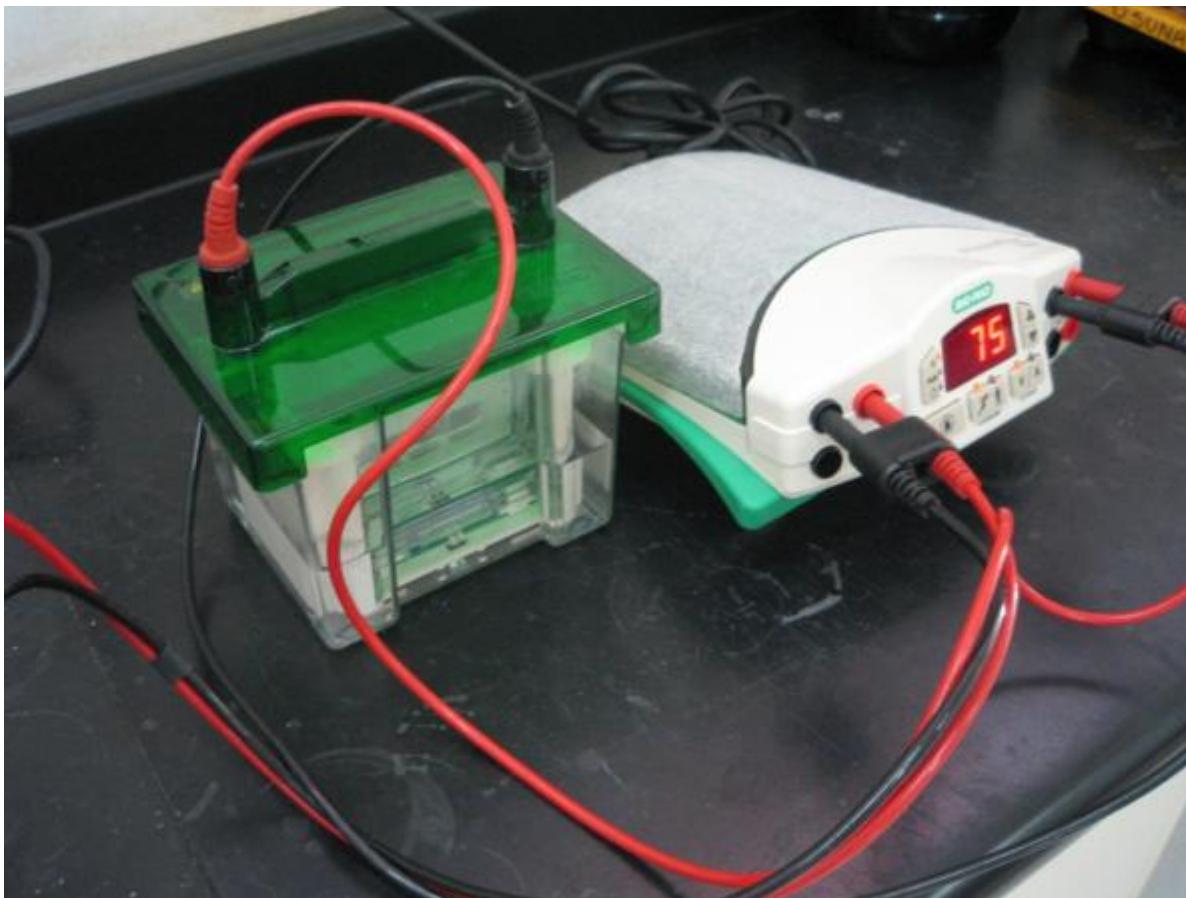


# ELECTROFORESIS

jareig@umh.es

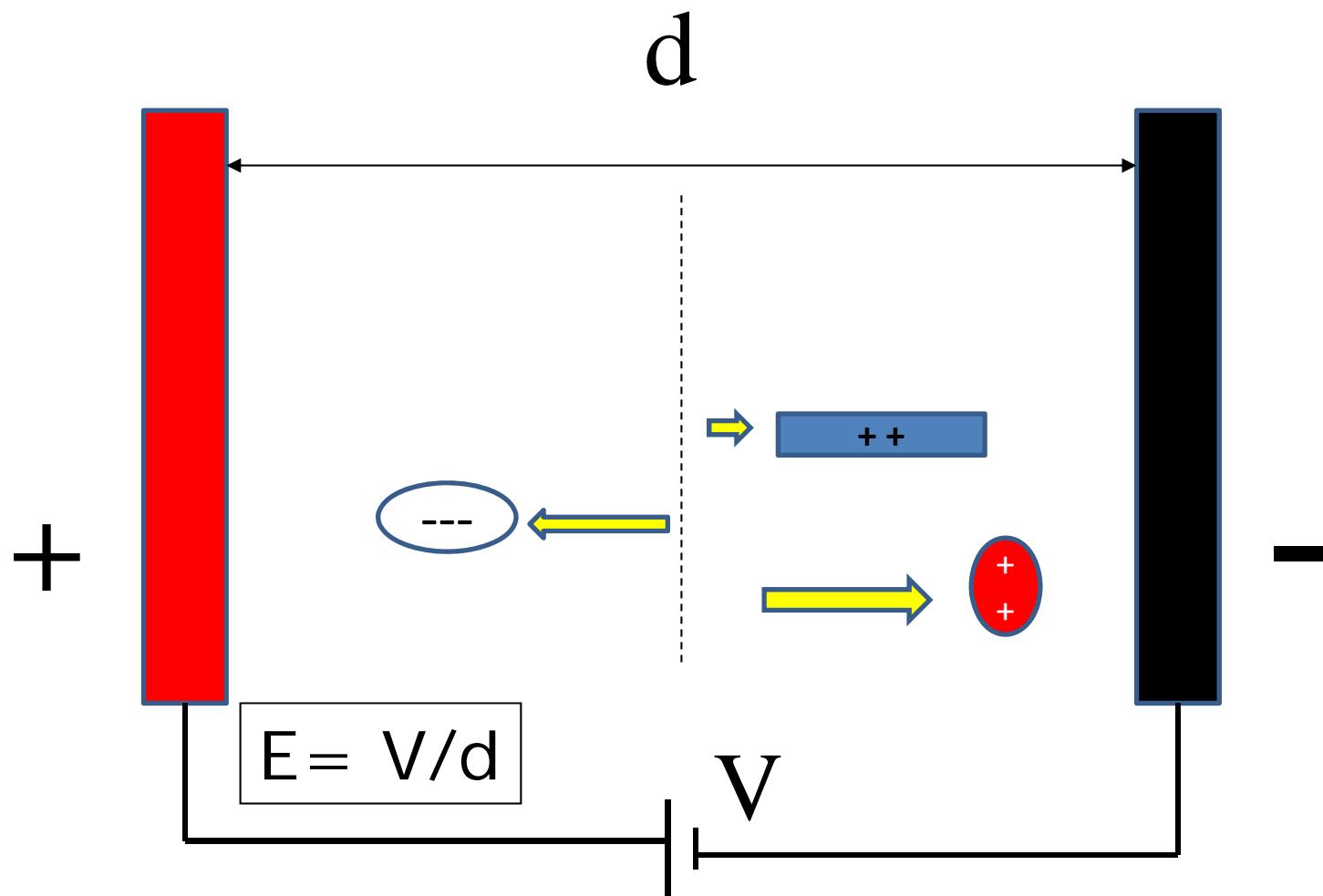


FUENTE  
DE ALIMENTACIÓN

CUBETA

**ELECTROFORESIS:** TRANSPORTE DE MOLÉCULAS A TRAVÉS DE UN CAMPO ELECTRICO  
TECNOLOGÍA ANALÍTICA O TECNOLOGIA PREPARATIVA → PROTEÓMICA

LA CARGA ELECTRICA ESTÁ ASOCIADA FRECUENTEMENTE CON LAS BIOMOLÉCULAS  
 $\text{-NH}_3^+$     $\text{-COO}^-$     $\text{-PO}_4^{3-}$

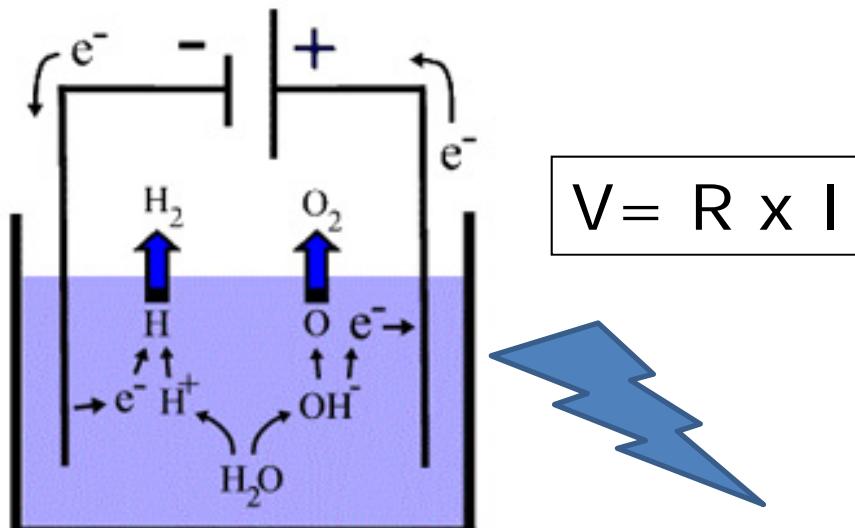


$$\mu_r = \phi \text{ (carga, tamaño, forma, viscosidad...)}$$

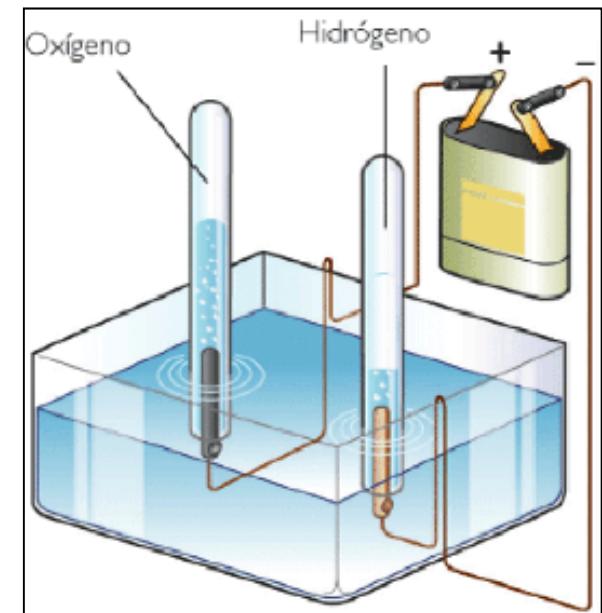
ELECTROFORESIS →: ELECTROLISIS → "burbujas y calor"

La corriente entre electrodos se mantiene por los electrolitos del buffer que mantienen **pH constante**

**Electrolysis:** Splitting water with electricity to produce hydrogen and oxygen:



$$V = R \times I$$



$$W = I^2 \times R$$

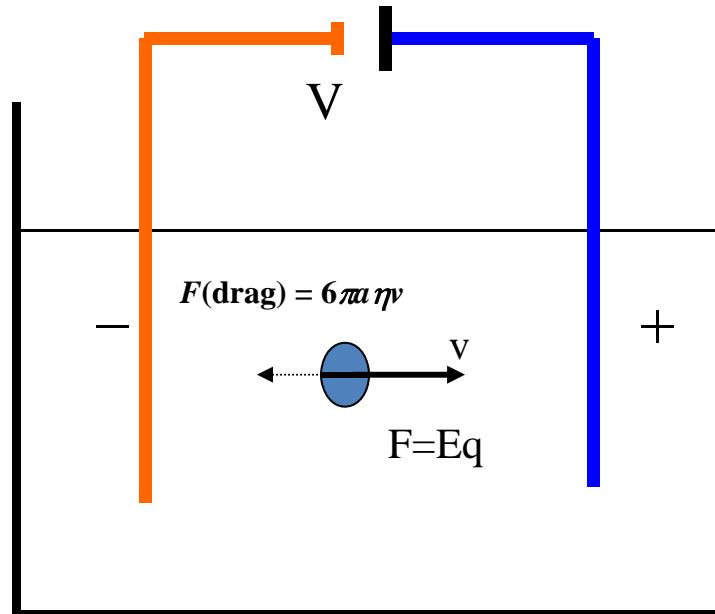


$F$   $q_1$   $q_2$   $F$

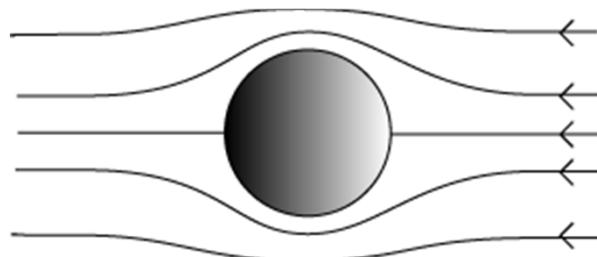
Like charges repel  
Unlike charges attract

$q_1 F$   $F q_2$

$$F = \frac{kq_1q_2}{r^2} = \frac{q_1q_2}{4\pi\epsilon_0 r^2} \text{ Coulomb's Law}$$

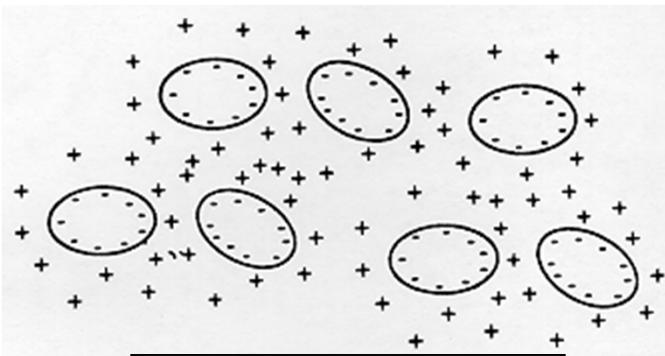


$v = F/f$   
 $f$ ; coeficiente friccional  
 $v = Eq/f = Eq/6\pi\eta r$   
 $\mu = v/E = q/6\pi\eta r$   
 $\mu$ ; Movilidad electroforética



Ley Stokes en condiciones ideales

## ATMOSFERA IONICA TEORIA DEBYE-HUCKEL



Función de Henry

$$v = Eq / 6\pi r \eta \times 1 / 1 + KR$$

ideal x factor de corrección

$$K = [8\pi Ne^2 / 1000 DkT]^{1/2} \times I^{1/2}$$

D=constante dieléctrica; I=fuerza iónica

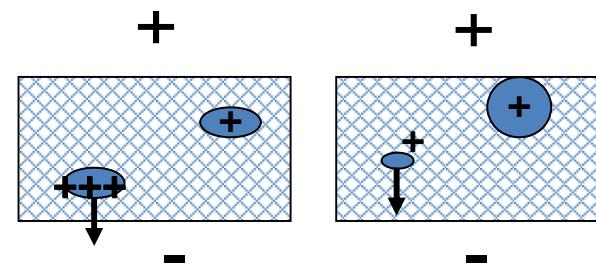
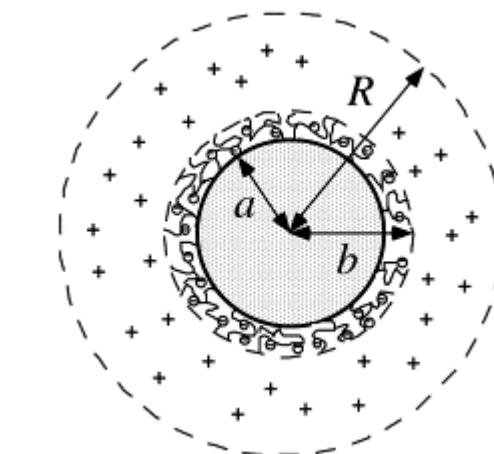
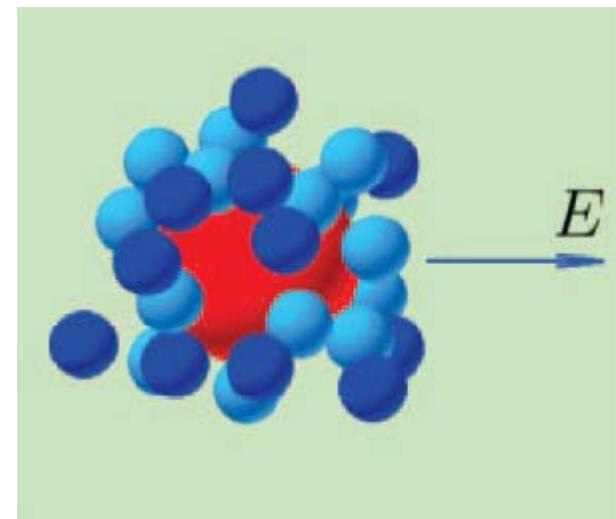
**Bioquimica física, Van Holde ED Exedra**

Caso ideal  $v = Eq / 6\pi r \eta$

$$\mu = v / E = \phi (q/m)$$

No existe formulación matemática accesible que represente la movilidad electroforetica en función de parametros moleculares

**Electroforesis un técnica empírica.**

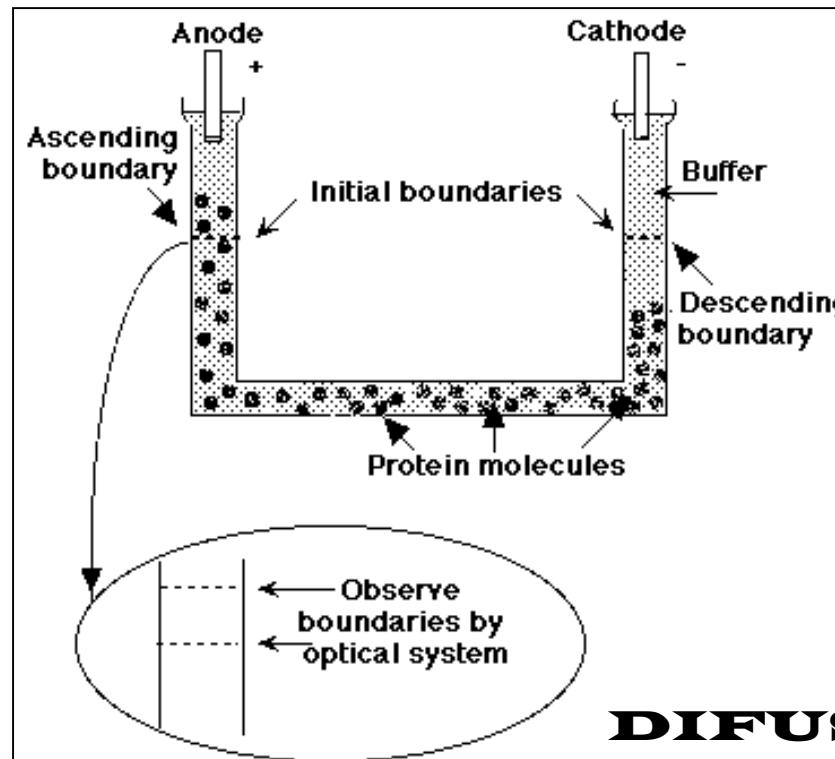


## ELECTROFORESIS LIBRE: En solución acuosa

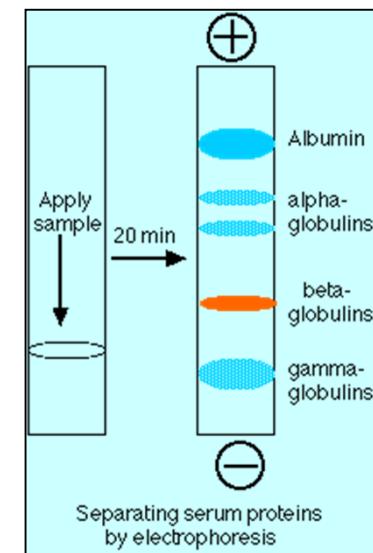


TISELIUS  
PRIMEROS  
EXPERIMENTOS  
ELECTROFORETICOS

PREMIO NOBEL 1948  
"Estudios del plasma sanguíneo"



ELECTROFORESIS ZONAL → En soporte  
No restrictivo: papel, agarosa (poro >> molécula)  
Restrictivo: poliacrilamida (mayor entramado)

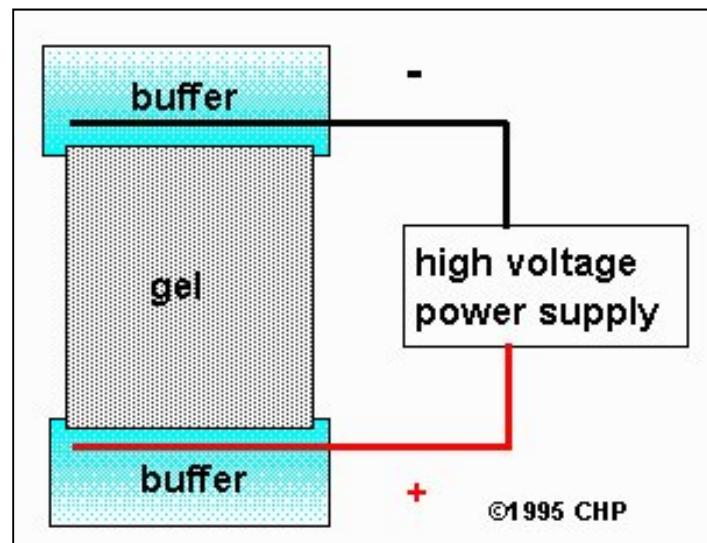


## EQUIPO BASICO:

FUENTE Y CUBETA

Un tampón determinado (pH)

$\mu = \phi$  ( $q/m$ )    $q = f$  (pH)



## SEGÚN EL SOPORTE

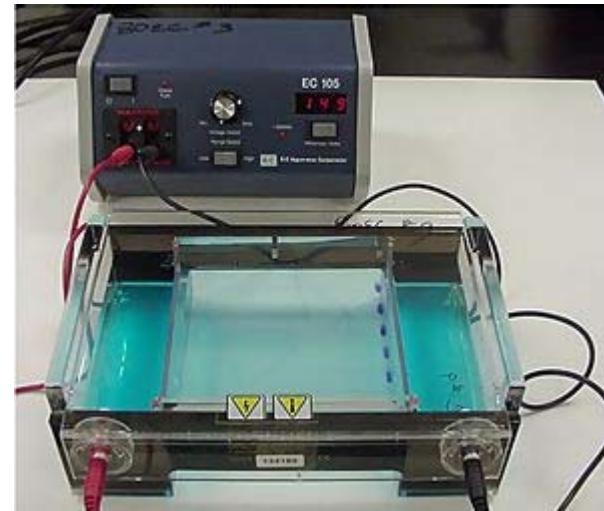
PAPEL

CAPILAR

GELES.

AGAROSA

POLIACRILAMIDA (PAGE)



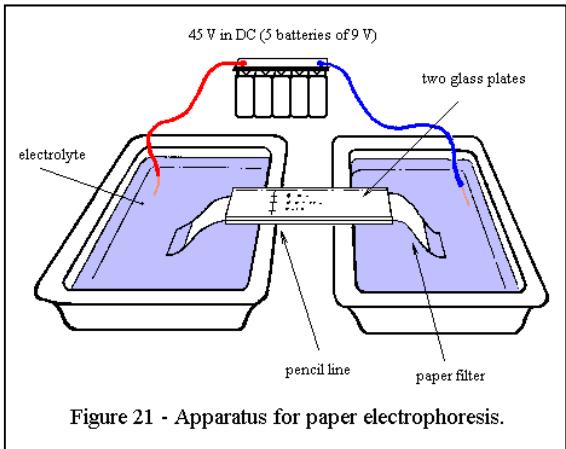
HORIZONTAL → AGAROSA  
ACIDOS NUCLEICOS



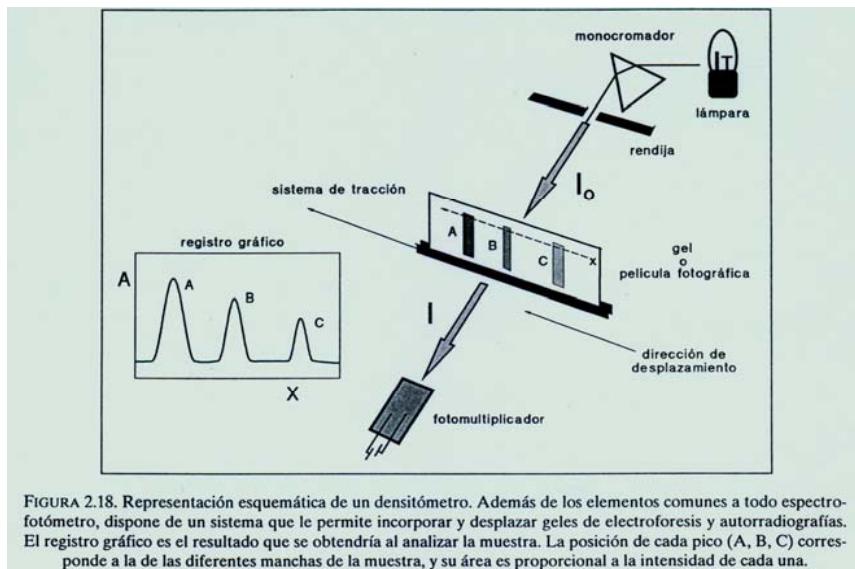
VERTICAL → ACRILAMIDA (PAGE)  
PROTEINAS

## ELECTROFORESIS EN PAPEL

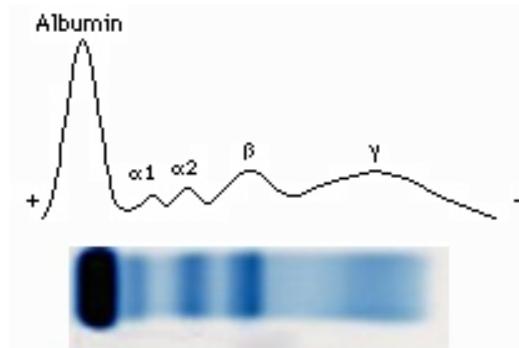
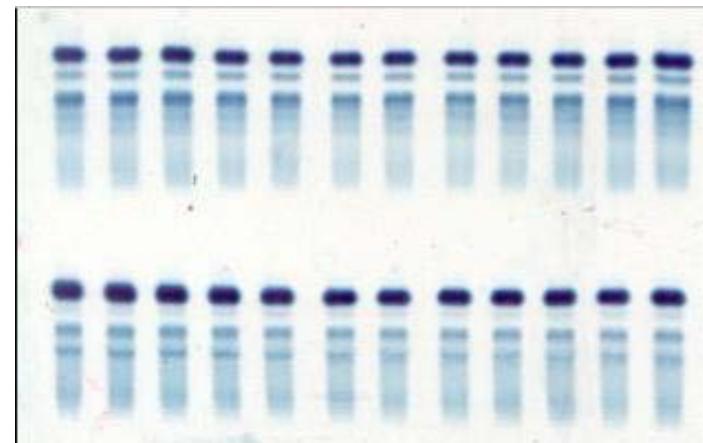
### Moléculas pequeñas



## ACETATO DE CELULOSA

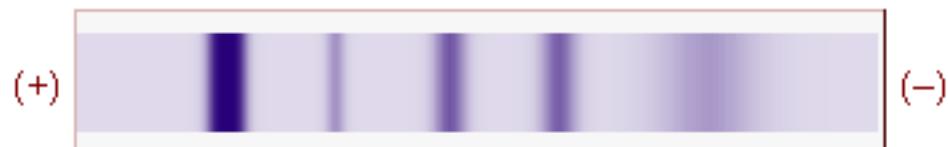


## densitometría

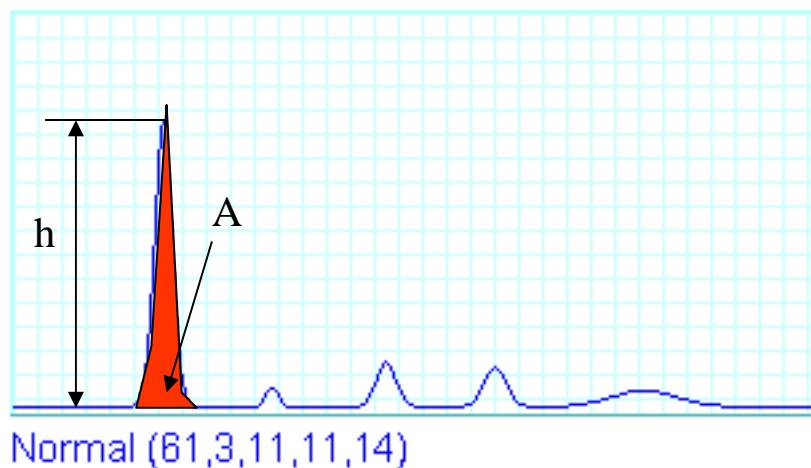


## DENSITOGRAMA → METODO DE CUANTIFICACION ELECTROFORETICA

Tira de acetato de celulosa:



Densitograma:



Tinción con Azul de Coomassie ▾

Densitograma en color Azul ▾

Superponer curvas

Datos:

61% albúmina

3% globulinas alfa-1

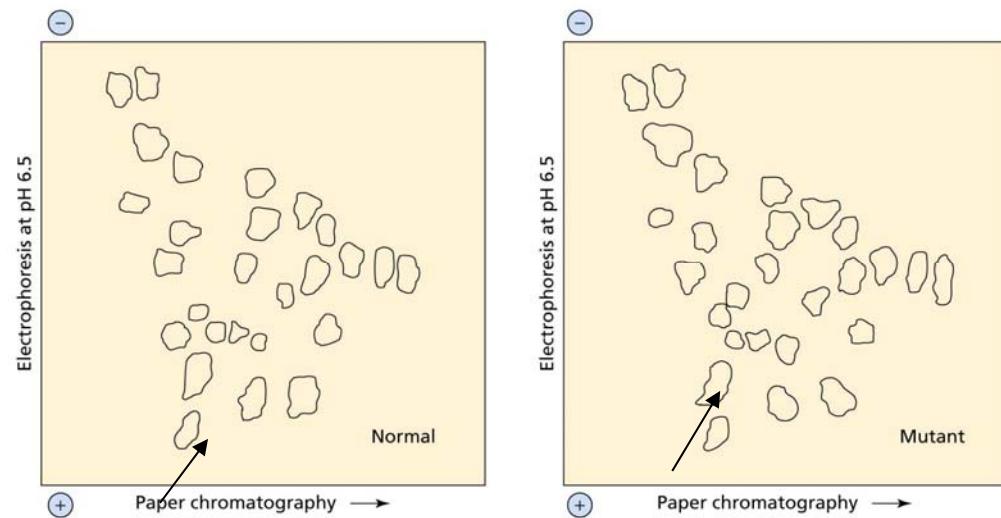
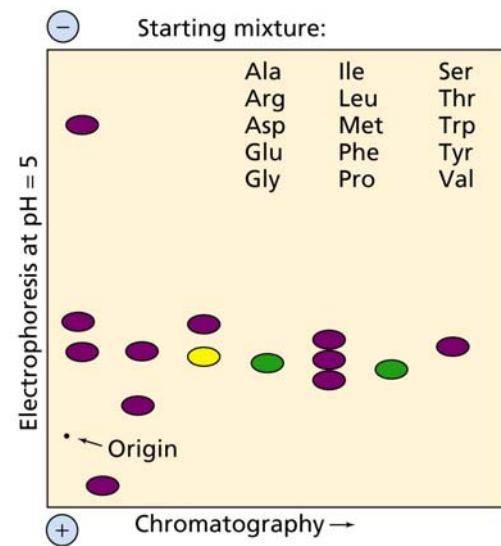
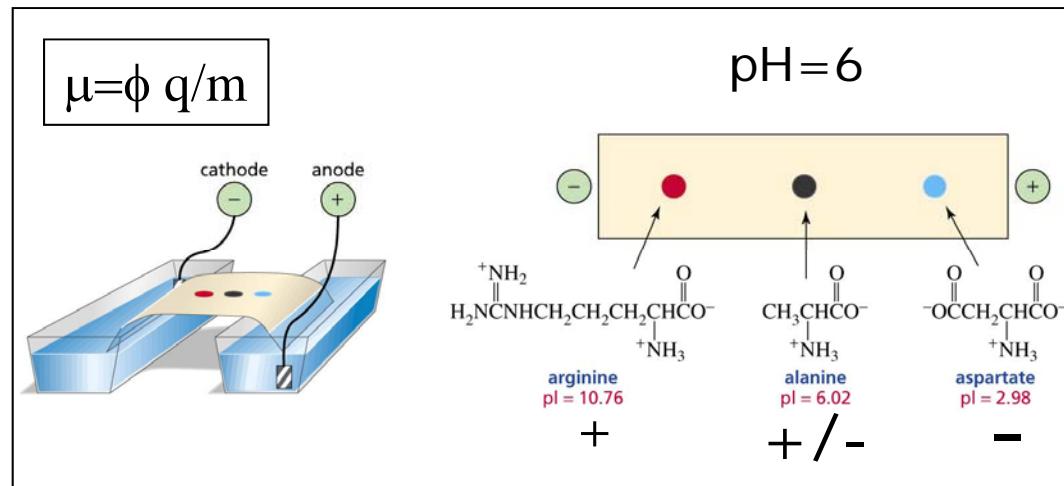
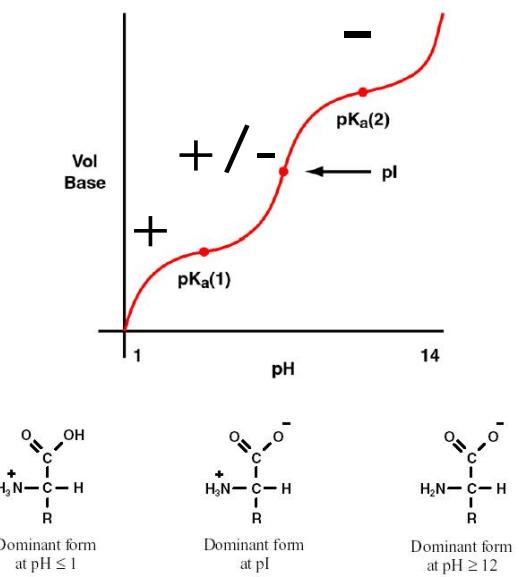
11% globulinas alfa-2

11% globulinas beta

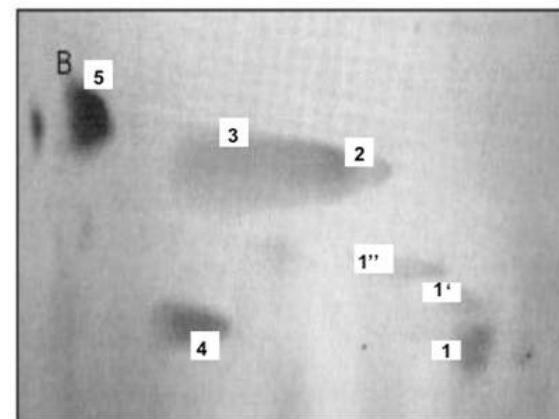
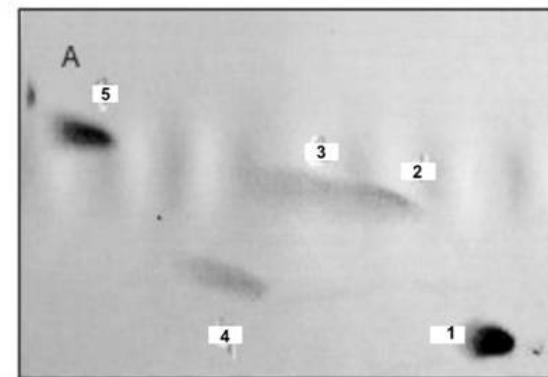
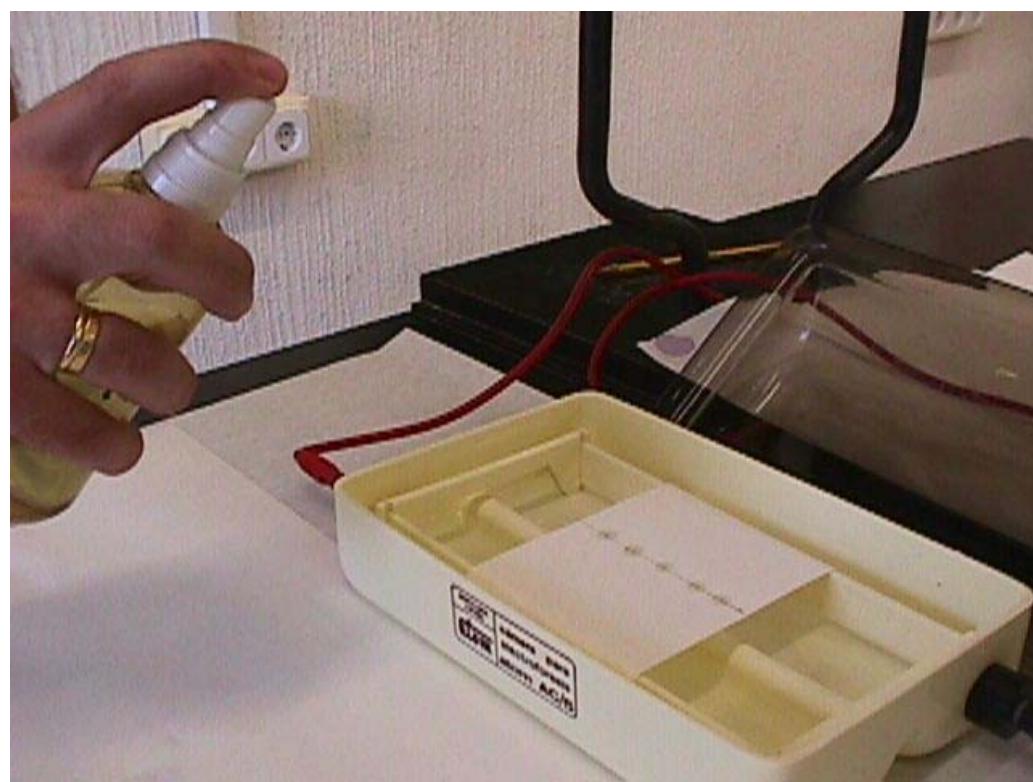
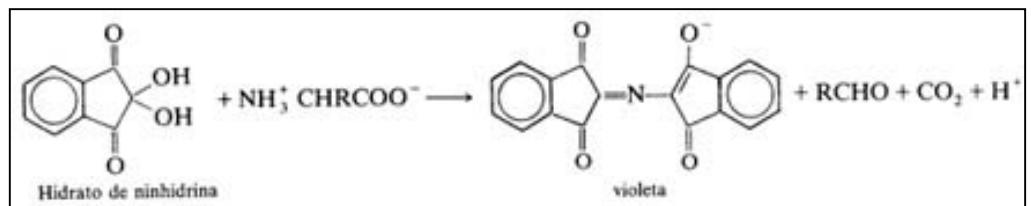
14% globulinas gamma

[Ver resultado](#)

Perfiles patológicos: ...elige uno... ▾

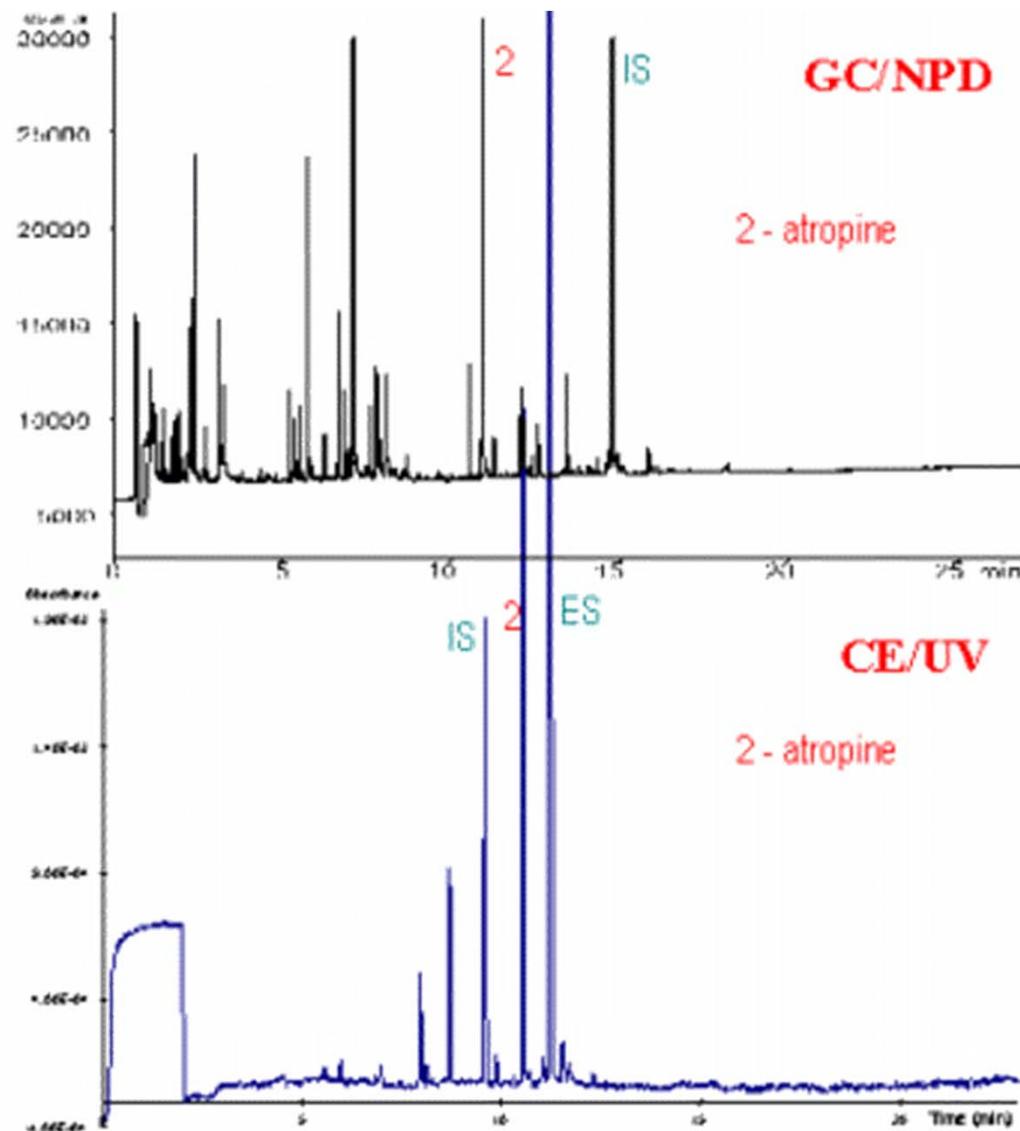
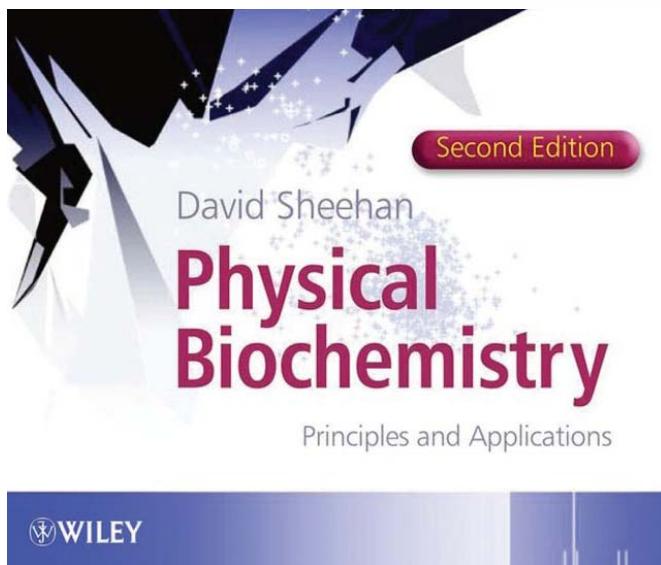


SEPARACION BIDIMENSIONAL: CROMATOGRAFIA+ELECTROFORESIS

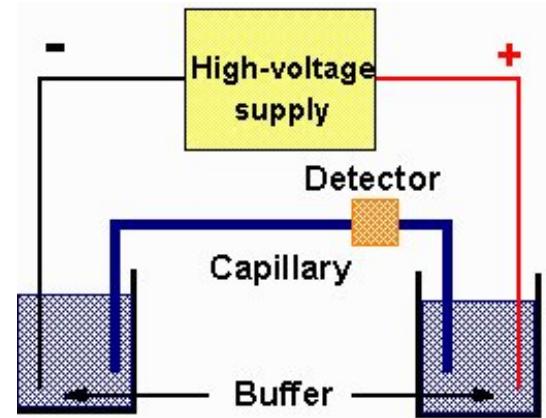


**ALTO VOLTAJE!**  
**(5-50KV)**

## ELECTROFORESIS CAPILAR

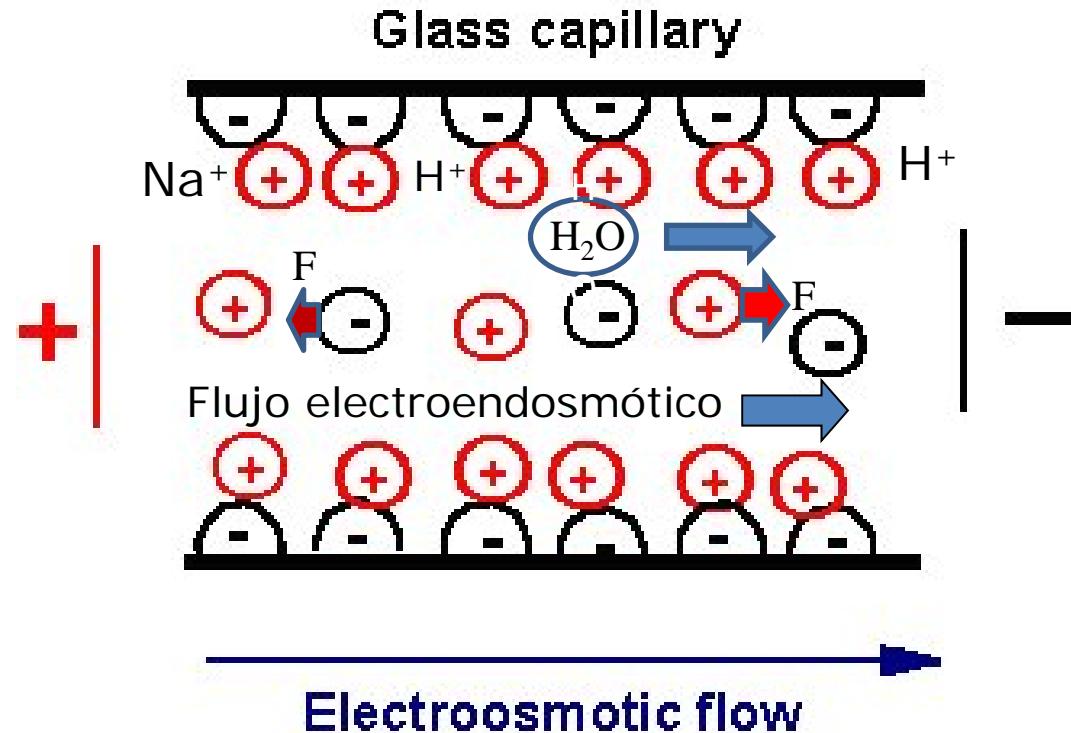


## Free solution CE Capilar Gel Electroforesis



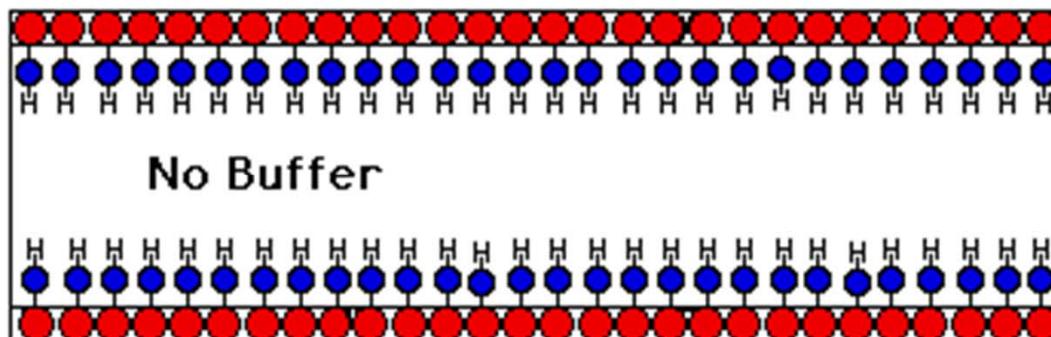
$\varnothing=20\text{-}200 \mu\text{m}$   
 $V=$  hasta 50.000V

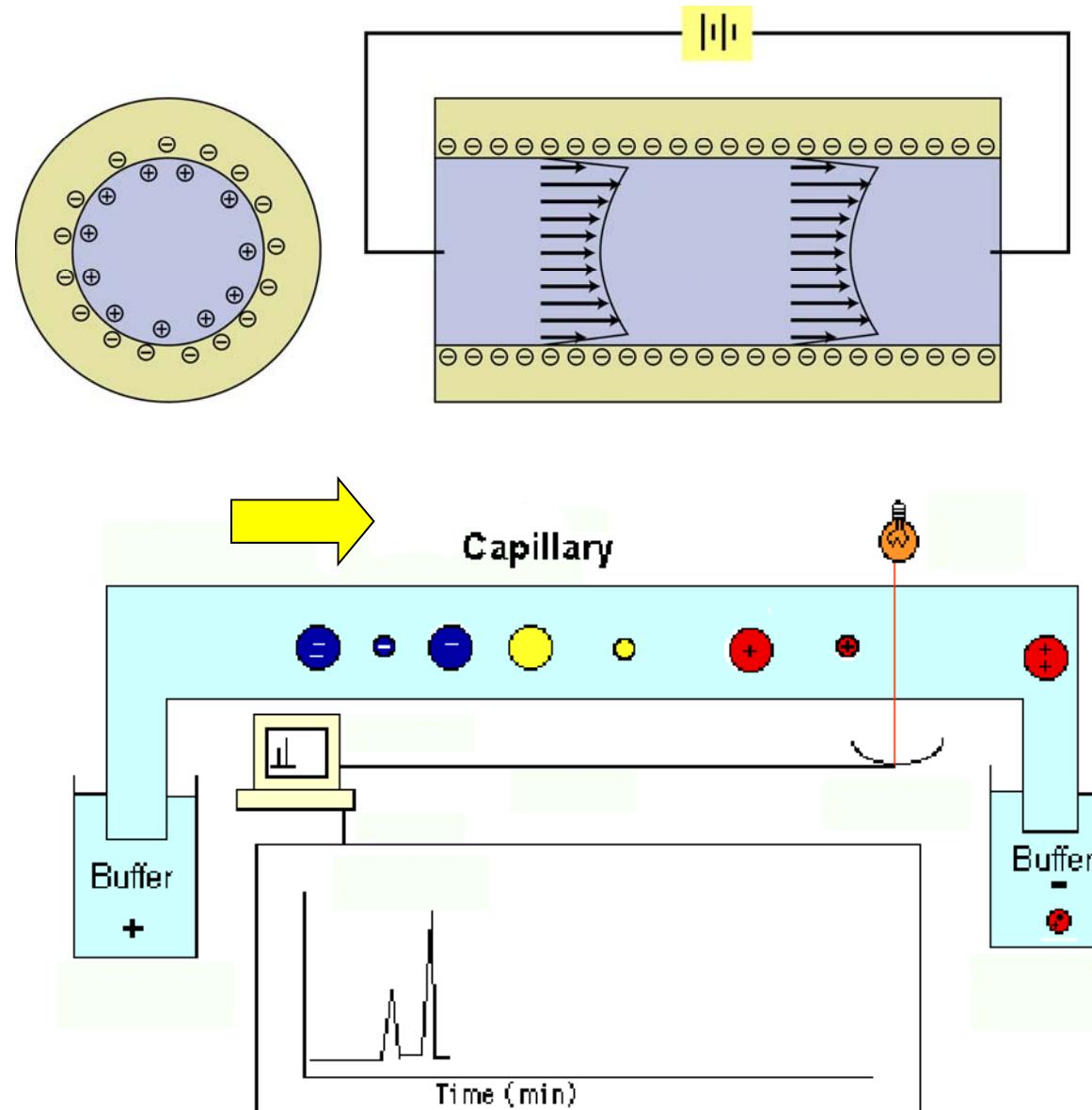
Pared de capilar compuesto de silicatos  
Grupos silanol :  $\text{R}_3\text{Si-OH} \rightleftharpoons \text{R}_3\text{Si-O}^- + \text{H}^+$

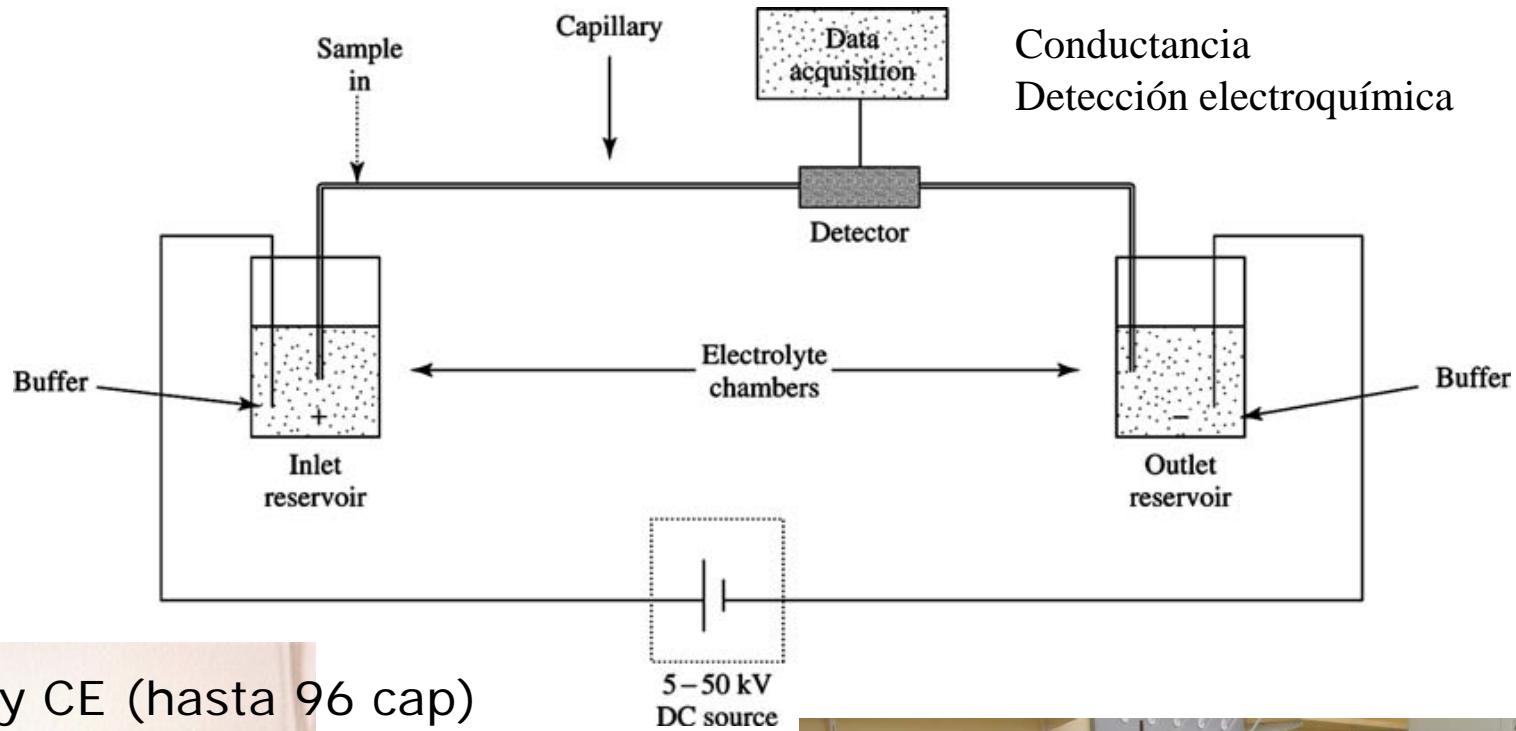


## Endoosmotic Flow

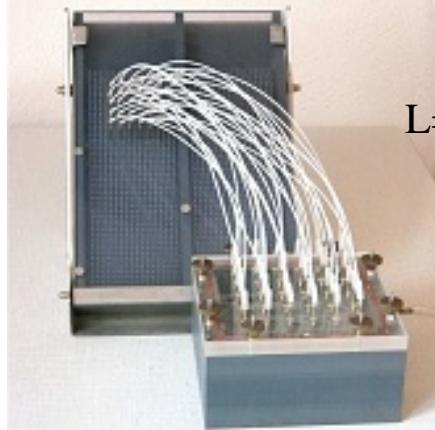
— Silica  
— Oxygen  
— Hydrogen







Array CE (hasta 96 cap)



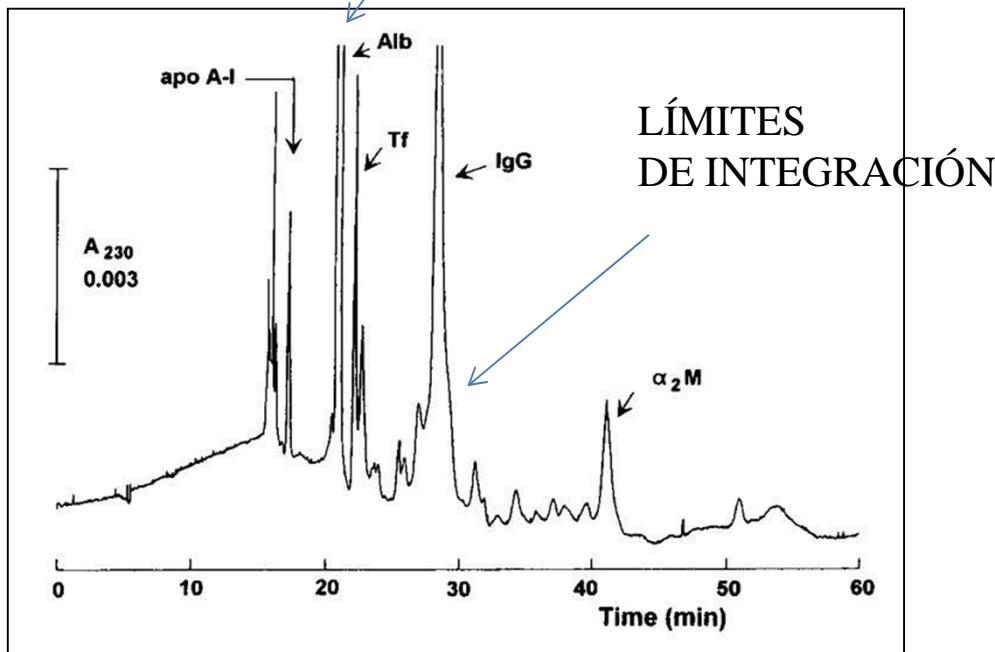
$\phi_i = 10-100 \mu\text{m}$



# Rápido y cuantificable ¡HAY QUE OPTIMIZAR CONDICIONES DE SEPARACIÓN!

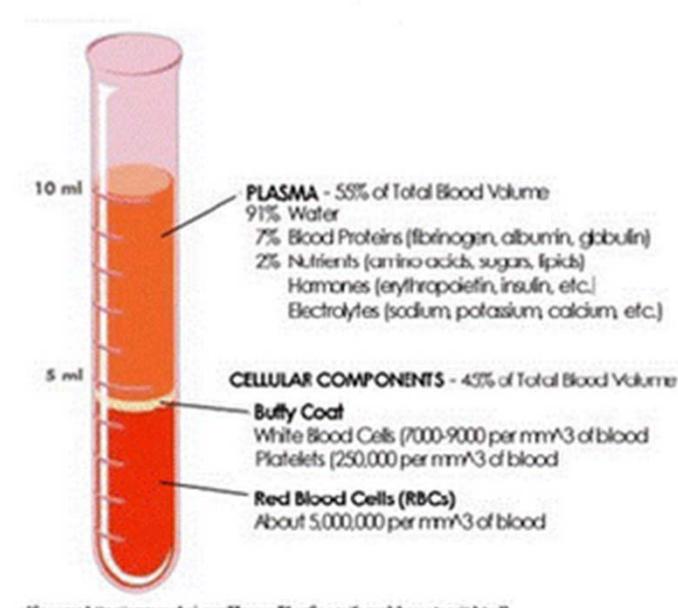
Proteinas del plasma  
(incluye los componentes de la coagulación)

## SATURACIÓN DE ALBUMINA



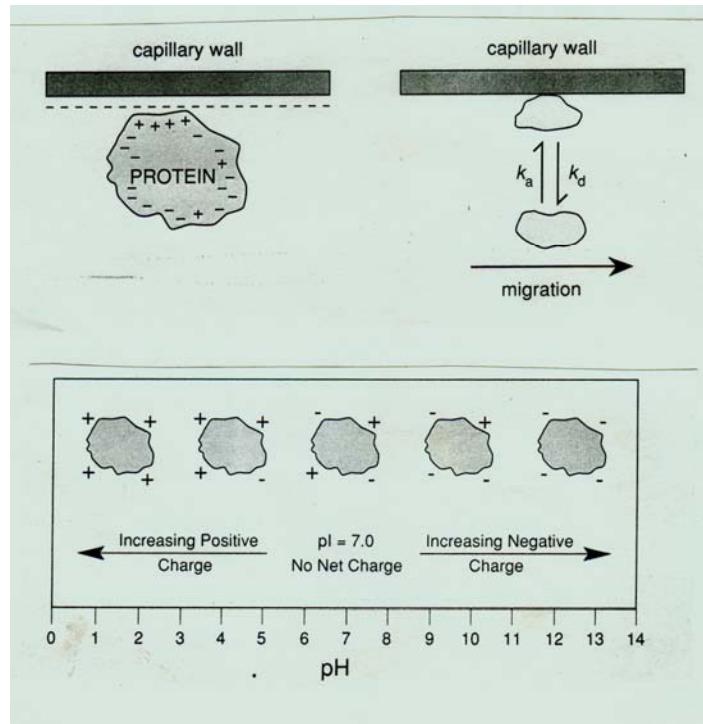
## Optimizing CE Separations-POINT method

- pH
  - First parameter to control
  - Effects EOF and mobility (charge)
- Organic Solvent
  - Analyte solvation
- Interacting agent
  - Ion-pairing, solvation, etc.
- Non-aqueous Conditions
  - Solvation and charge
- Temperature
  - Solvation, chemical equilibria

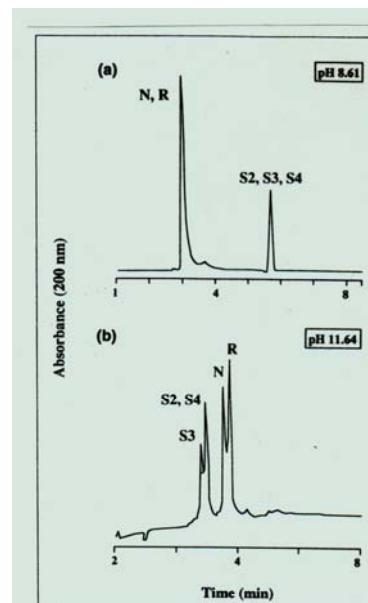


Grupos silanol :  
 $\text{R}_3\text{Si-OH} \rightleftharpoons \text{R}_3\text{Si-O}^- + \text{H}^+$

A menor pH, más protonada la pared,  
 menos carga y menor EOF.  
 En teoria, mejor resolucion...pero a pH bajo proteinas  
 con más carga + interaccionan más con la pared...  
 Hay que optimizar condiciones!!

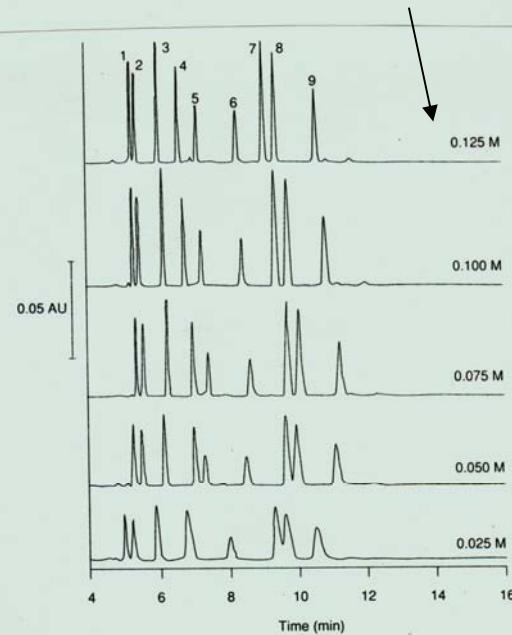


Aquí: A mayor pH  
 mejoramos resolución!

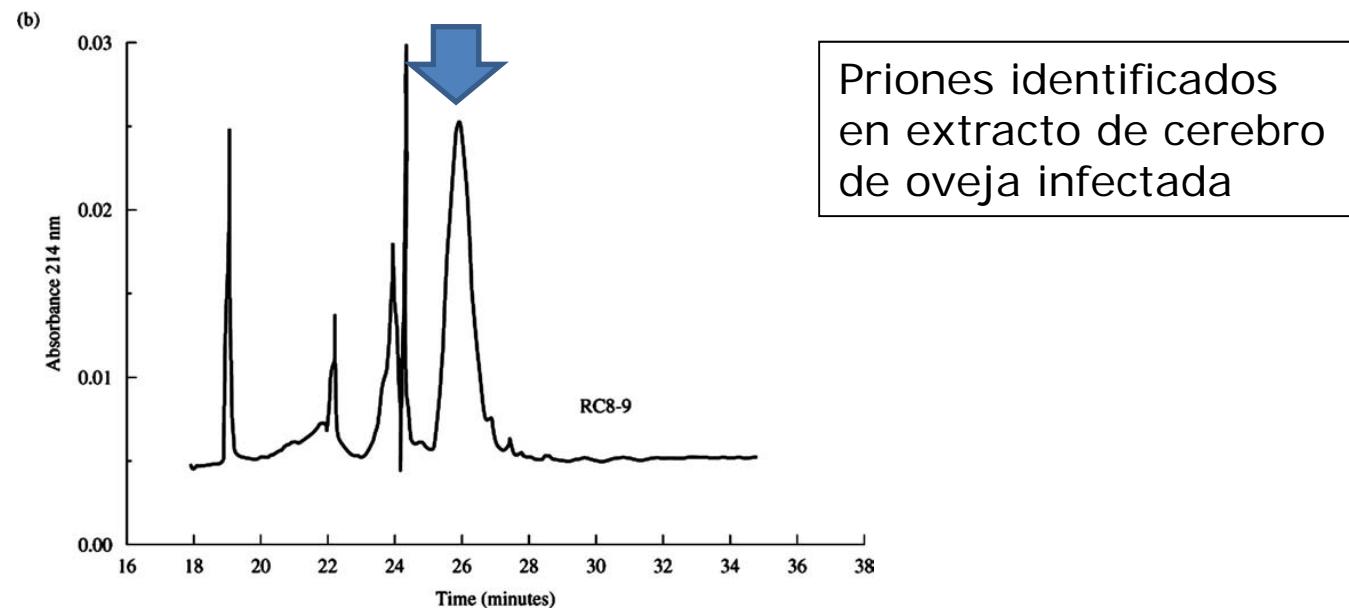
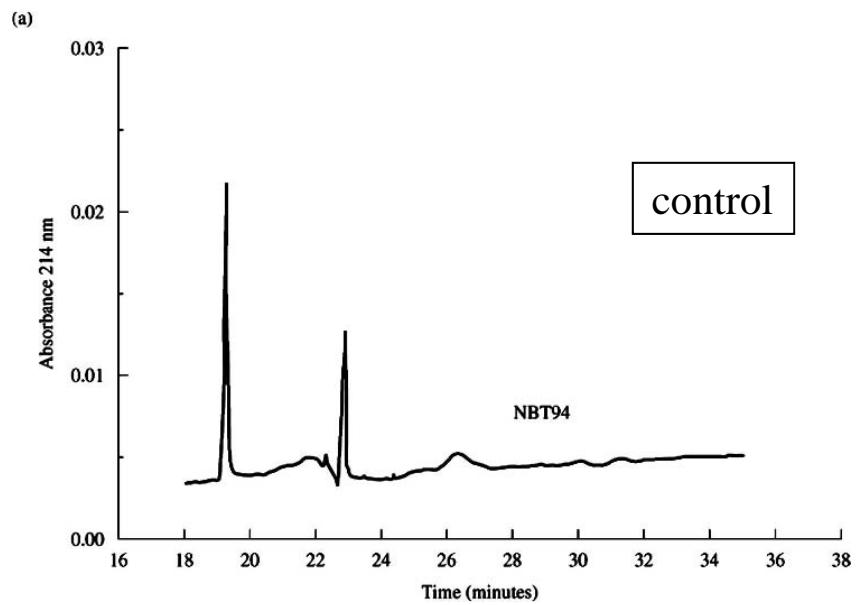


**Figure 3**  
 The pH-dependent separation of peptides with 'shuffled' amino acid sequences. The model peptide system used was a 12 amino acid sequence, KTNYC-TKQPKSY, from residues 101–112 of the thyroid-stimulating-hormone receptor. Separation was carried out in a 50  $\mu\text{m}$   $\times$  20 cm capillary in 100 mM borate, 10 mM diaminopentane, with pH adjusted to (a) 8.61 and (b) 11.64 with NaOH. N, R and S represent the 'native', 'reverse' and 'shuffled' peptide sequences, respectively. The instrumentation used was a Beckman P/ACE 2050 with detection at  $\lambda = 200$  nm.

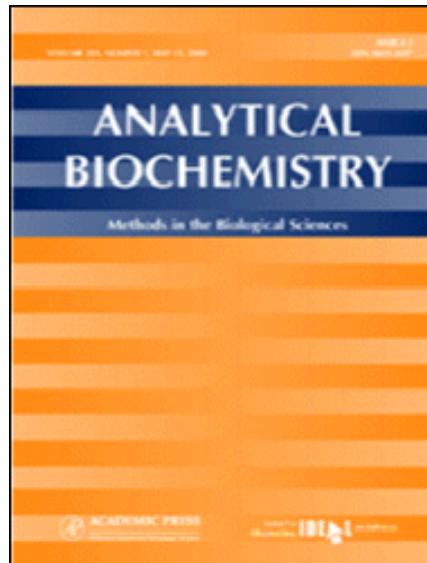
Incrementando fuerza iónica  
 disminuye interacción con pared y  
 mejora resolución..



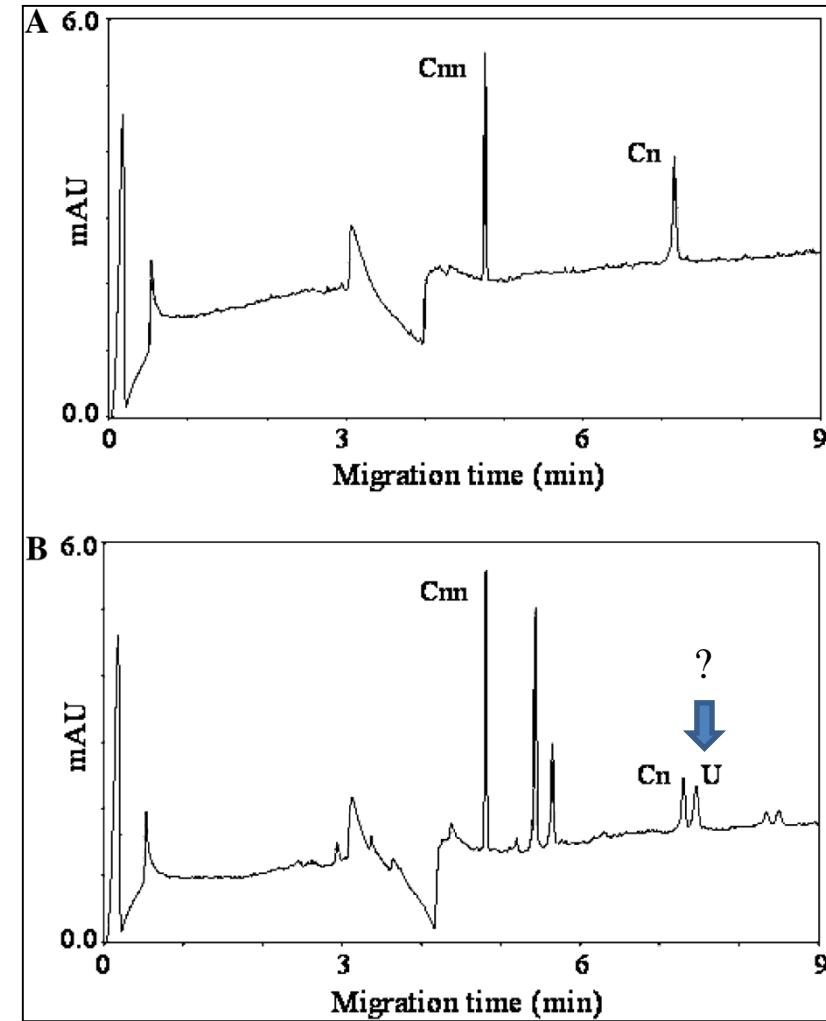
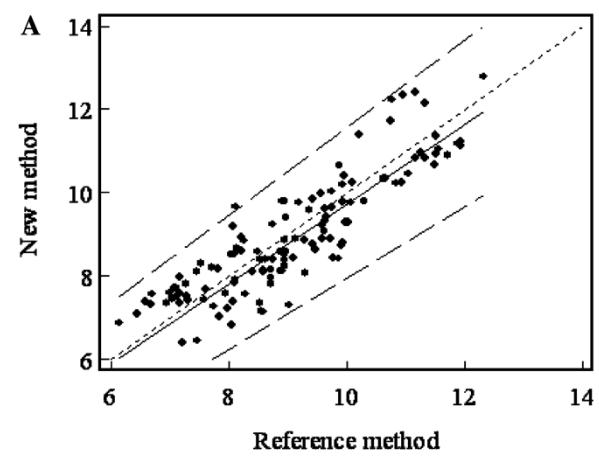
**Figure 1-26.** Effect of the buffer ionic strength on peak shape and migration time of peptides. Buffer: sodium phosphate, pH 2.44; 30 kV; 20°C; 200 nm; 57 (50) cm  $\times$  75  $\mu\text{m}$  capillary. Peak identification: (1) dynorphin; (2) bradykinin; (3) angiotensin II; (4) TRH; (5) LHRH; (6) bombesin; (7) leu-enkephalin; (8) met-enkephalin; (9) oxytocin. From McLaughlin et al., Beckman Technical Information Bulletin TIBC-106 (1991).



"Plasma creatinine and creatine quantification by capillary electrophoresis diode array detector"  
Analytical Biochemistry vol342 (2005) 186–193



Sangre +  
anticoagulante  
⊕  
3000g x 5min  
PLASMA  
↓  
TCA 5%  
⊕  
SOBRENADANTE



75mM Tris/fosfato pH=2,25

<http://www.ncbi.nlm.nih.gov/pubmed>

Capillary electrophoresis

The screenshot shows the PubMed homepage within a Windows Internet Explorer window. The address bar at the top displays the URL <http://www.ncbi.nlm.nih.gov/pubmed>. The browser's toolbar includes icons for Back, Forward, Stop, Refresh, and Home, along with links for Favorites, Google, Bing, and other NCBI resources. The main content area features a large image of an open book on the left and a dark blue sidebar on the right containing the word "PubMed". Below the sidebar, a text box states: "PubMed comprises more than 21 million citations for biomedical literature from MEDLINE, life science journals, and online books. Citations may include links to full-text content from PubMed Central and publisher web sites." The main content area is divided into three sections: "Using PubMed" (with links to Quick Start Guide, Full Text Articles, FAQs, Tutorials, and New and Noteworthy), "PubMed Tools" (with links to PubMed Mobile, Single Citation Matcher, Batch Citation Matcher, Clinical Queries, and Topic-Specific Queries), and "More Resources" (with links to MeSH Database, Journals in NCBI Databases, Clinical Trials, E-Utilities, and LinkOut). The bottom of the screen shows the Windows taskbar with various pinned icons and the system tray.

The analysis of human amniotic fluid using capillary electrophoresis [Electrophoresis. 2001] - PubMed - NCBI - Windows Internet Explorer

http://www.ncbi.nlm.nih.gov/pubmed/11358139

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The analysis of human amniotic fluid using capillary electrophoresis [E...]

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Performing your original search, *amniotic fluid capillary electrophoresis*, in PubMed will retrieve 19 records.

Electrophoresis. 2001 Apr;22(6):1136-42.

**The analysis of human amniotic fluid using capillary electrophoresis.**

Stewart CJ, Iles RK, Perrett D.

Department of Medicine, St Bartholomew's & the Royal London School of Medicine & Dentistry, St Bartholomew's Hospital, West Smithfield, UK.

**Abstract**

This study has investigated the composition of amniotic fluid (AF) using capillary electrophoresis (CE). A detailed optimisation investigation was made to determine the major peaks in amniotic fluid. In the final method, capillary zone electrophoresis (CZE) of AF was performed on a Hewlett Packard 3D CE instrument with a total length of 36 cm from the detector to the inlet. The background electrolyte was 20 mM sodium tetraborate containing 1% acetonitrile and diluted 1 plus 1 with deionised water prior to hydrodynamic injection for 3 s at 50 mbar. The separation was performed at +22.5 kV and resulted in a run time of less than 10 min. A scheme for the identification of peaks once they had been separated was also developed. Four peaks have been identified and tentatively assigned to albumin, transferrin, α<sub>1</sub>-antitrypsin and α<sub>1</sub>-antichymotrypsin. Surprisingly, one major peak was shown to be the purine catabolite, xanthine.

PMID: 11358139 [PubMed - indexed for MEDLINE]

MeSH Terms, Substances

LinkOut - more resources

http://www.ncbi.nlm.nih.gov/guide/



[www.electrophoresis-journal.com](http://www.electrophoresis-journal.com)

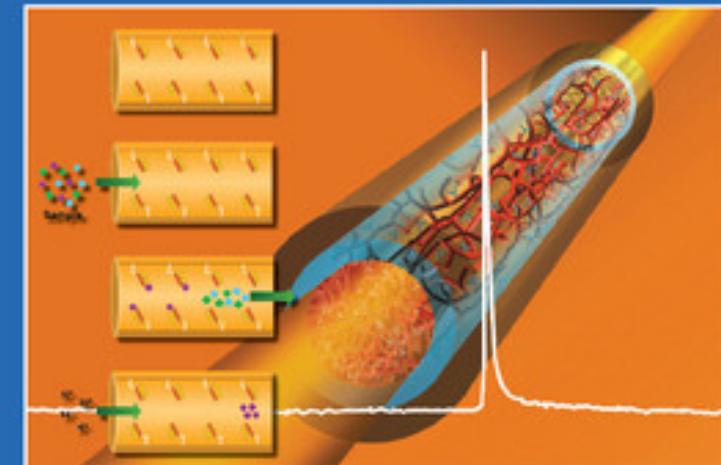


# ELECTROPHORESIS

## SPECIAL ISSUE

16'08

www.electrophoresis-journal.com



Affinity and Immunoaffinity  
CE and CEC

Editor:  
**Terry M. Phillips**

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Google Capillary gel electrophoresi...

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# ELECTROPHORESIS

Short communication

**Capillary gel electrophoresis for precise protein quantitation**

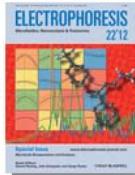
Claudia Cianciulli, Thomas Hahne, Hermann Wätzig\*

Article first published online: 12 SEP 2012  
DOI: 10.1002/elps.201200177  
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ELECTROPHORESIS

Special Issue: MicroScale Bioseparations and Analyses

Volume 33, Issue 22, pages 3276–3280, November 2012



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**Keywords:**  
CGE; Monoclonal antibody; Protein; Quantitation; SDS-CGE

CGE, also known as SDS-CGE, is being established in the pharmaceutical industry replacing SDS-PAGE. In most cases, the method is applied for the identity and purity control of proteins, for example monoclonal antibodies. In order to quantify these components with sufficient precision using the same quality control method, a RSD for the quantitative analysis under 2% is required. A reliable and highly precise CGE method could be obtained after thorough optimization. It was crucial to increase the sample concentration and the injection volume in order to achieve sufficiently high S/N ratios (>70). The application of hydrodynamic injection is beneficial for the precision of the method compared to the traditionally used electrokinetic one. Linearity was demonstrated and LOD and LOQ were estimated. Both injection modes were compared in long series runs ( $n = 48$ ). Furthermore, the use of an internal standard was investigated. Thus, the RSD% of the migration time was reduced from 0.9 to 0.2% and the RSD% of peak areas was greatly improved. However, the normalization to the total area further reduced the influence of the injection error. RSD% for the peak area ratios of typically between 1 and 2% was provided.

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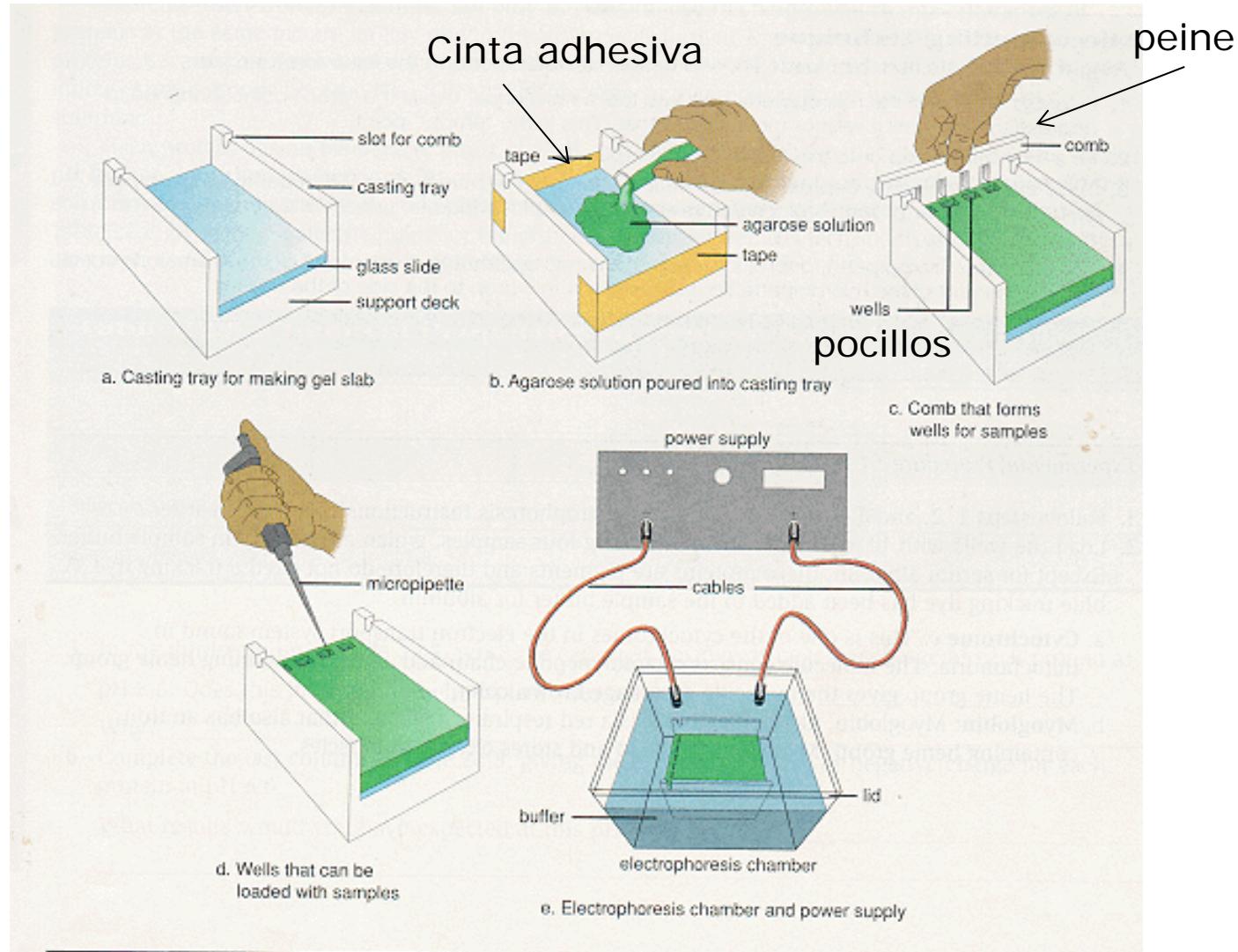
PAGE Microsoft PowerPo... Presentación de Po... Google - Windows I... Capillary gel electro...

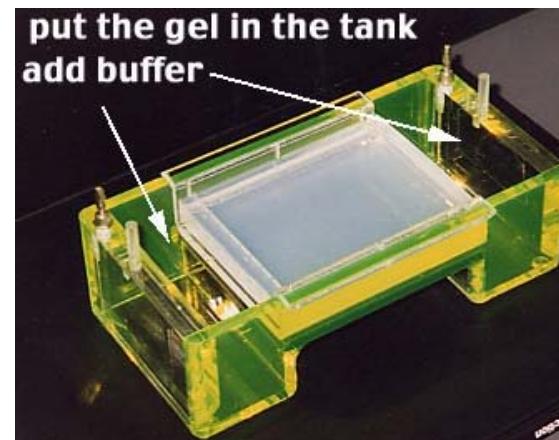
ES 15:31

## **ELECTROFORESIS EN GELES , MATRICES HIDRATADAS ESTABLES**

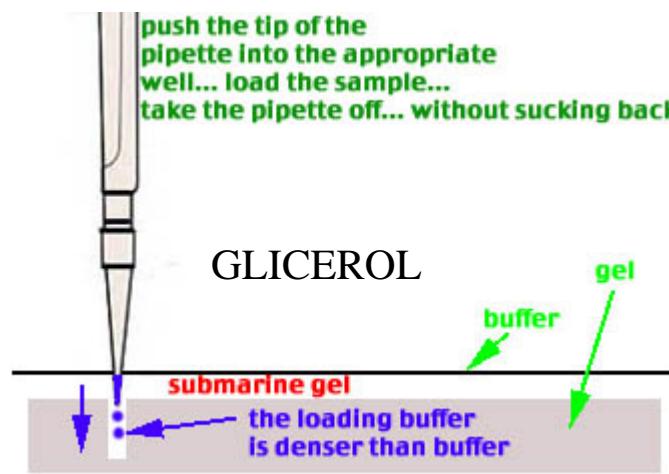
AGAR, AGAROSA: ACIDOS NUCLEICOS

POLIACRILAMIDA (PAGE): PROTEINAS



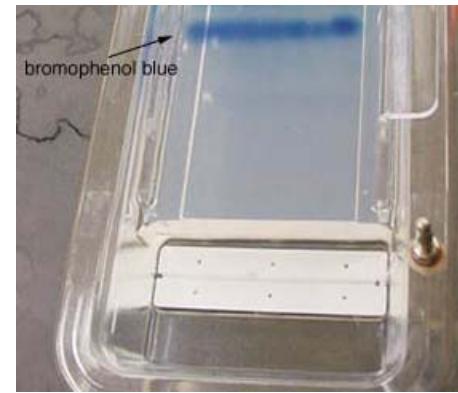
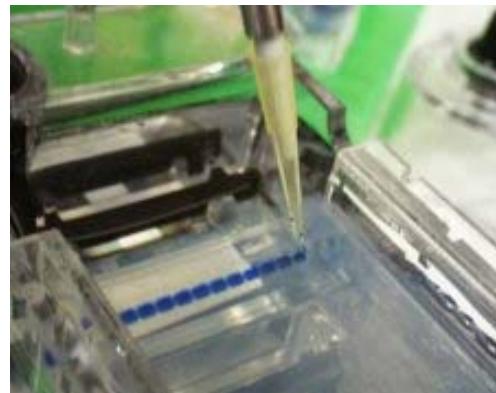
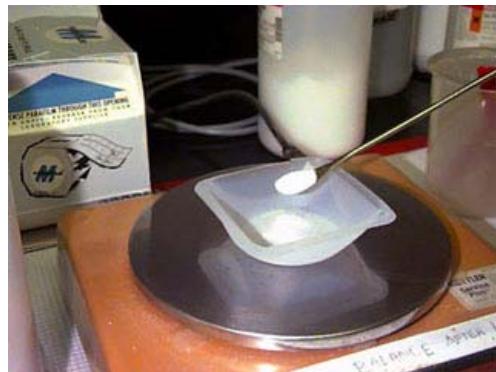
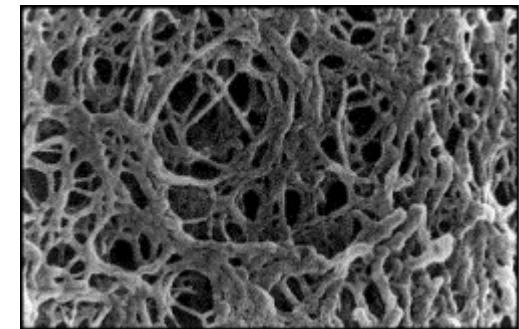
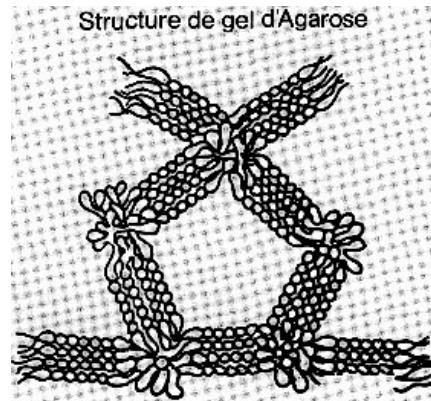
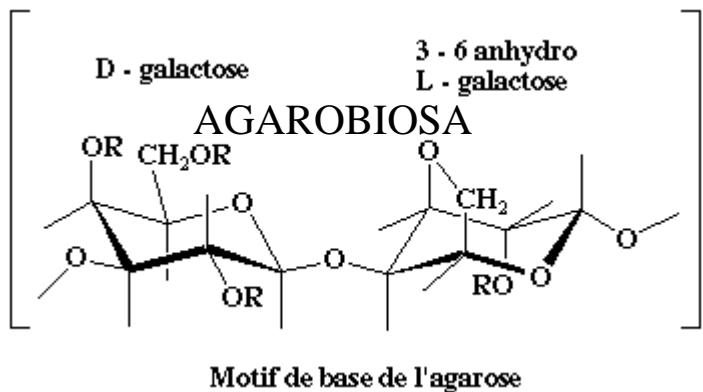


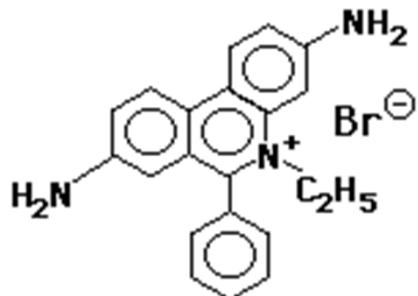
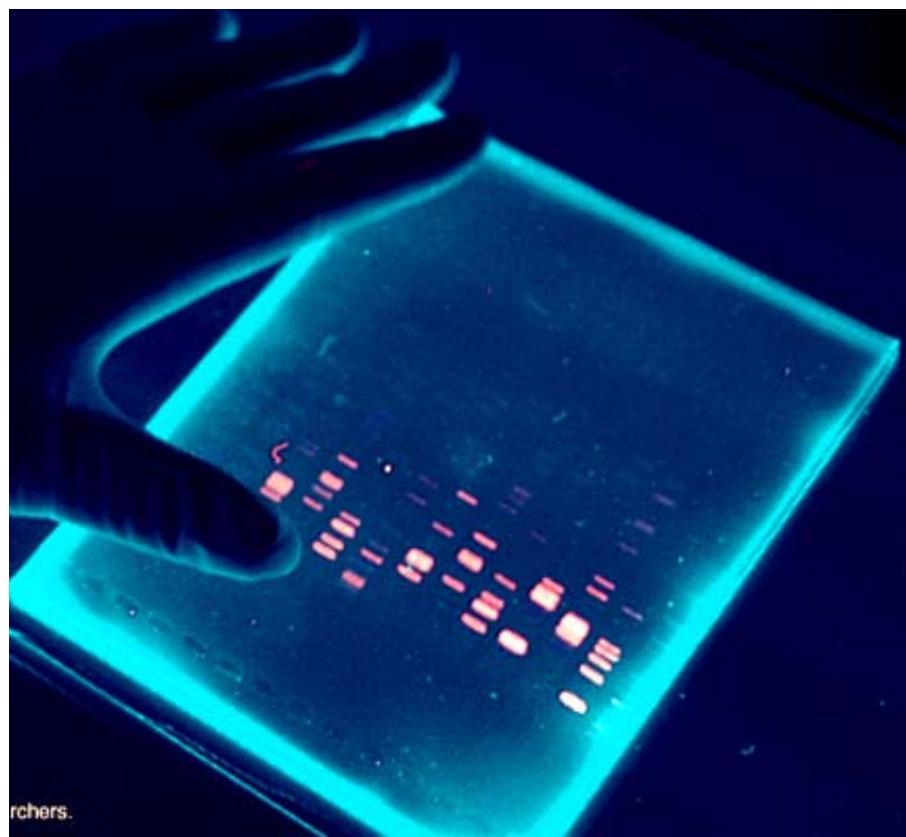
## PREFORMAR EL GEL



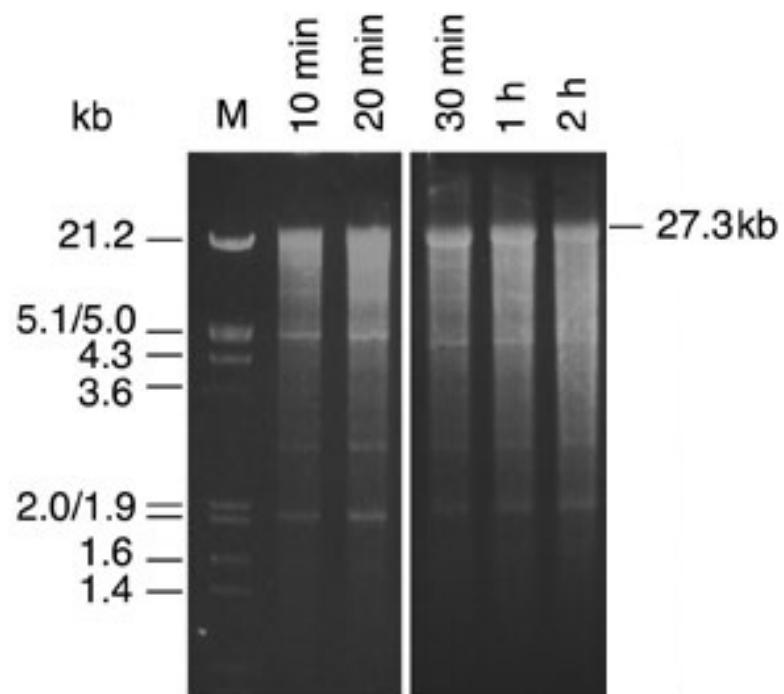
## APLICAR LA MUESTRA

## GELES DE AGAROSA: ACIDOS NUCLEICOS

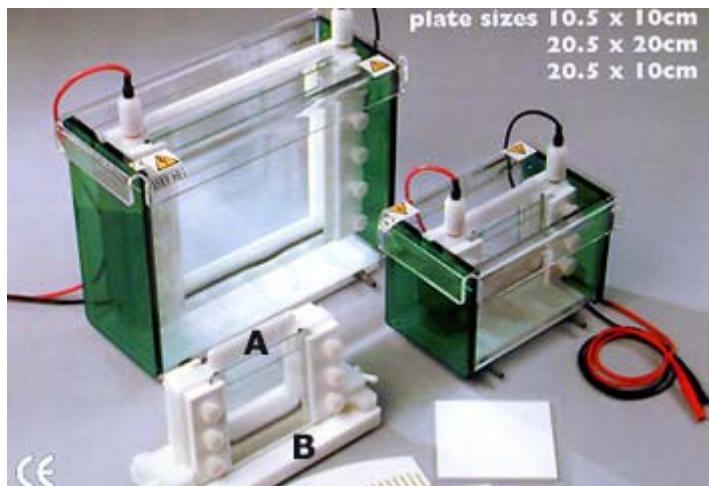
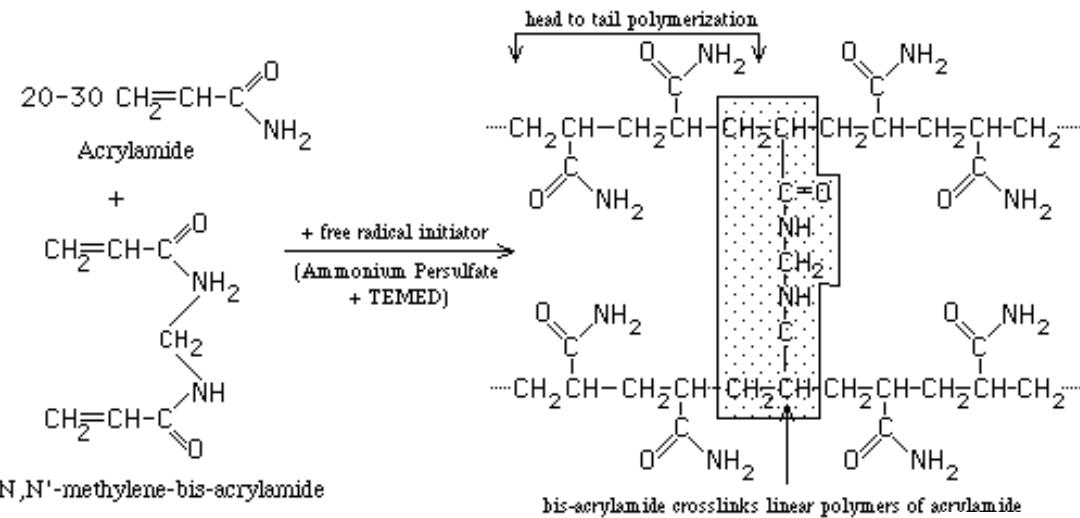




**Ethidium  
Bromide**

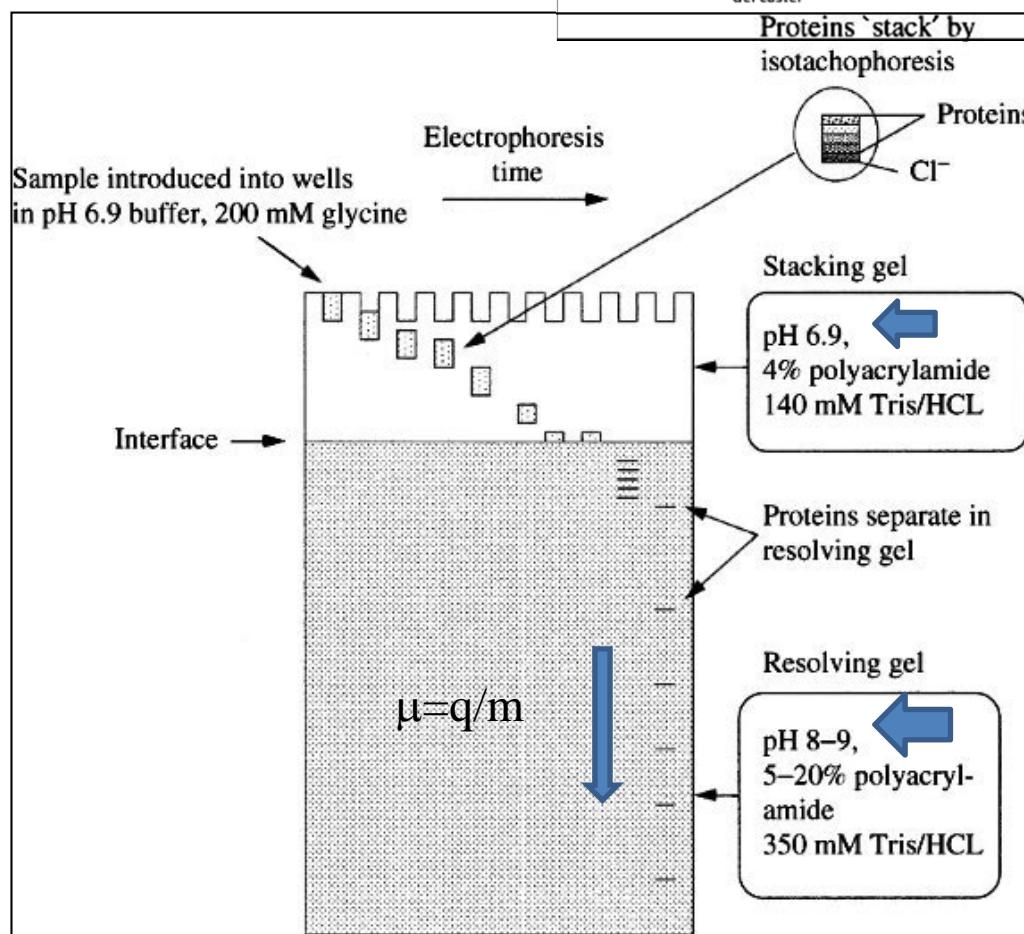
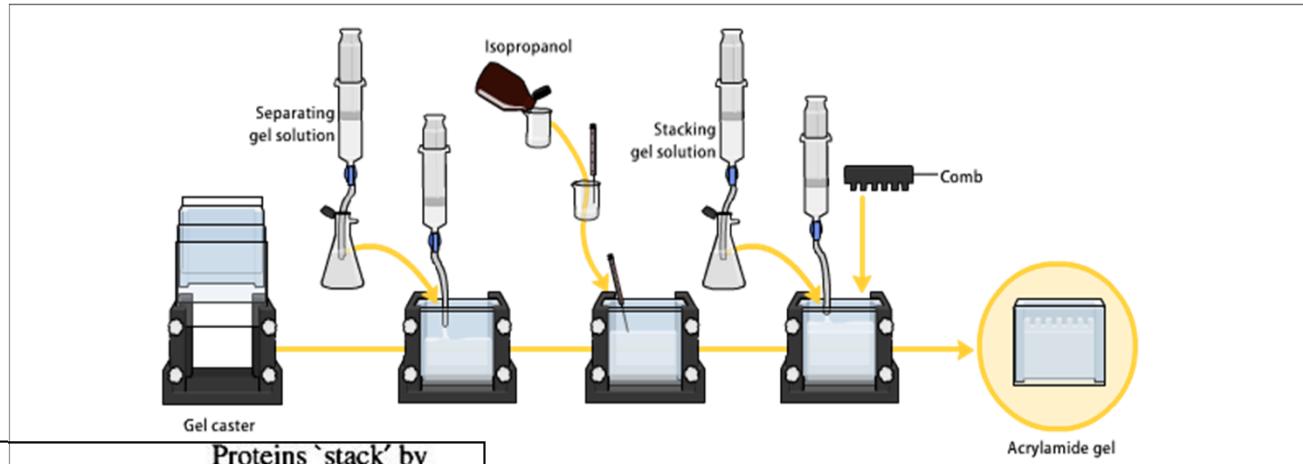


## ELECTROFERESIS EN GELES DE POLIACRILAMIDA: PAGE

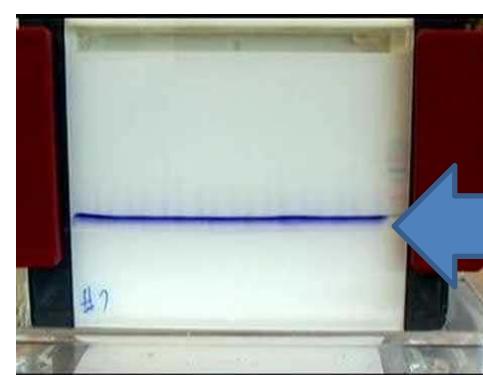
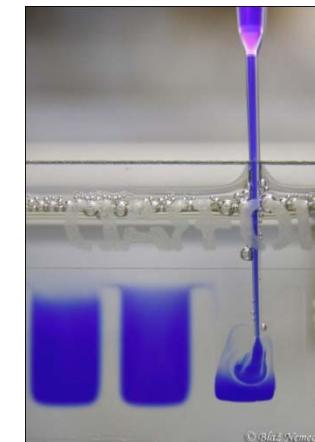
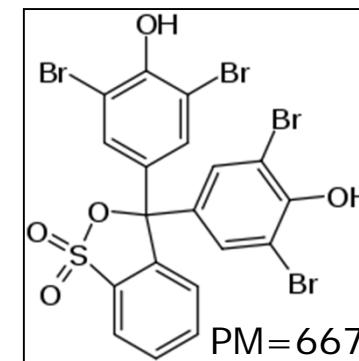


**CATALIZADORES DE LA POLIMERIZACION**  
 TEMED = NNN'N' TETRAMETILENEDIAMINA  
 PERSULFATO DE AMONIO

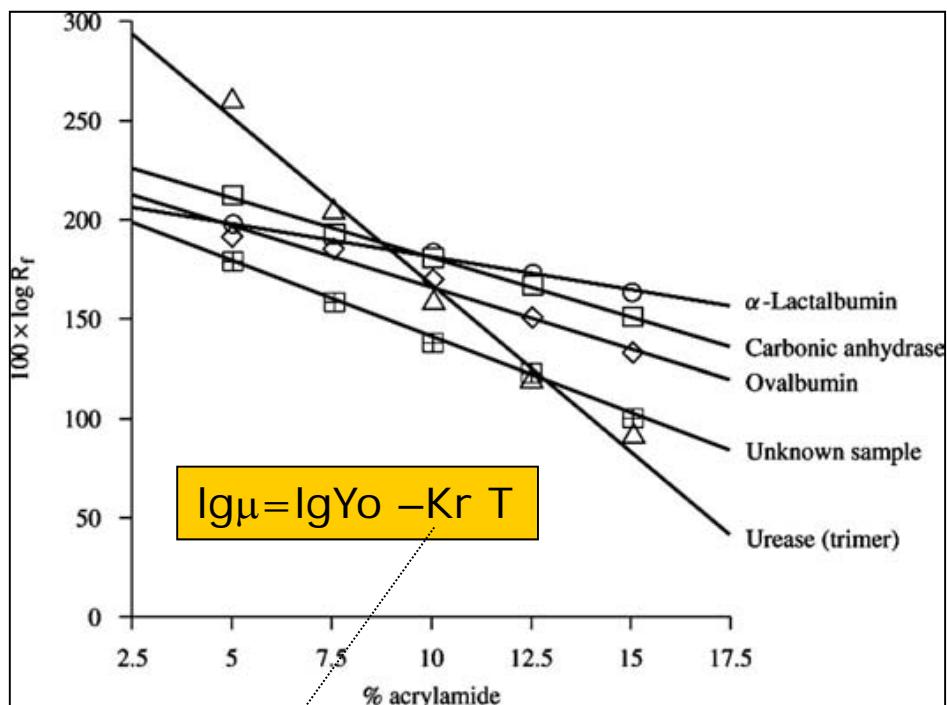
**GELES  
DISCONTÍNUOS:  
STACKING GEL(pH6)  
RESOLVING GEL(pH8)**



Azul de bromofenol + glicerina (densidad)



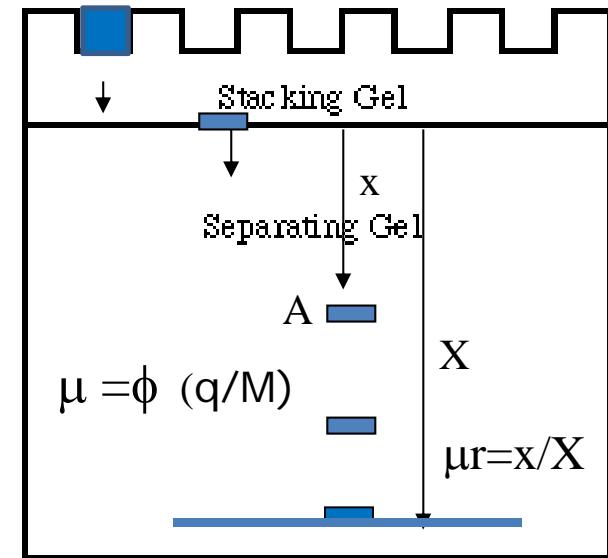
## ELECTROFORESIS NO DESNATURALIZANTE



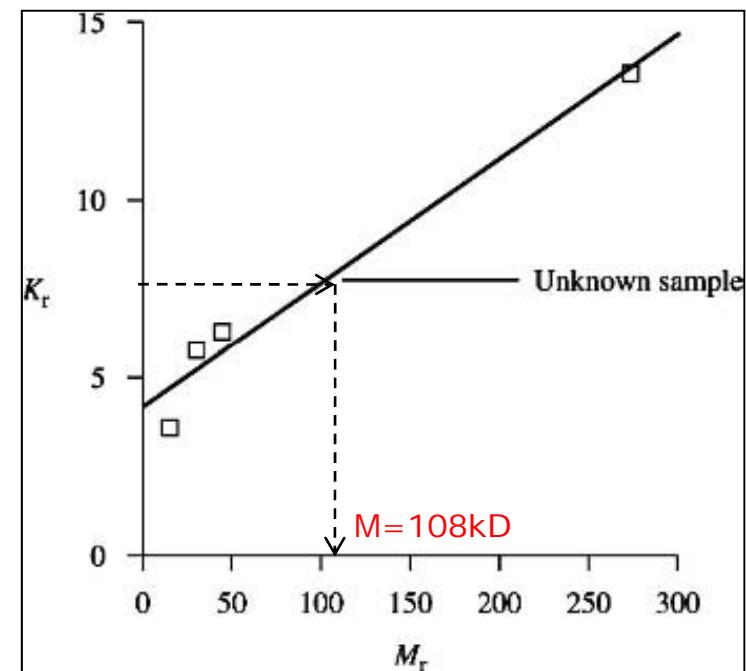
A más % de acrilamida, para una proteína menos movilidad, menos  $R_f$

### Representaciones de Ferguson

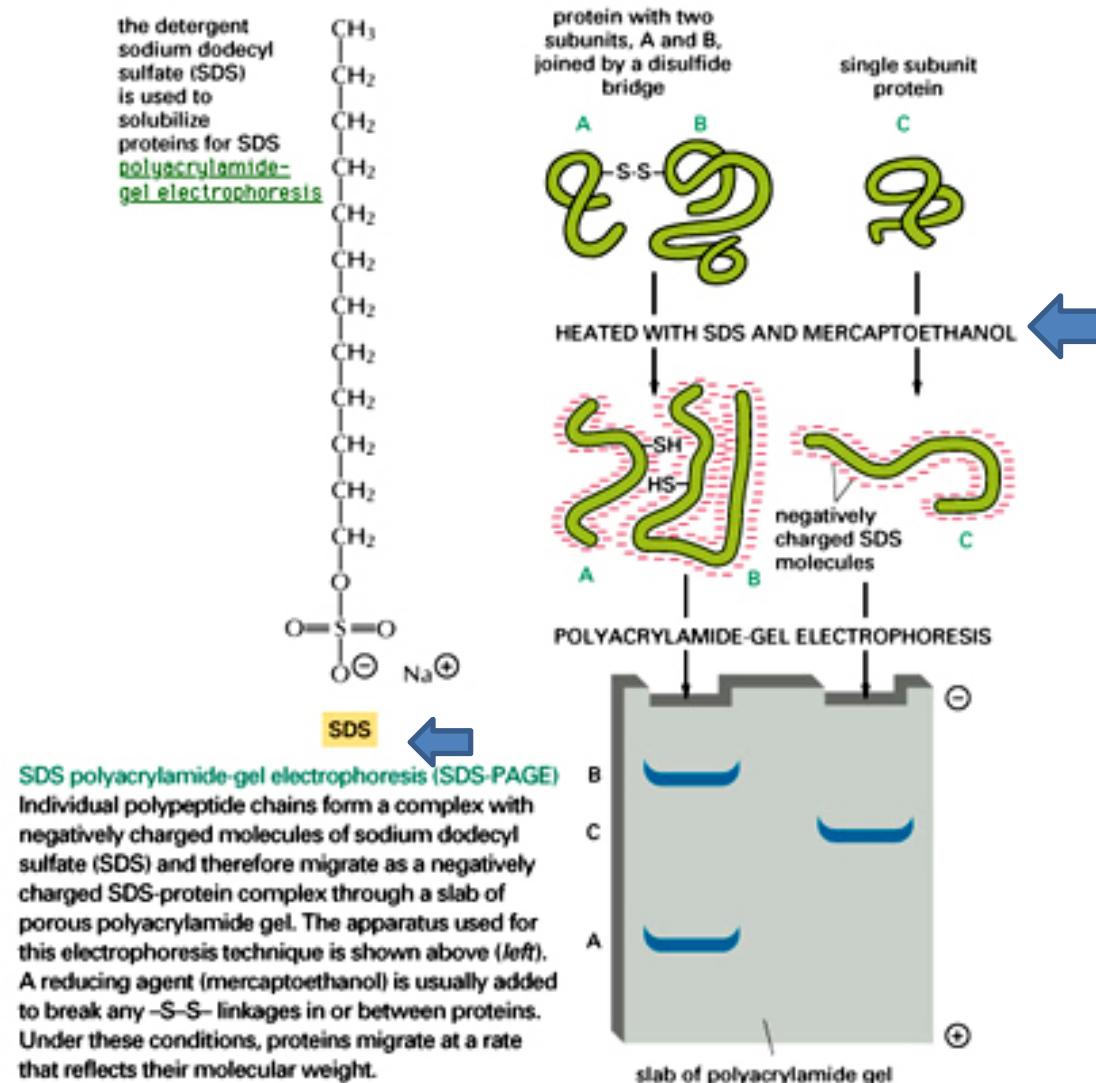
$K_r$  (coeficiente de retardo) =  $\phi$  (radio molecular)



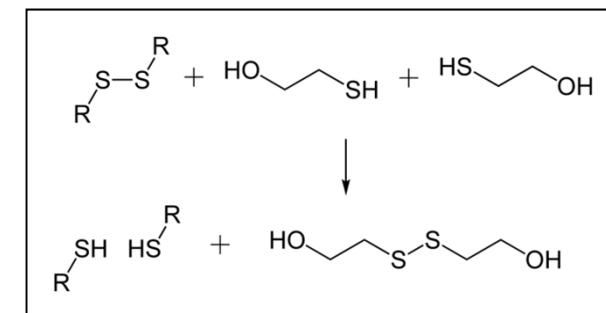
272kD



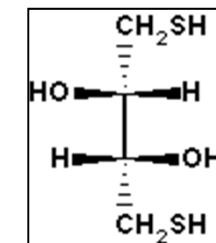
## ELECTROFORESIS DESNATURALIZANTE SDS-PAGE



$\mu = \phi Q/M \rightarrow$  no desnaturizante  
 $\mu = \phi 1/M \rightarrow$  con SDS

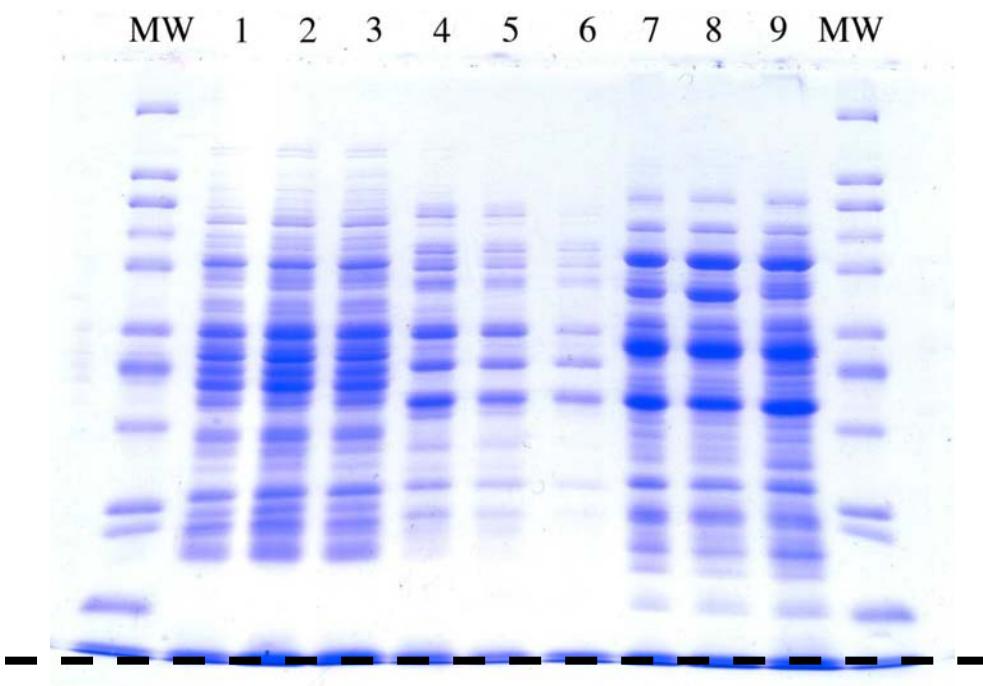
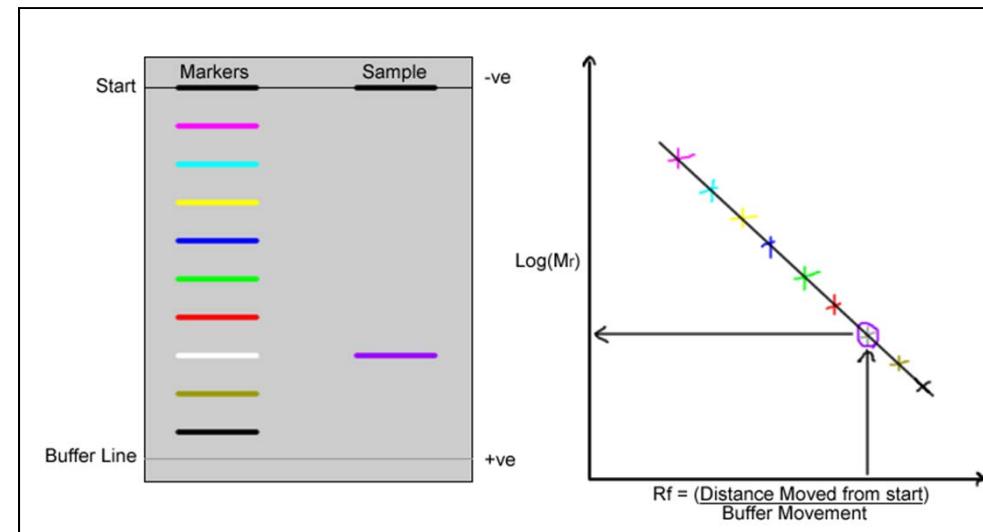
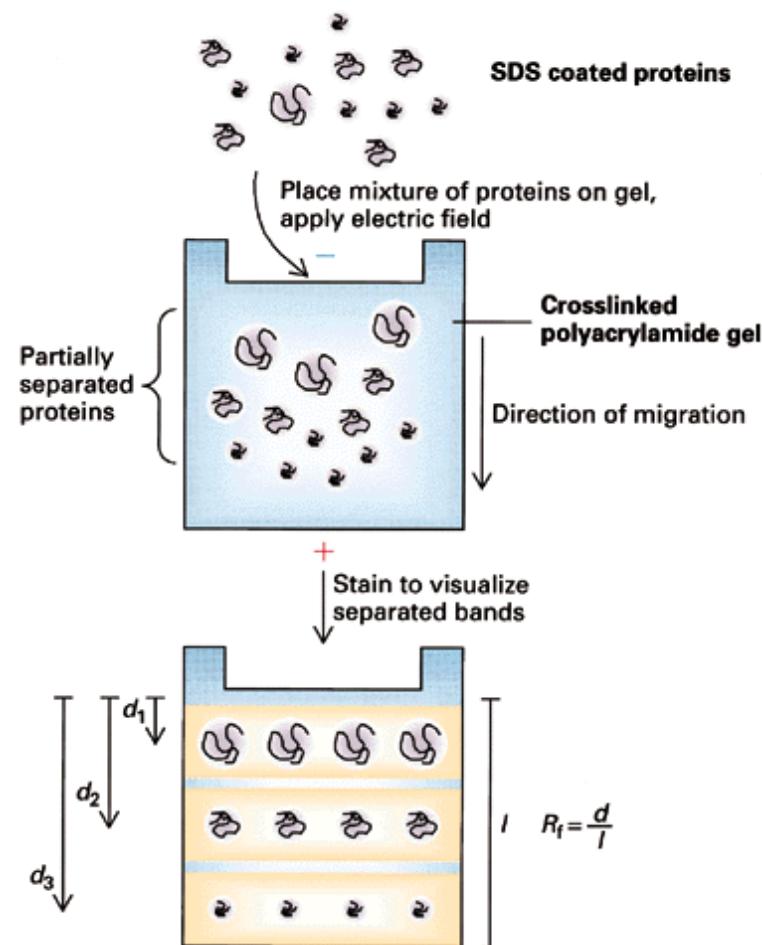


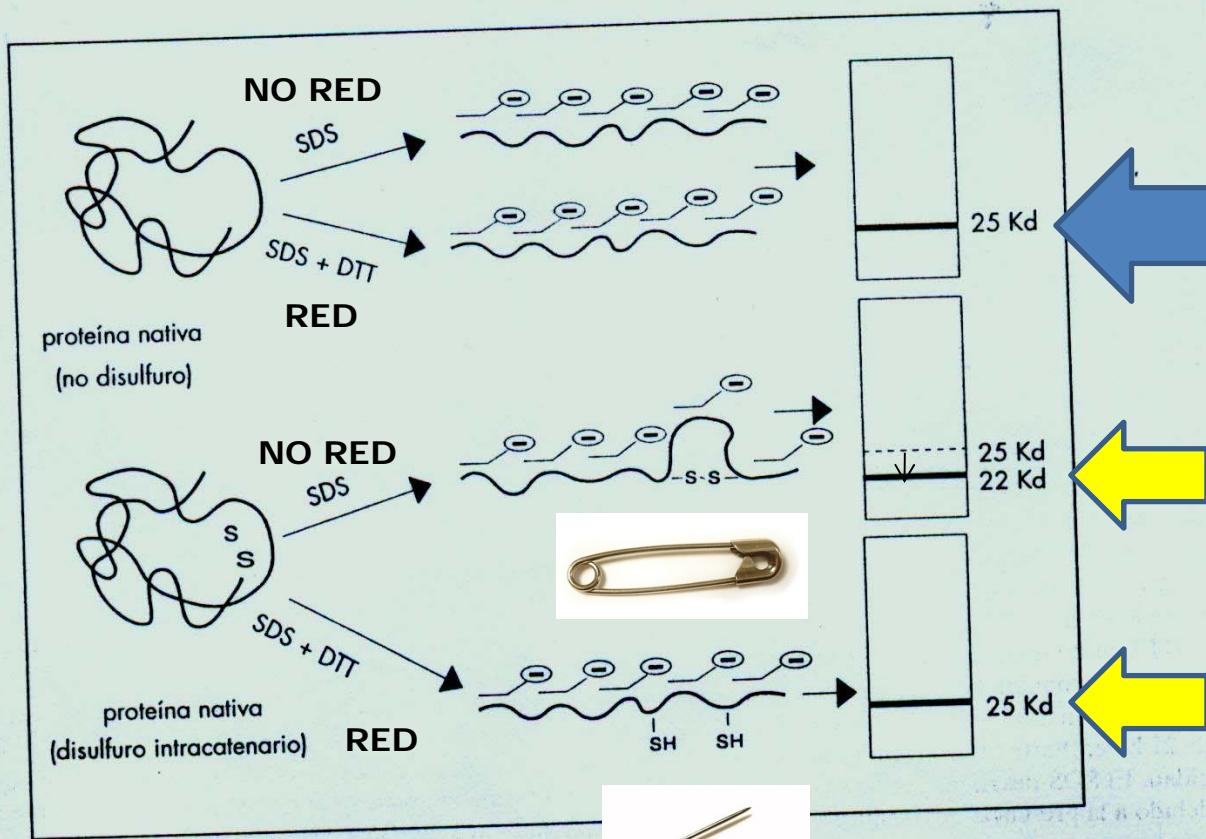
Beta.mercaptoethanol



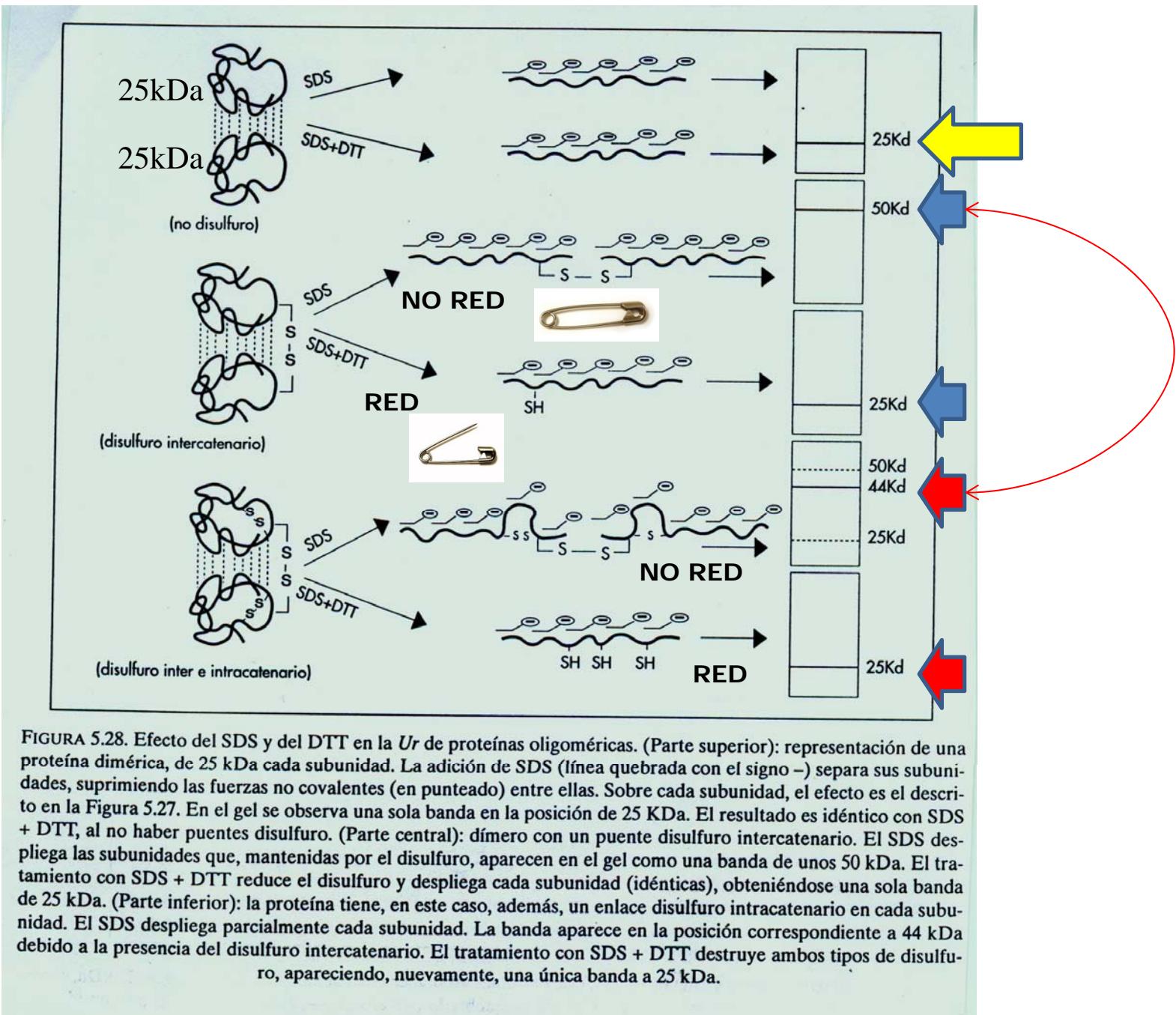
Ditiotreitol  
DTT

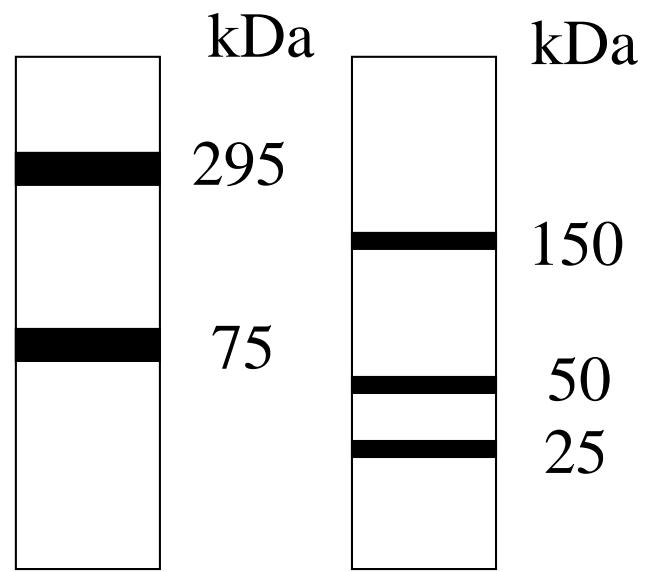
AGENTES REDUCTORES





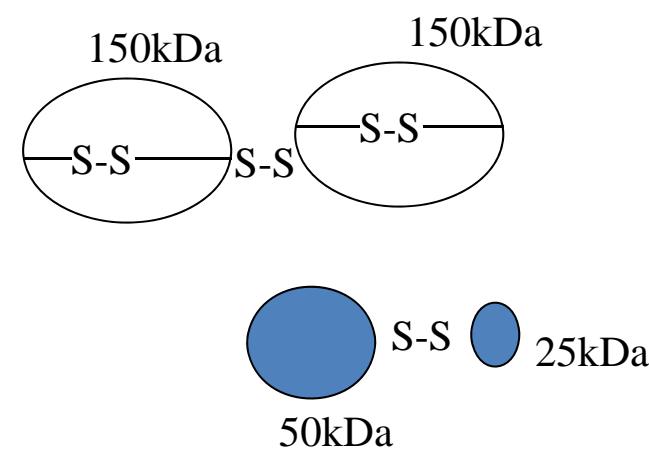
**FIGURA 5.27. Efecto del SDS y el DTT en la *Ur* de proteínas mediante puentes disulfuro.** El tratamiento con SDS (línea quebrada) o SDS + DTT, producen idéntico resultado: una banda en el gel (rectángulo vertical) en la posición correspondiente a la M de la proteína, 25 kDa. El SDS despliega la proteína convirtiéndola en una molécula aproximadamente lineal. La proteína con un disulfuro intracatenario, en presencia de SDS, se desnaturiza parcialmente, manteniendo sin desplegar la zona que abarca el disulfuro. La proteína se comporta como si fuera de menor tamaño, apareciendo en la posición correspondiente a un valor de M de 22 kDa. Cuanto mayor sea el segmento englobado por el disulfuro (cuanto más distantes estén en la estructura primaria las Cys que lo forman), mayor es el cambio observado en la posición de la banda. En trazo discontinuo se señala la posición de 25 kDa. El tratamiento con SDS + DTT rompe el enlace disulfuro y permite el despliegue total de la proteína, que aparece en la posición correspondiente a 25 KDa. El cambio de posición de la banda, con ambos tratamientos, permite identificar la presencia de uno (o más) puentes disulfuro intracatenarios.



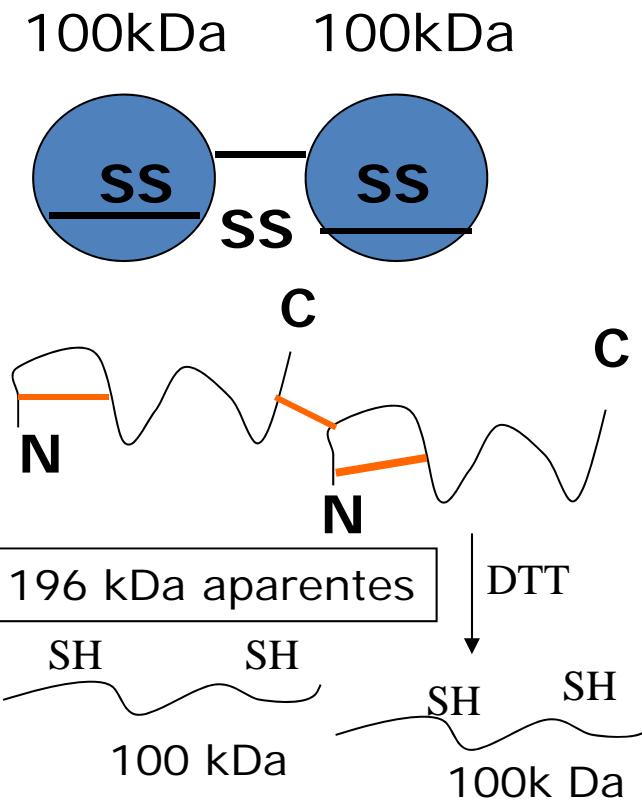
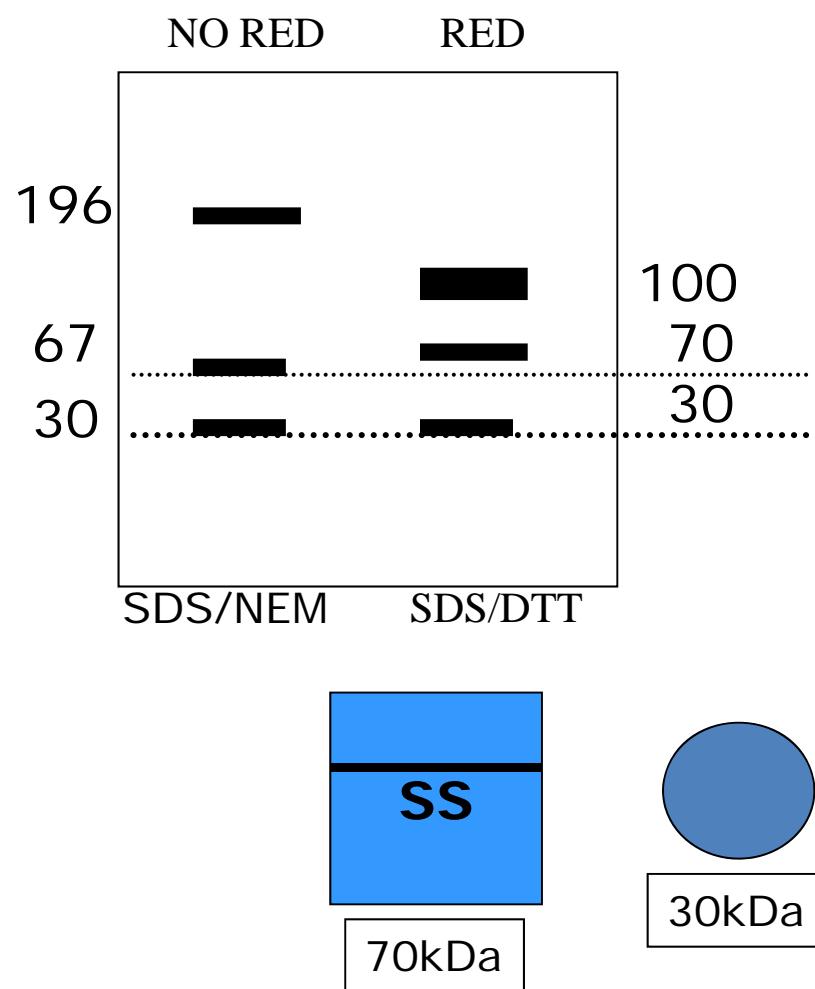


A  
SDS

B  
SDS+DTT



En una muestra de proteínas purificada se obtienen los resultados esquematizados tras la electroforesis en SDS-poliacrilamida. En la condición 1 no reductora: 3 bandas de 196, 67 y 30 kDa y en la condición 2 en condiciones reductoras, en presencia de ditiotreitol (DTT) 3 bandas de 100, 70 y 30 kDa  
 ¿Qué características podrían deducirse sobre las proteínas de la mezcla?





### **SOLUCIONES PARA ELECTROFORESIS MONODIMENSIONAL 2 PLACAS**

	7.5%	10%	12.5%	15%
AGUA DESTILADA	21	17.4	14	10.4
LOWER BUFFER	10.5	10.5	10.5	10.5
ACRILAMIDA (30:1)	10.5	13.8	17.3	20.8
UREA añadir 15,12g y llevar con agua hasta 42ml				
TEMED.....	22 $\mu$ l			
PERSULFATO.....	137 $\mu$ l			

#### **RUNNING BUFFER**

TRIS .....6 g  
 GLICINA .....28.75 g  
 AGUA DESTILADA... 2 litros  
 SDS .....2 g

#### **UPPER BUFFER**

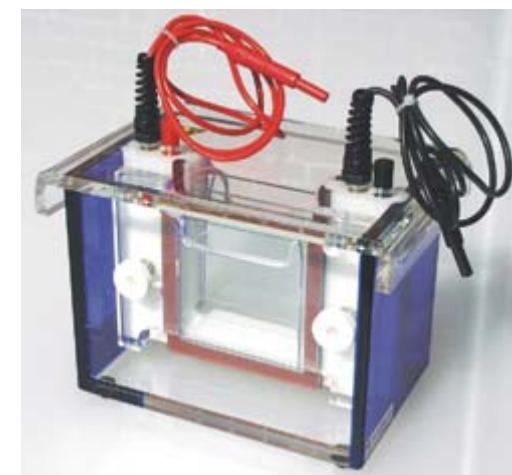
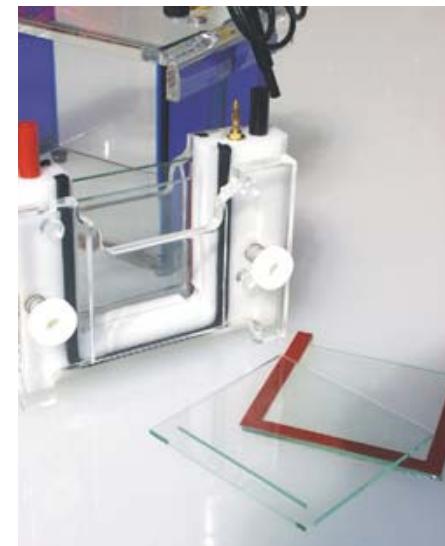
TRIS .....60.54 g (0.5M) pH 6.8  
 AGUA DESTILADA...1 litro  
 SDS.....4 g

#### **LOWER BUFFER**

TRIS.....181.1 g (1.5M) pH 8.8  
 AGUA DESTILADA.. 1 litro  
 SDS.....4 g

#### **WESTERN BLOT BUFFER**

TRIS.....9 g  
 GLICINA.....42 g  
 AGUA DESTILADA.....2.5 litros  
 METANOL.....500 ml



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Buscar

IBMC Universidad Miguel Hernández

Técnicas Instrumentales Básicas

Práctica 5: Electroforesis en geles de poliacrilamida (PAGE).

1.- INTRODUCCIÓN

Muchas moléculas importantes en Bioquímica, tales como aminoácidos, péptidos, proteínas y ácidos nucleicos, poseen grupos ionizables que en disolución se encuentran en forma de especies cargadas eléctricamente (negativa o positivamente). Las moléculas que tengan cargas similares poseerán diferentes relaciones carga/masa ( $q/m$ ) debido a las inherentes diferencias de peso molecular. Estas diferencias constituyen la base para la migración diferencial de dichas moléculas cargadas, cuando se someten a la acción de un campo eléctrico.

La **electroforesis en geles de poliacrilamida (PAGE)** se utiliza mayoritariamente para la separación de proteínas, aunque también puede ser útil para ácidos nucleicos. Se preparan de modo que sus poros sean de un tamaño comparable al de las proteínas, de manera que produzcan un efecto de tamizado molecular; la separación electroforética depende entonces de la densidad de carga de las moléculas y de su tamaño, por lo que dos proteínas con idéntica densidad de carga, pero de tamaño diferente pueden ser separadas, ya que el

ES 16:20  
06/12/2011

**Gel separador 12 %****1 gel**

Agua destilada → 1,7 ml

Tampón Tris 1,5 M; pH=8,8 → 1,25 ml

Acrilamida:Bis acrilamida 30:0,8 → 2 ml

SDS 10% → 50 µl

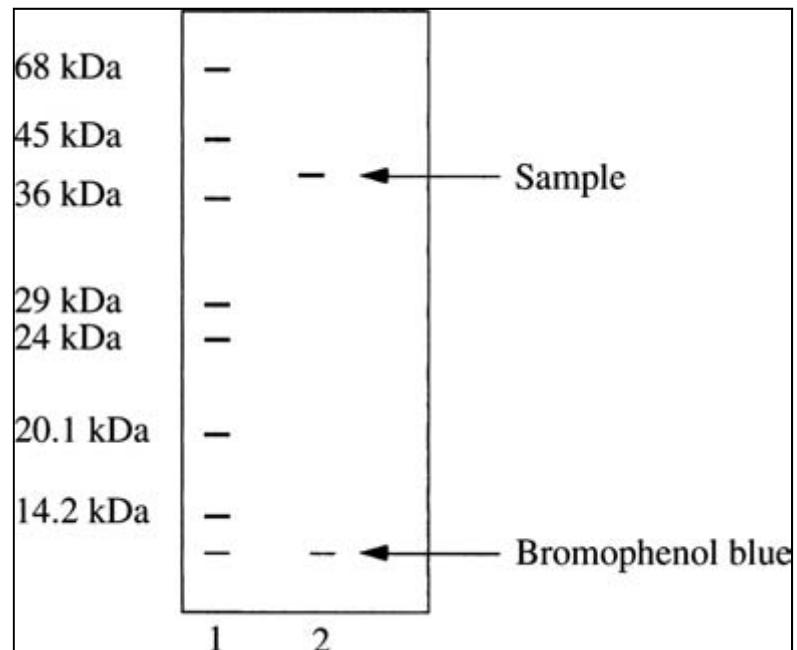
**Gel concentrador 4 %****1 gel**

Agua destilada → 1,5 ml

Tris-HCl 0,5 M; pH=6,8 → 0,625 ml

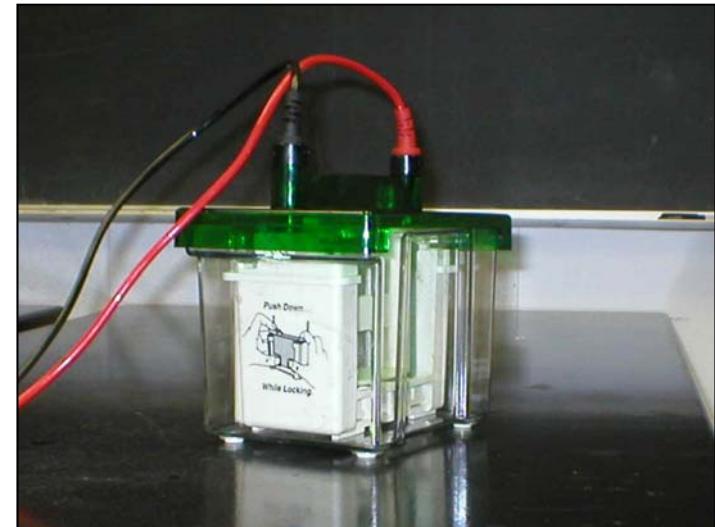
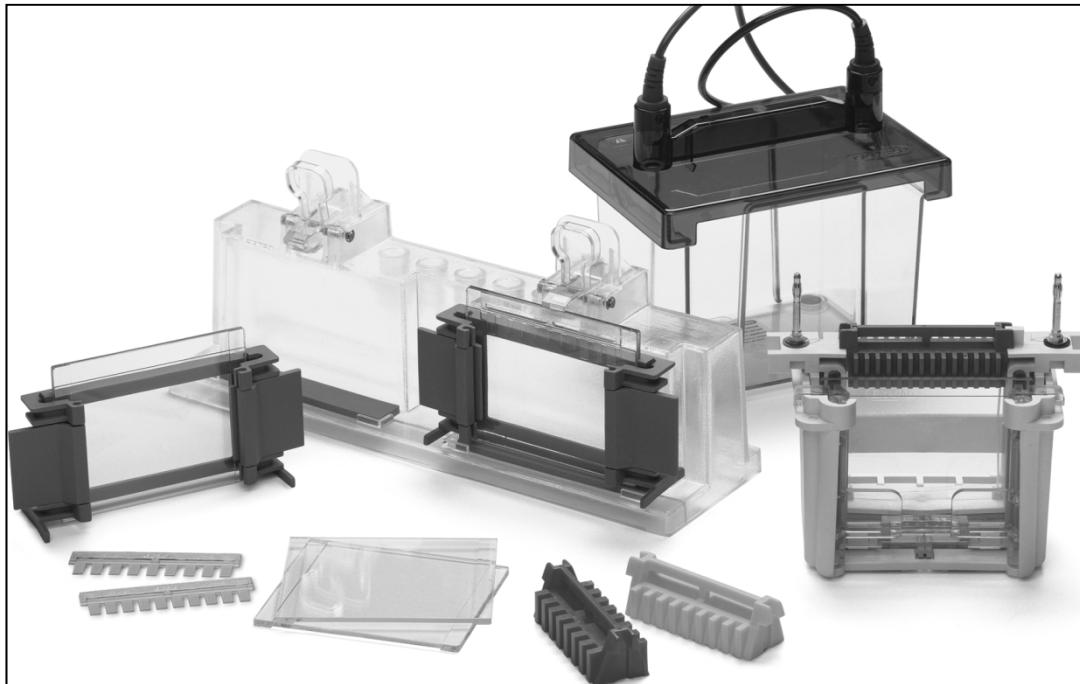
Acrilamida:Bis acrilamida 30:0,8 → 0,325 ml

SDS 10% → 25 µl

**Tampón para muestras Reductor (5X)      No reductor (5X)**

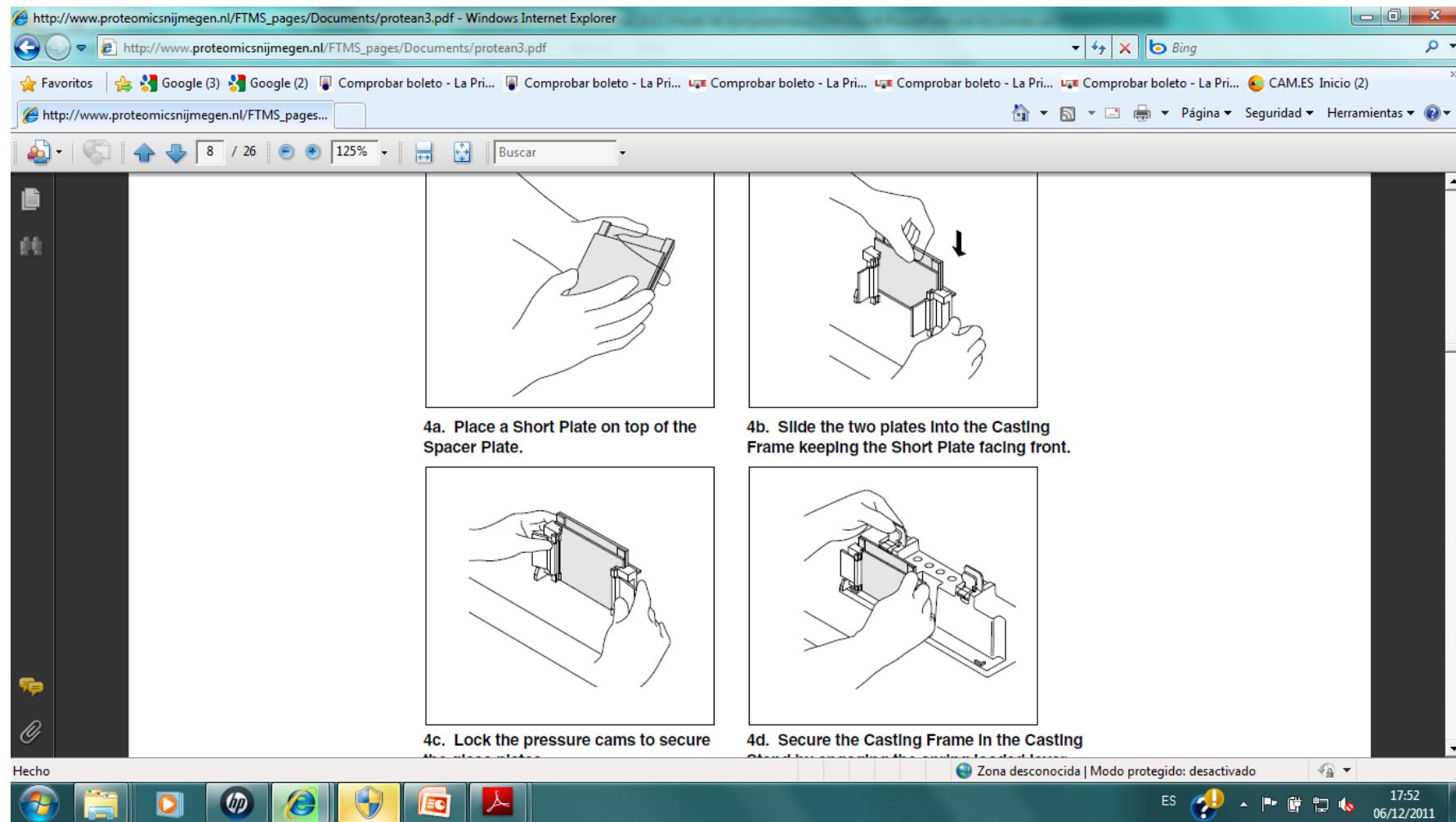
H <sub>2</sub> O	-----	5 ml
Tris-HCl 1,5 M pH 6,8	4 ml	4 ml
Glicerol 100 %	10 ml	10 ml
SDS	2 g	2 g
2-Mercaptoetanol	5 ml	-----
Azul de Bromofenol 0.1%	1 ml	1 ml





<http://www.youtube.com/watch?v=pnBZeL8nFEo>

[http://www.youtube.com/watch?v=gcJg\\_GtLgt8&feature=related](http://www.youtube.com/watch?v=gcJg_GtLgt8&feature=related)



http://www.proteomicsnijmegen.nl/FTMS\_pages/Documents/protean3.pdf - Windows Internet Explorer

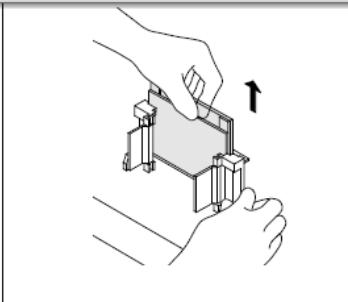
http://www.proteomicsnijmegen.nl/FTMS\_pages/Documents/protean3.pdf

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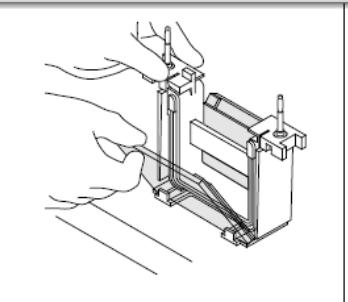
http://www.proteomicsnijmegen.nl/FTMS\_pages... Buscar

10 / 26 125% Página Seguridad Herramientas

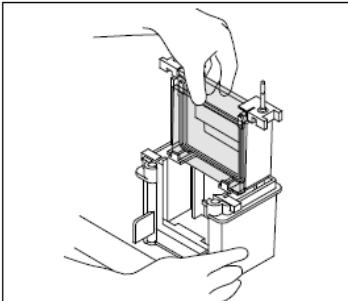
5a. Remove the Gel Cassette Sandwich from the Casting Frame.



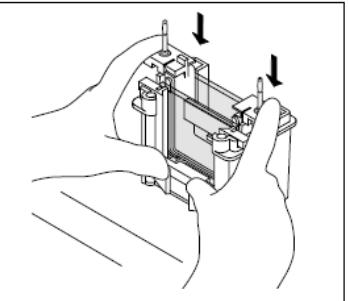
5b. Place Gel Cassette Sandwich into the Electrode Assembly with the Short Plate facing Inward.



5c. Slide Gel Cassette Sandwiches and



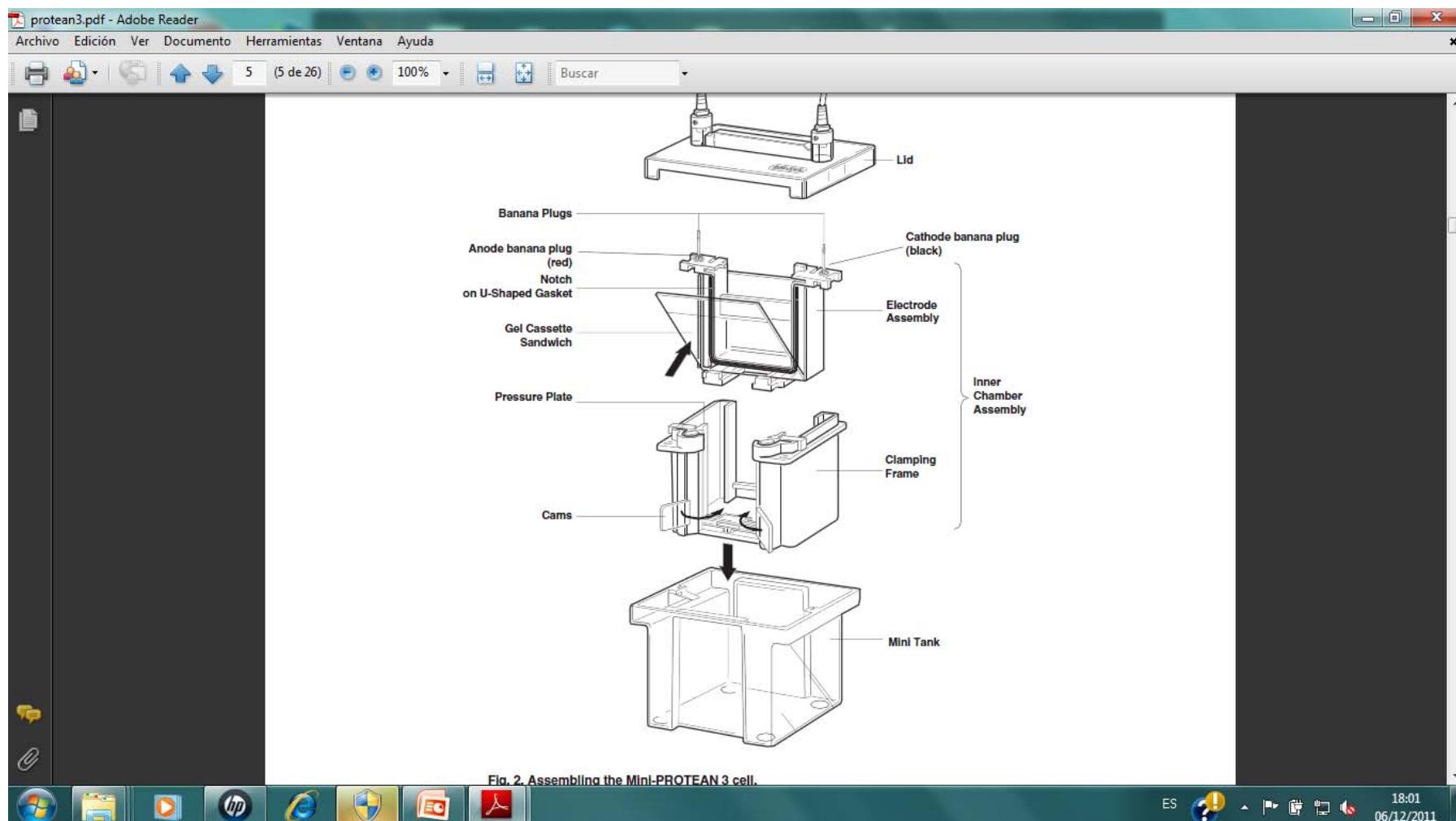
5d. Press down on the Electrode Assembly

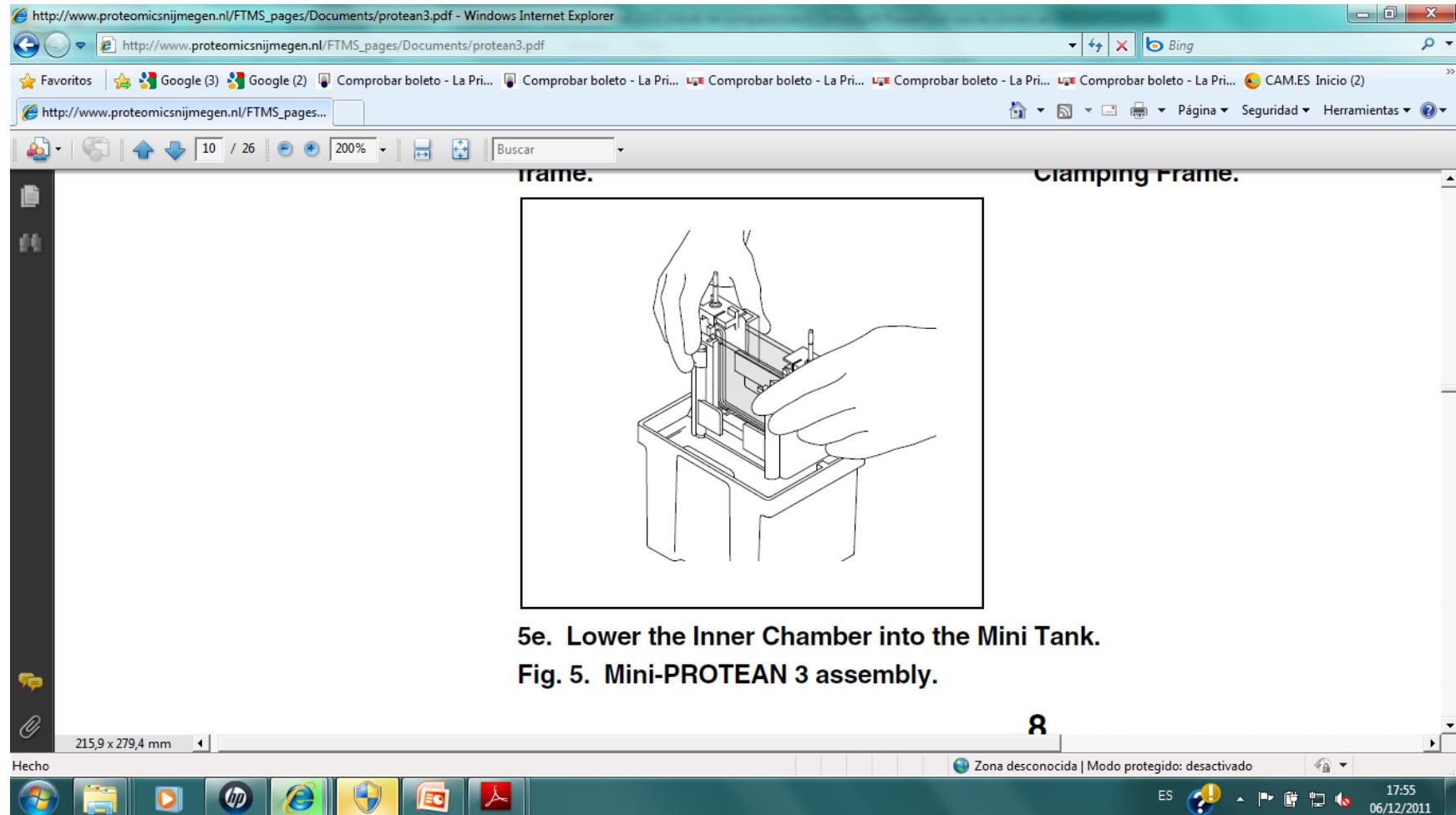


Hecho

Zona desconocida | Modo protegido: desactivado

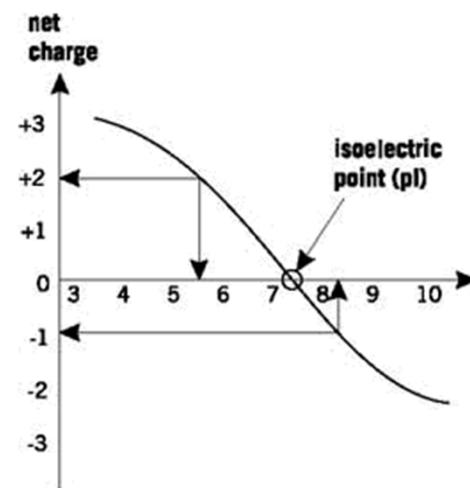
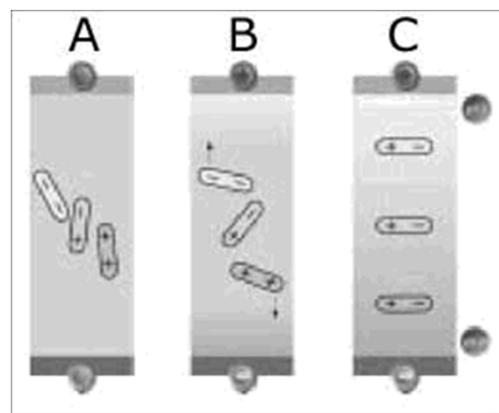
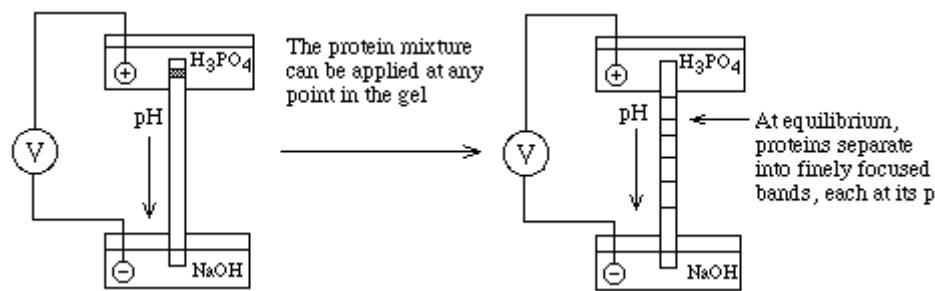
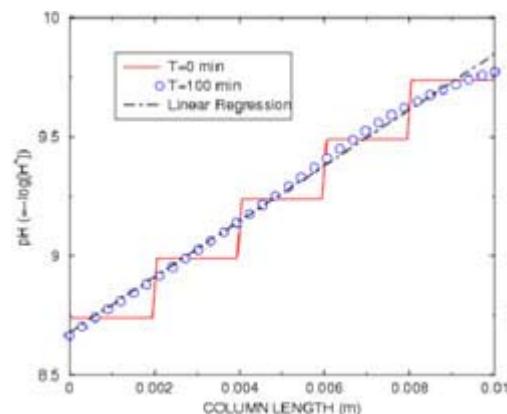
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06/12/2011

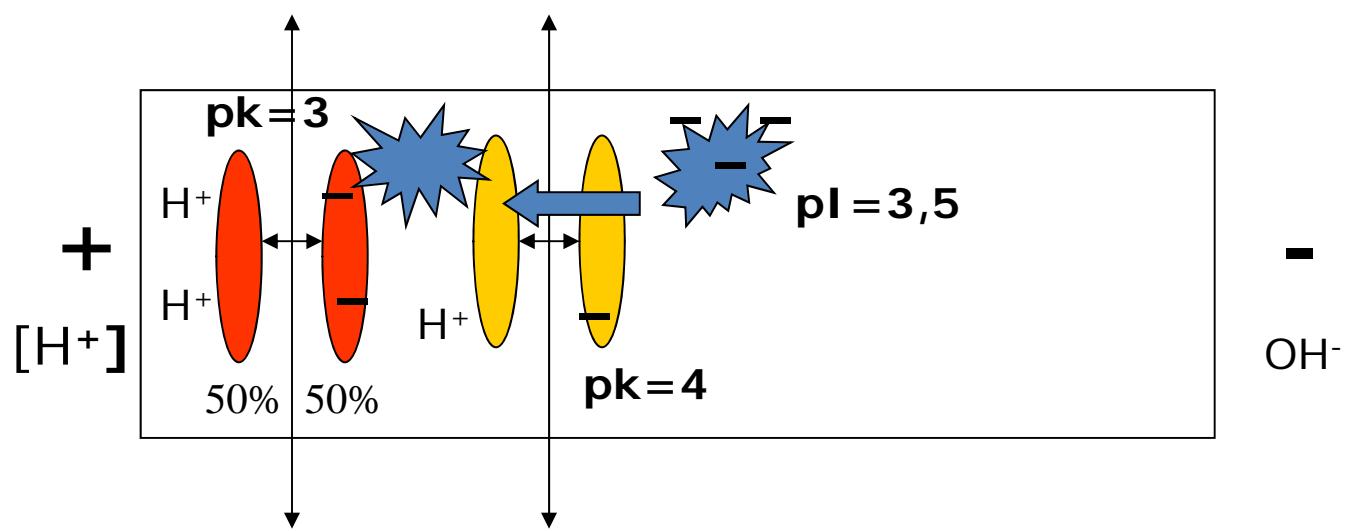
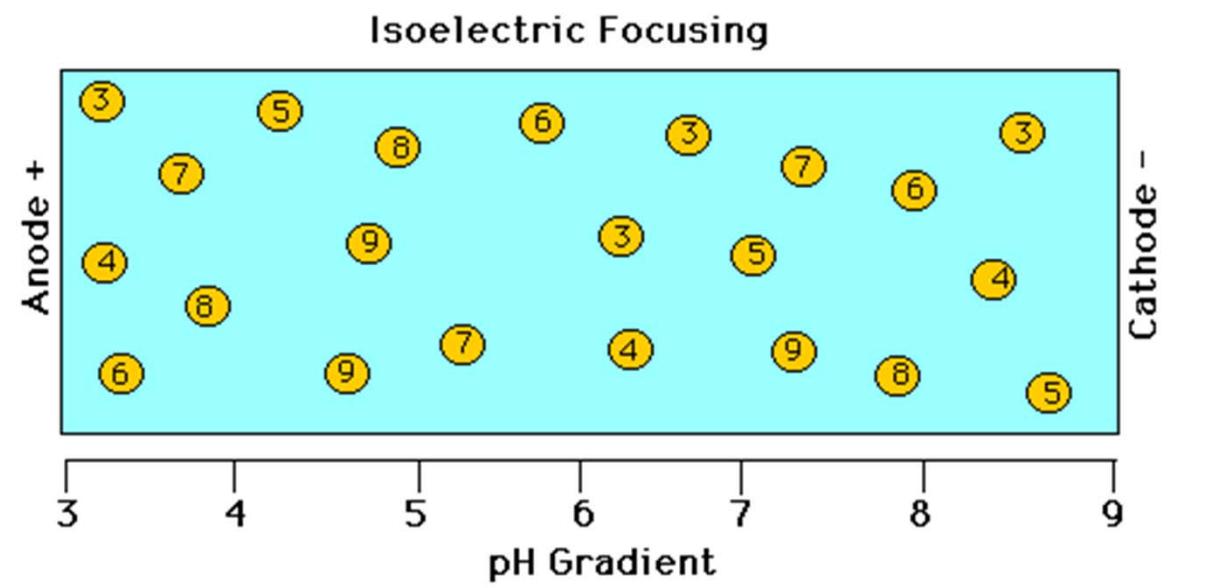


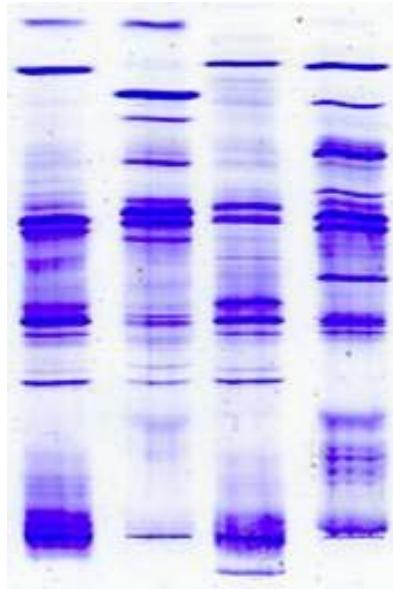


## ELECTROENFOQUE :IEF O ISOELECTRIC FOCUSING

Migración de moléculas a través de un campo eléctrico en el que existe un gradiente de pH



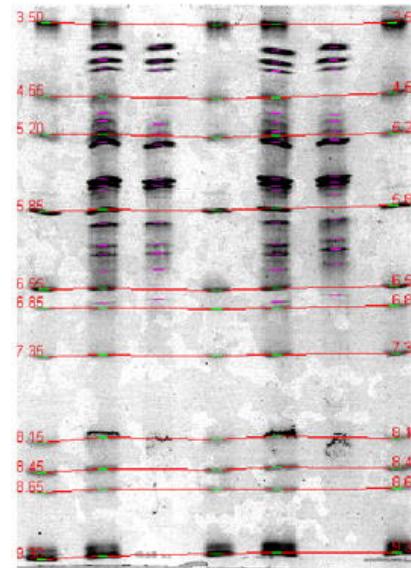




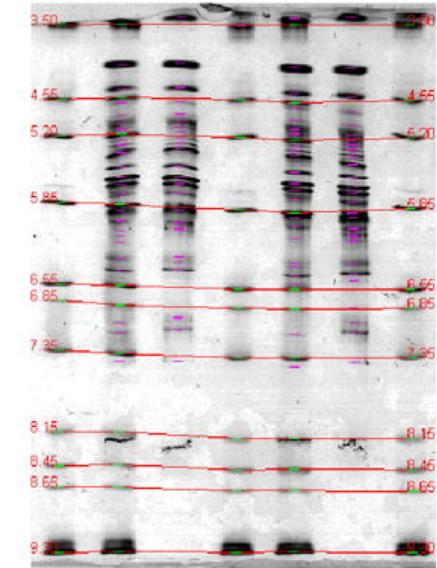
**ISOENZIMAS**  
Mismo PM  
Distinto PI



## Calibrated Gel Comparison



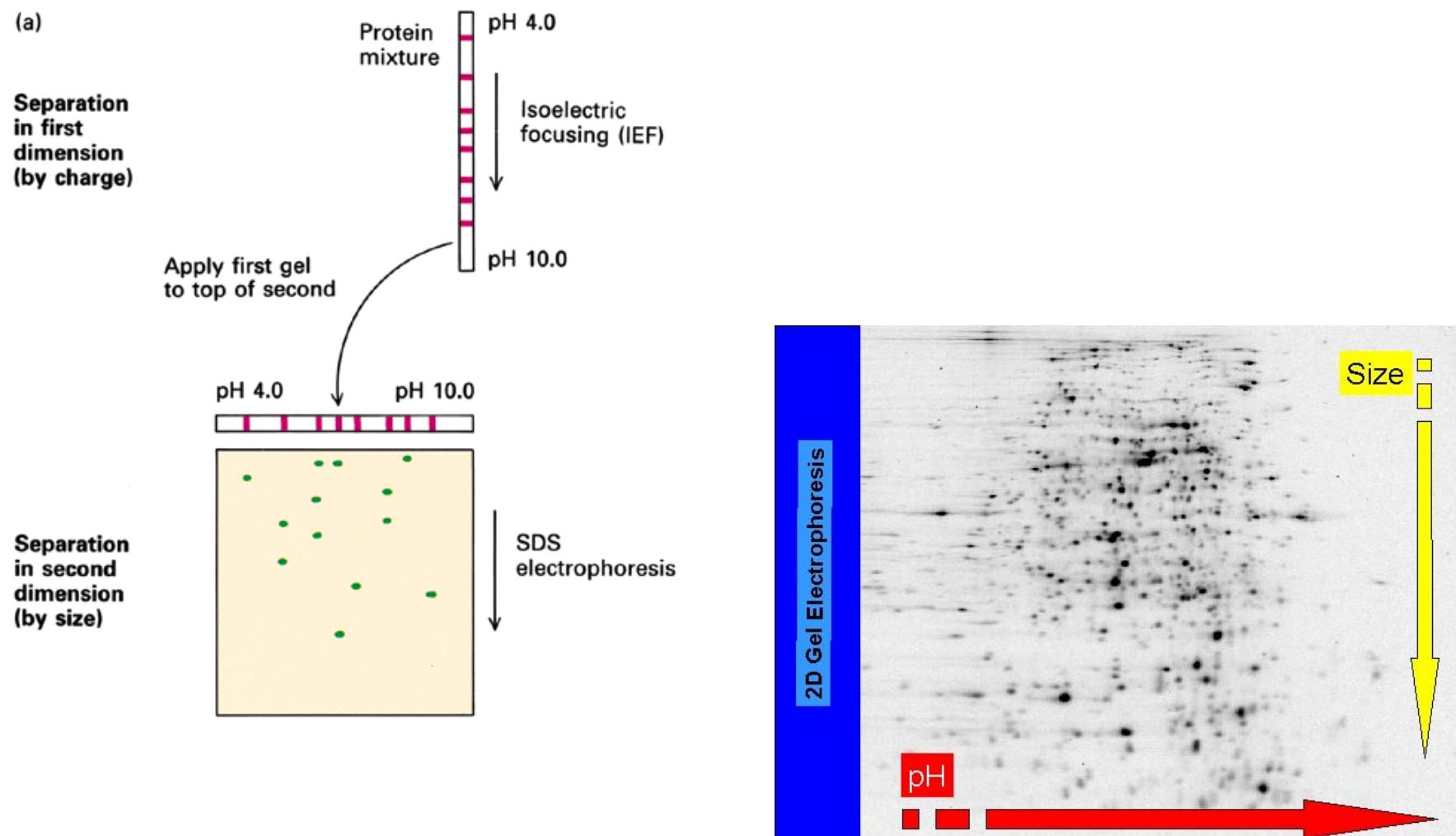
Arrowtooth Flounder  
athestom 002 #1  
081393 gel #3



English Sole  
pleuvetu 002 #1  
081793 gel #2

RFE Team: Tenge, Barnett, Savary, Rogers, Fry, Dang  
 RFE Contact: btenge@fdaem.ssw.dhhs.gov  
 ffr@fdacf.ssw.dhhs.gov  
 ndang@fdaem.ssw.dhhs.gov

## 2D PAGE: ELECTROFORESIS BIDIMENSIONAL = IEF + PAGE



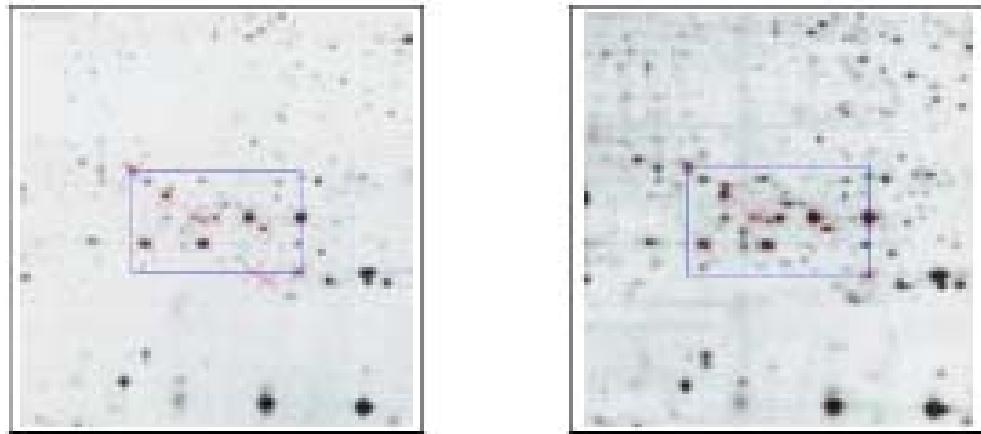
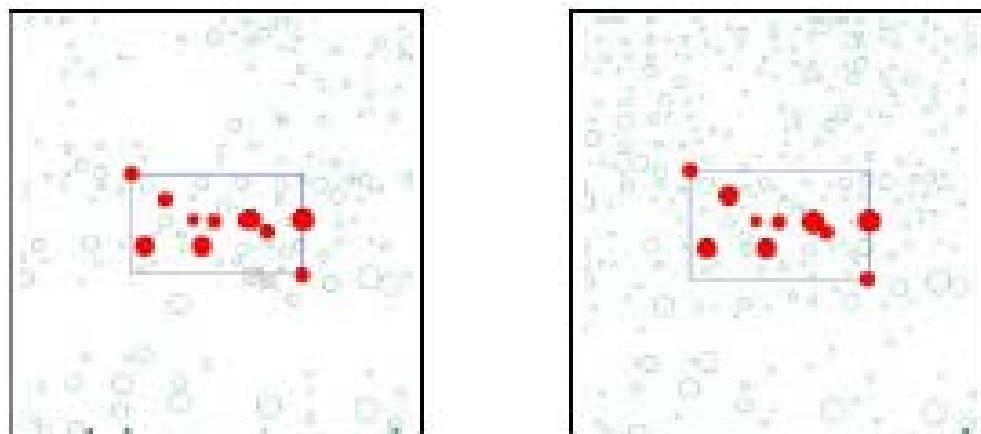
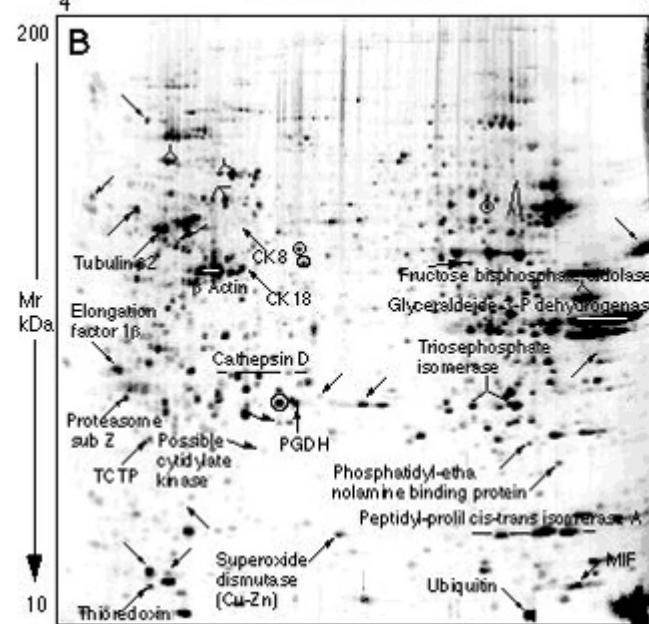
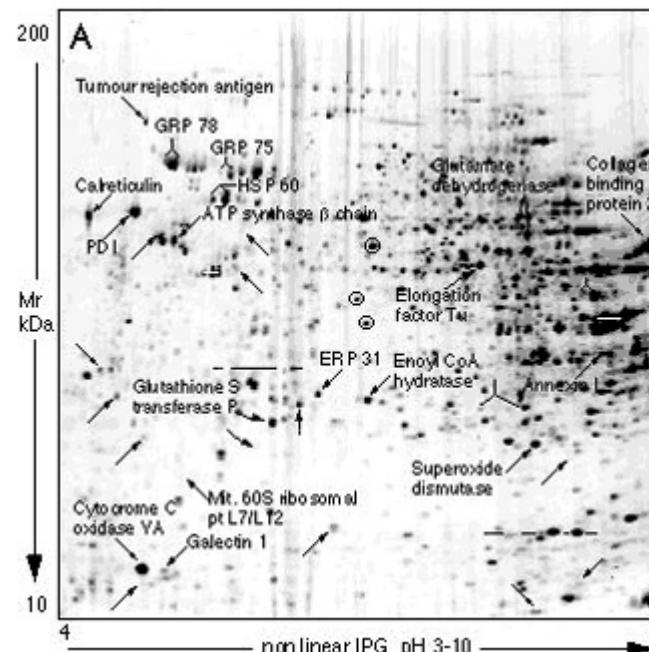
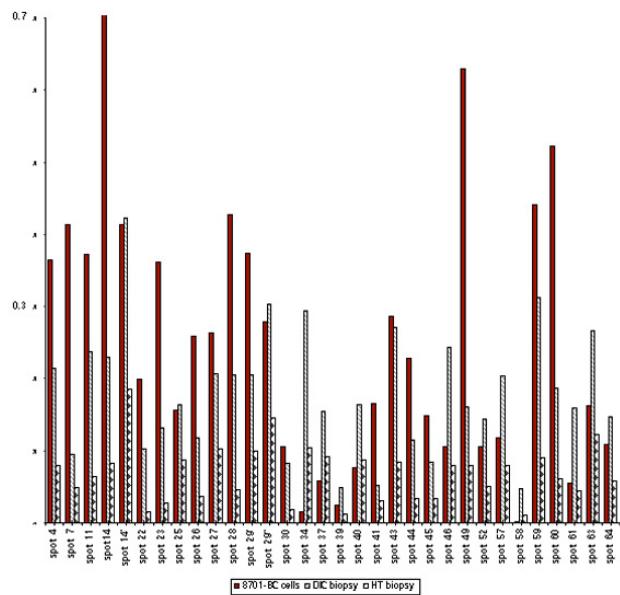
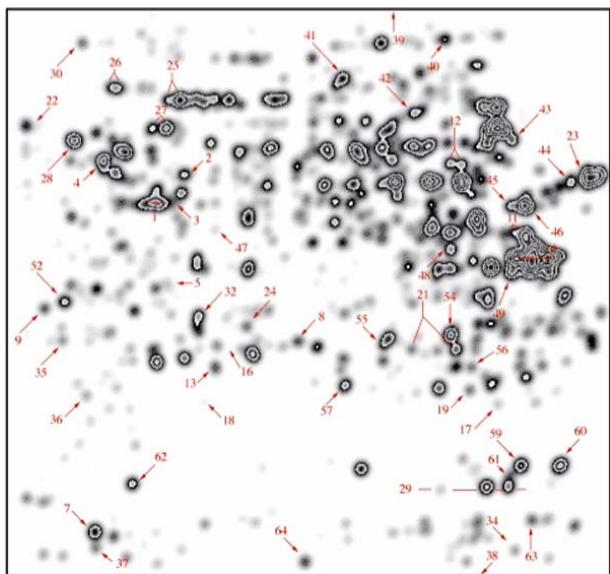
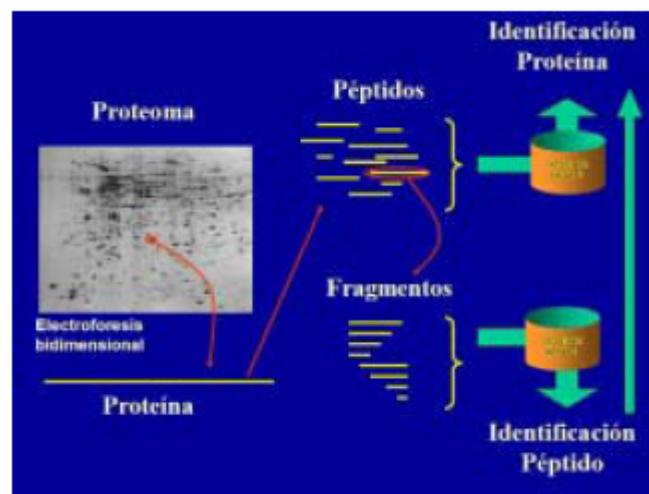
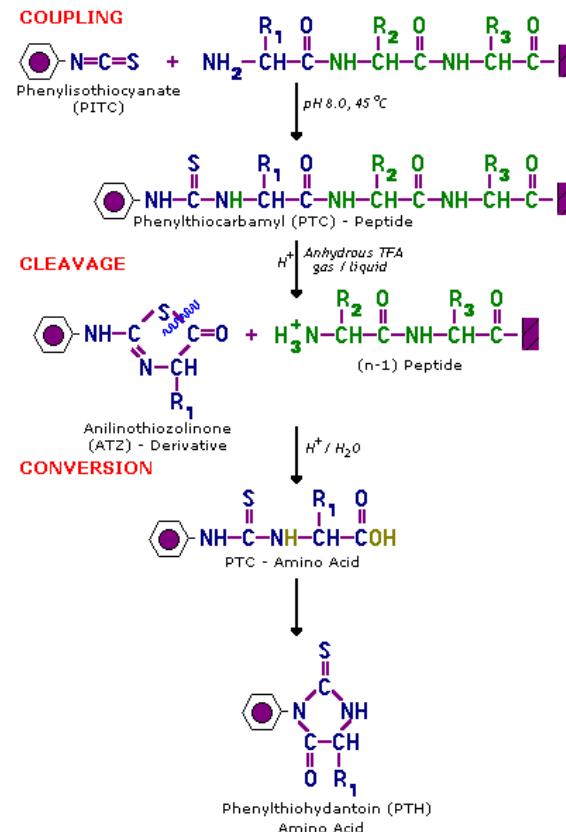
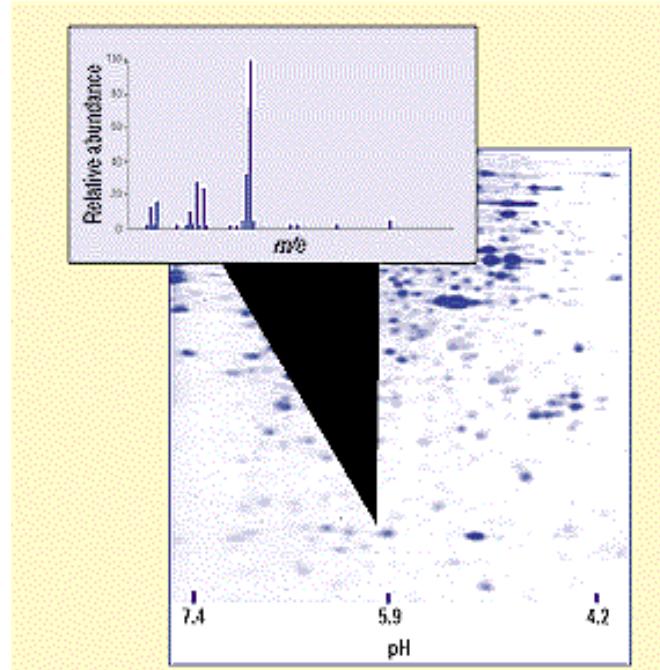


Figure 3: Detailed local images of Fig 2, a selected pattern on the left side and a partial matching.







## **DETECCION Y ANALISIS DE PROTEINAS TRAS ELECTROFORESIS**

- 1.-PREPARACION DEL GEL**
- 2.-APLICACIÓN DE LA MUESTRA**
- 3.-CORRER EL GEL (RUNNING)**
- 4.-DETECCION DE PROTEINAS**
- 5.-SECAR Y PROCESAR: DENSITOMETRIA, PROTEOMICA**

**TINCION:** DETECCION POR COLORANTES O SUSTRATOS QUIMICOS

**AUTORADIOGRAFIA:** DETECCION MEDIANTE MARCAJE CON RADIOISOTOPOS

**INMUNODETECCION:** DETECCIÓN ESPECIFICA CON ANTICUERPOS

**TINCION RAPIDA: COOMASSIE (microgramos)**  
**TINCION SENSIBLE: PLATA (nanogramos)**  
**TINCION PROTEOMICA: sypro ruby fluorescente**

## ELECTROPHORESIS

### Protein Staining Reagents

To meet the great diversity of protein analysis needs, Sigma offers a wide selection of protein visualization (staining) reagents. EZBlue™ and ProtoSilver™, designed specifically for proteomics, also perform impressively in traditional PAGE formats.

### Protein Dyes and Stains Selection Chart

		Optimize your protein detection by choosing the best reagent for your application.											
		Good Choice											
		BEST Choice											
Detection Targets	Method	Colorimetric	Fluorescent	Proteins	Glycoproteins	Lipids and/or Lipoproteins	Nucleic Acids	Phospholipids, Neutral Fats, & Sterols	PAGE	IF (Varylande)	Agarose Gel	PVDF Membrane	Cellophane Acetate Membrane
	Colorimetric	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓
	Fluorescent	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓
	Proteins	✓	✓	✓	✓	✓	★	✓	✓	✓	✓	✓	✓
	Glycoproteins	✓											
	Lipids and/or Lipoproteins							★					
	Nucleic Acids												
	Phospholipids, Neutral Fats, & Sterols												
	PAGE	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓
	IF (Varylande)	✓	✓	✓	★	✓	✓	✓	✓	✓	✓	✓	✓
	Agarose Gel	✓		★									
	PVDF Membrane	✓	★										
	Cellophane Acetate Membrane	✓	✓										
	Nitrocellulose Membrane	✓	✓										
	Hylon Membrane	✓	✓										
	MALDI-MS	✓	✓										
	High Sensitivity (10 ng)												
	Higher Sensitivity (1 ng)												
	Highest Sensitivity (0.1 ng/ml)												
	Fluor Reagent	✓	✓	✓	★	★							
	Requires Re-Destaining			★									
	Rapid Development (5 minutes)	✓							✓				
	Reversible								✓				

105

SIGMA



### Stain

Coomassie Brilliant Blue

### Sensitivity

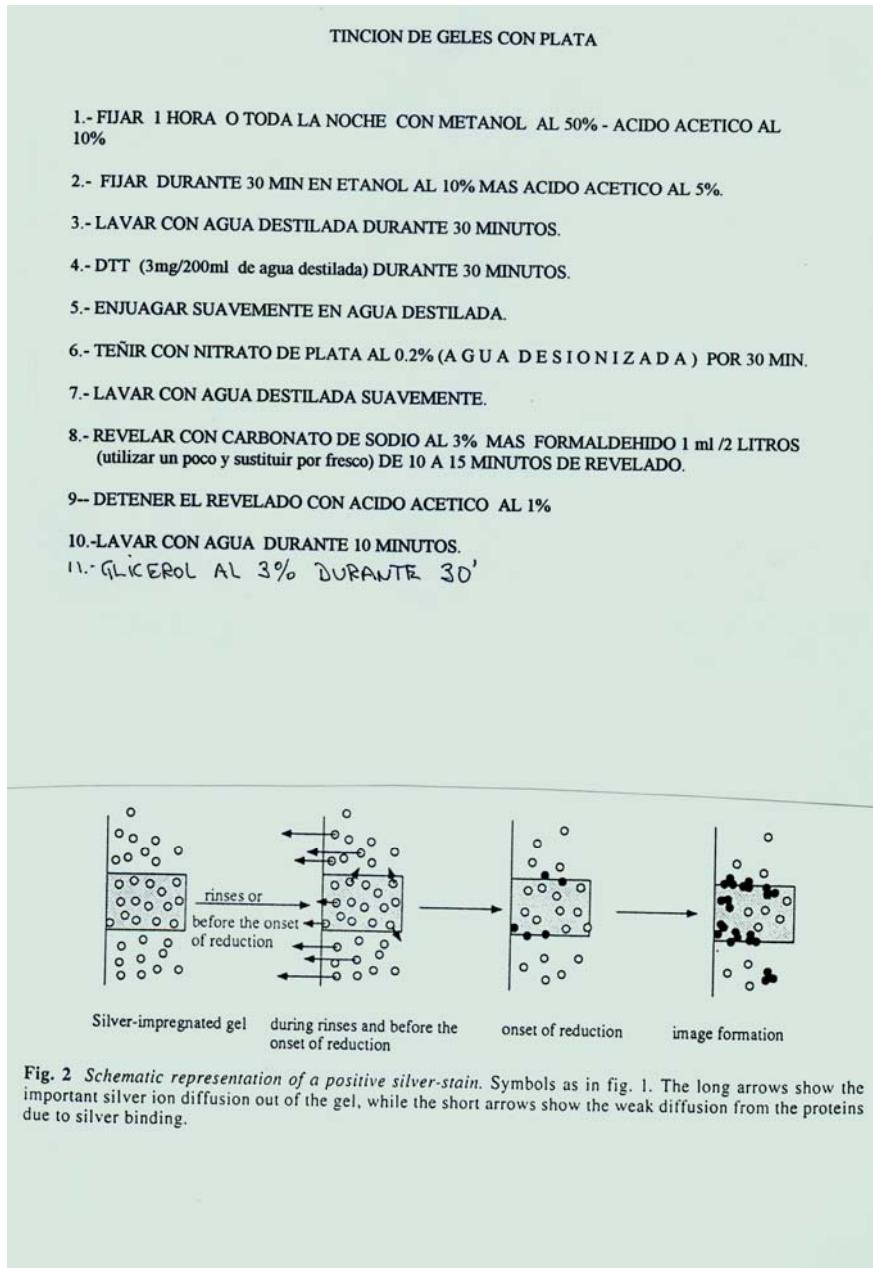
~30-50 ng

Sypro Ruby Fluorescent

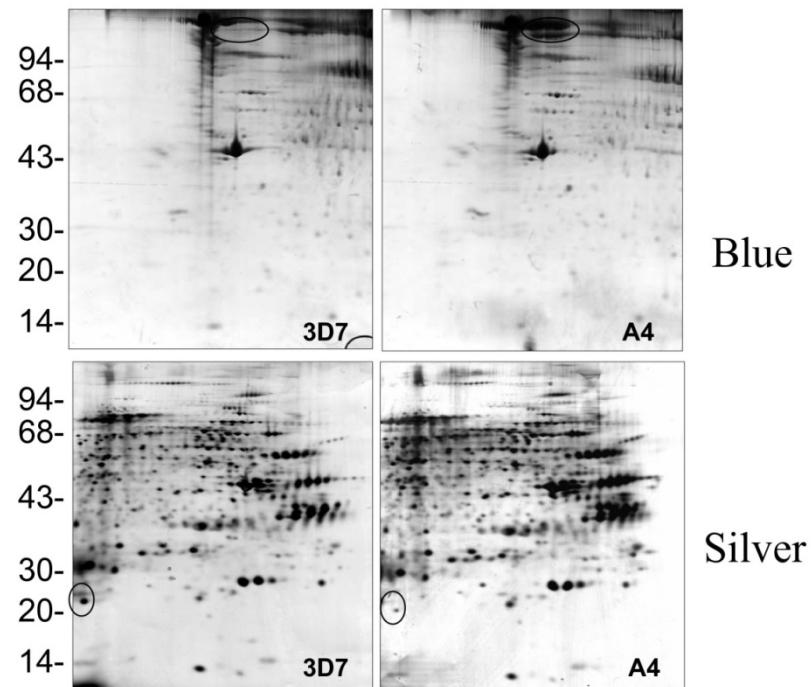
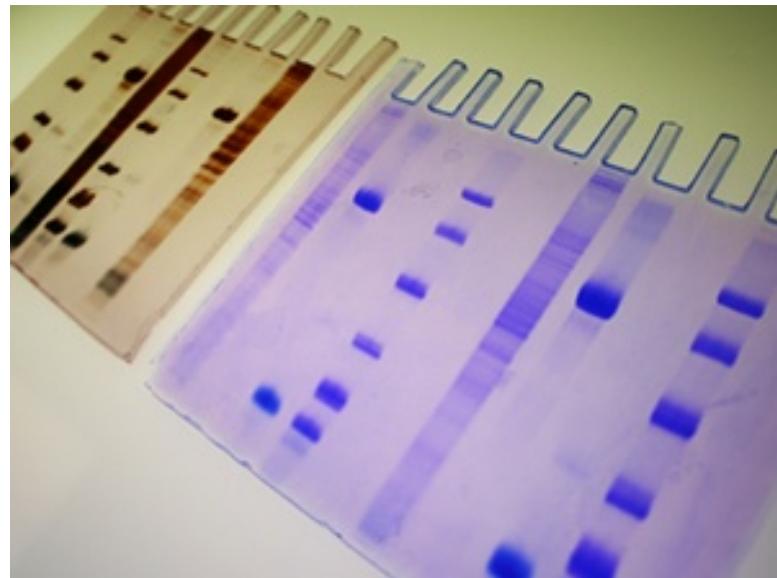
~1-10 ng

Silver Stain

~0.5 – 1.5 ng



**Fig. 2** Schematic representation of a positive silver-stain. Symbols as in fig. 1. The long arrows show the important silver ion diffusion out of the gel, while the short arrows show the weak diffusion from the proteins due to silver binding.



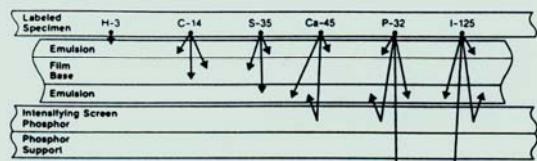
**Table 7.1. Radioisotopes Commonly Used in Autoradiography<sup>a</sup>**

Isotope	Radiation	Energy (MeV)	Half-life
<sup>3</sup> H	$\beta$	Low (0.018)	12.26 years
<sup>14</sup> C	$\beta$	Medium (0.156)	5730 years
<sup>35</sup> S	$\beta$	Medium (0.167)	88 days
<sup>45</sup> Ca	$\beta$	Medium (0.256)	165 days
<sup>32</sup> P	$\beta$	High (1.71)	14.3 days
<sup>125</sup> I	$\gamma$	High (0.035)	60 days
	X	(0.027)	
	$\epsilon$	(0.30)	

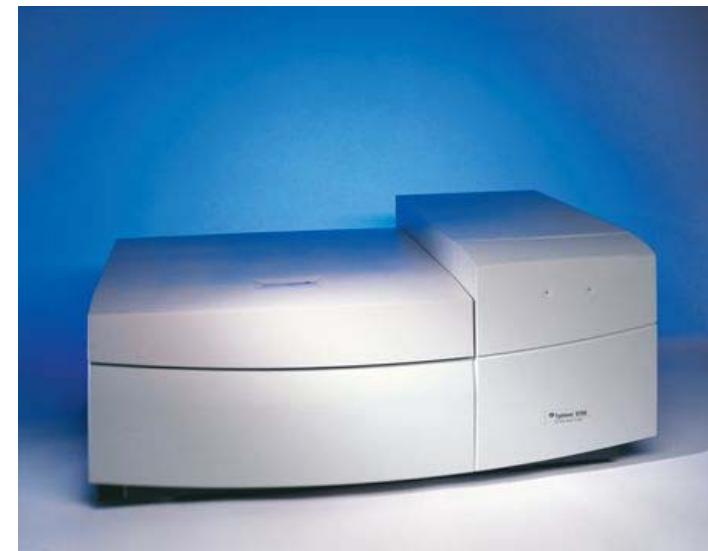
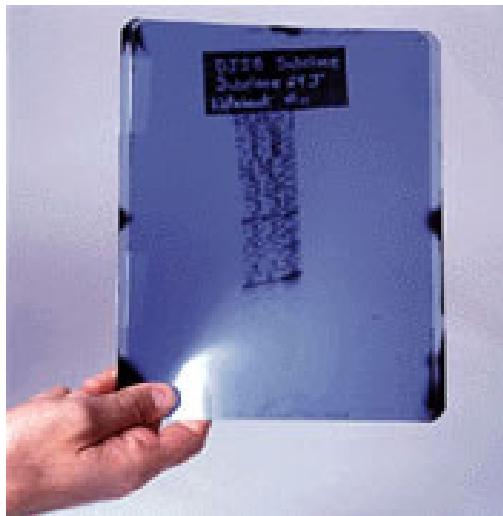
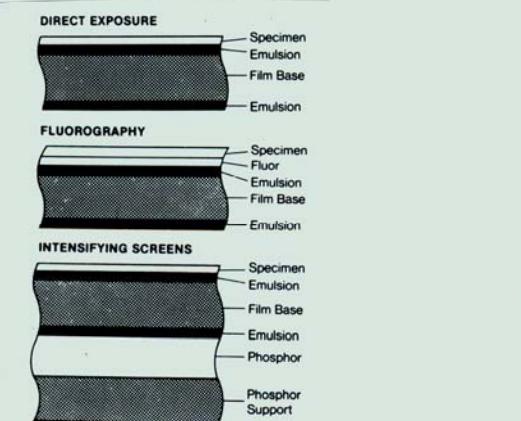
<sup>a</sup>Modified from Hahn (1983) and Freifelder (1976).

**Table 7.2. Recommended Conditions for Optimal Sensitivity of Exposure Using Different Radioisotopes for Autoradiography**

Radioisotope	Exposure	Temperature	Type x-ray film
<sup>3</sup> H	Fluorographic	-70°C	High sensitivity to UV and blue spectrum
<sup>35</sup> S, <sup>14</sup> C	Fluorographic	-70°C	High sensitivity to UV and blue spectrum
<sup>35</sup> S, <sup>14</sup> C	Direct exposure	20°C	High speed
<sup>32</sup> P	Direct exposure	20°C	High speed
<sup>32</sup> P	With intensifying screen	-70°C	High sensitivity to UV and blue spectrum
<sup>125</sup> I	Direct exposure	20°C	High speed
<sup>125</sup> I	With intensifying screen	-70°C	High sensitivity to UV and blue spectrum

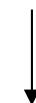


**Figure 7.2. Penetration levels of isotopes commonly used for autoradiography.** (From Hahn, 1983.)

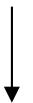
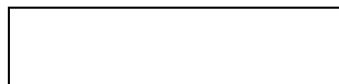


## AUTORADIOGRAFIA

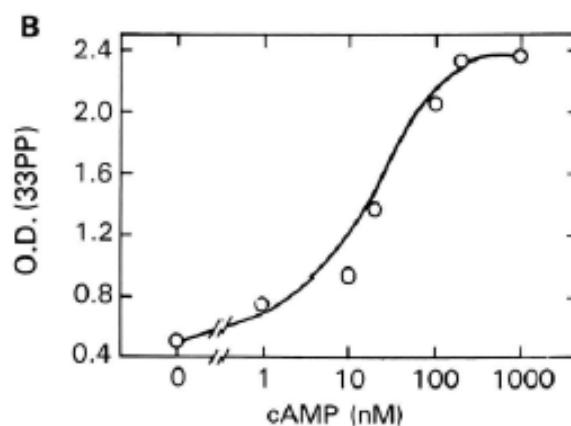
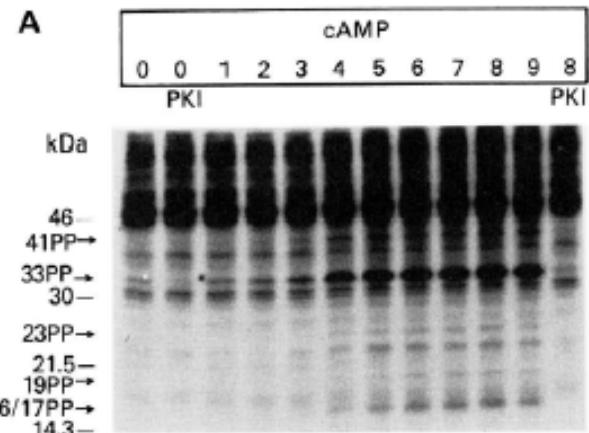
SUSTRADOS DE FOSFORILACION  $^{32}\text{P}$   
PROTEINAS TOTALES  $^{35}\text{S}$ -met



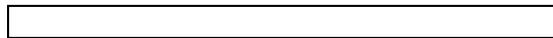
FOSFORO-32



EXPERIMENTO



FILM



REVELADO FOTOGRAFICO

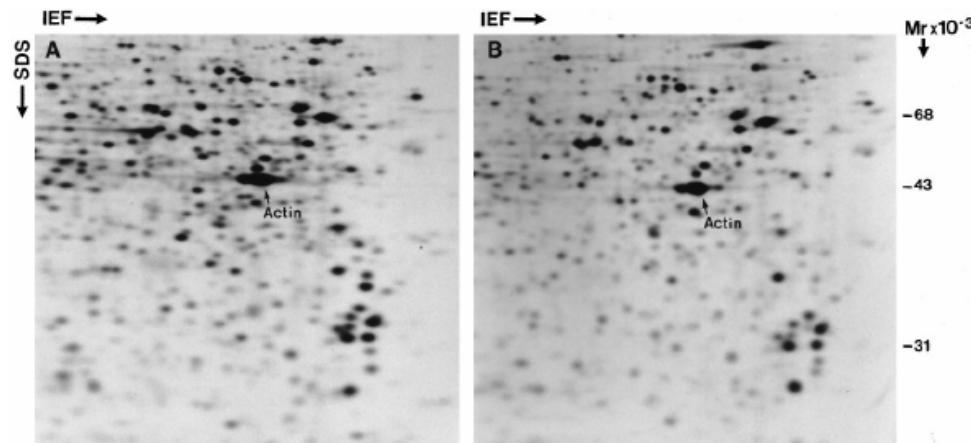
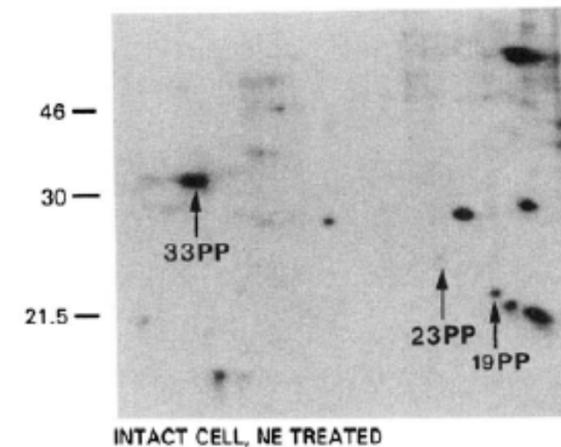
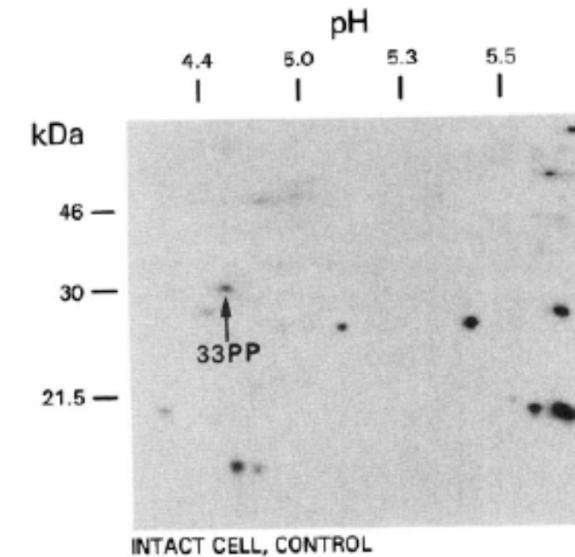


Fig. 9.  $[^{35}\text{S}]$ Methionine labeled proteins from newborn mouse kidney (A) and (B) lung. Only a fraction of the IEF gels are shown.

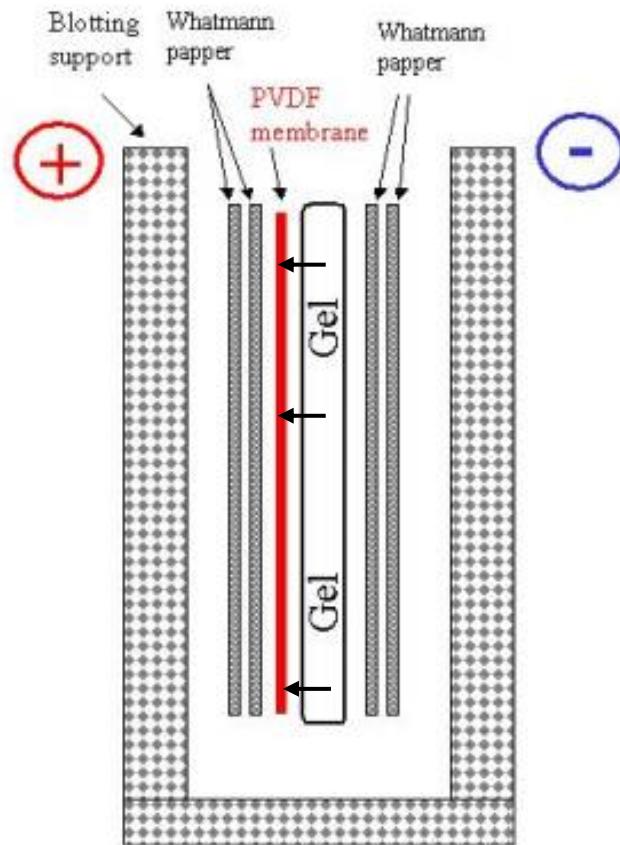
**MARCAJE CON  $^{35}\text{S}$ -MET**



**MARCAJE CON  $^{32}\text{PO}_4$**

# ELECTROTRANSFERENCIA DE PROTEINAS DESDE GELES DE POLIACRILAMIDA A MEMBRANAS DE NITROCELULOSA

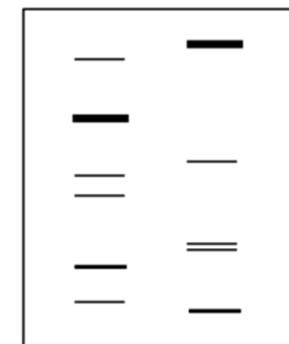
## WESTERN BLOT



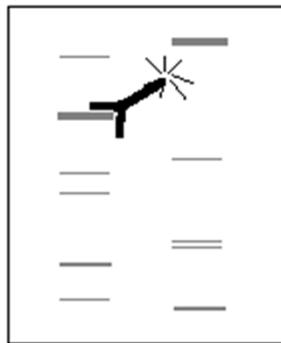
Protein Blot on  
Nitrocellulose



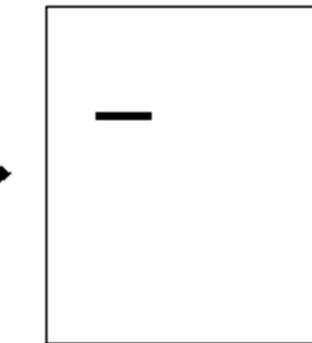
SDS Polyacrylamide  
Gel Electrophoresis



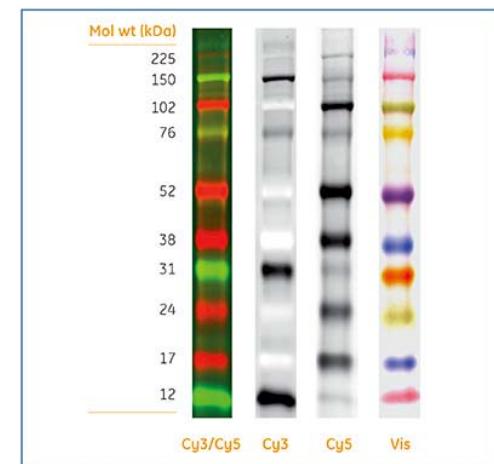
Label with Specific  
Antibody



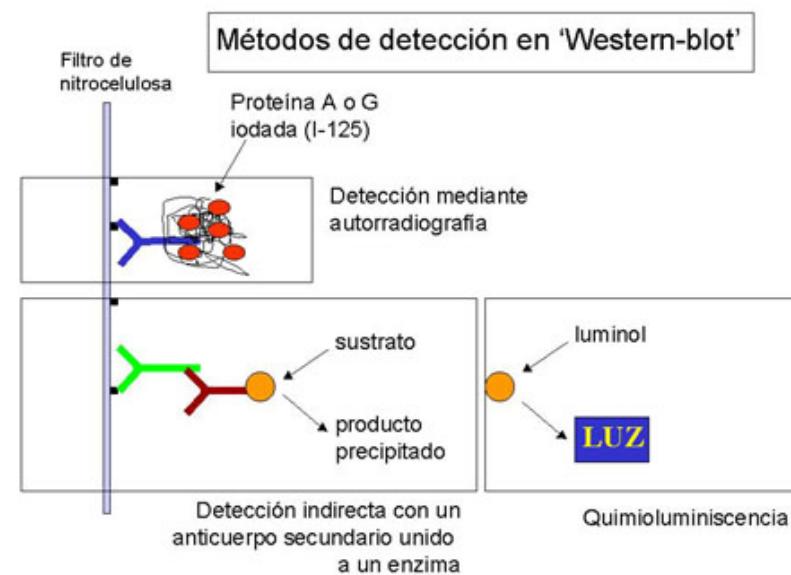
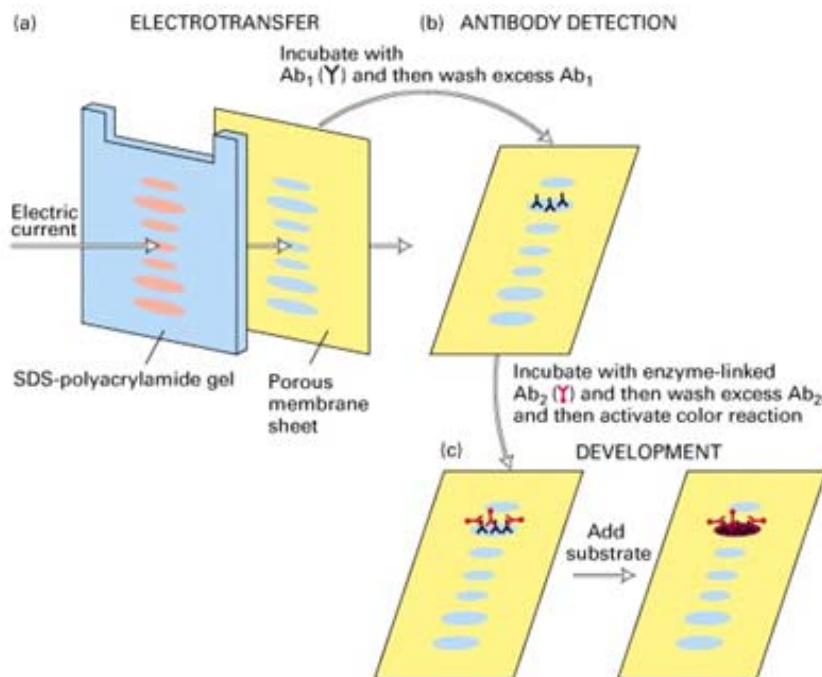
Detect Antibody



Reveals Protein  
of Interest



Typhoonで撮影したECL Plex Fluorescent Rainbow Markersです。  
左から順にFull-color Cy3/G5, Cy3 channel, Cy5 channel, 可視スペクトルになります。



32P-proteinas

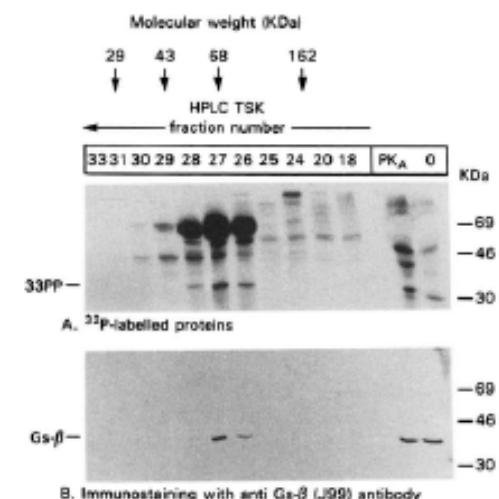
PAGE

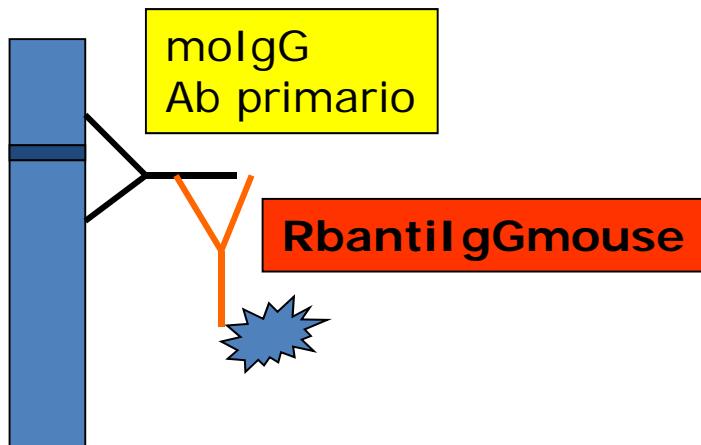
WESTERN

NITROCELULOSA

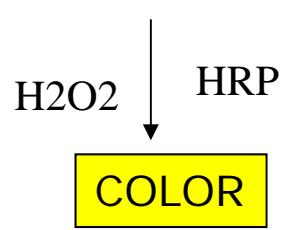
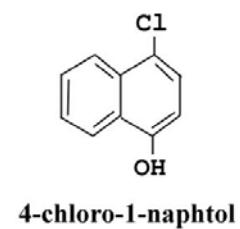
AUTORRADIOGRAFIA

INMUNODETECCION

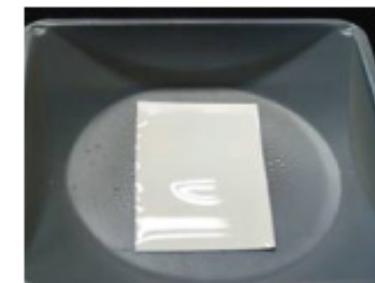
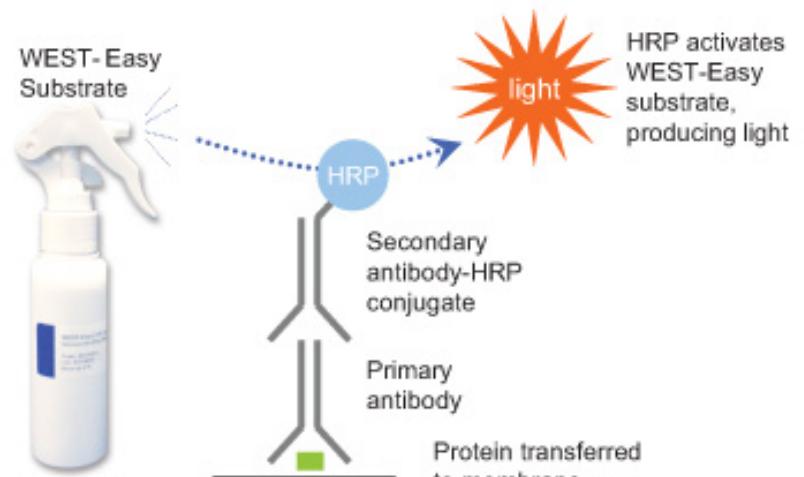


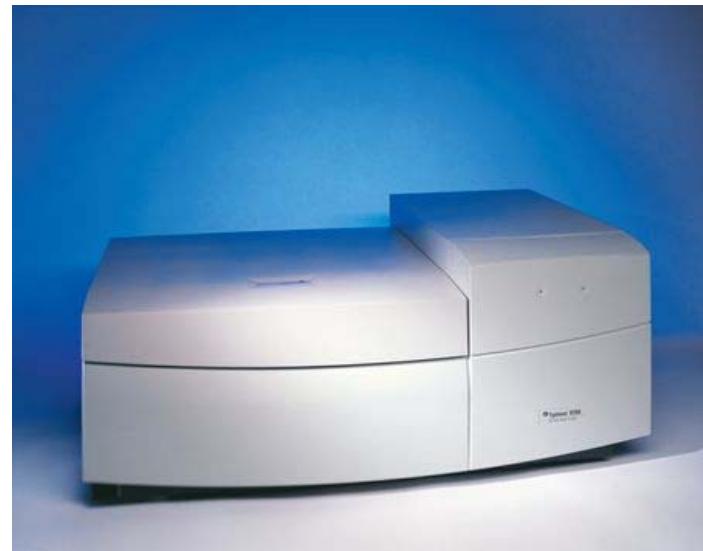
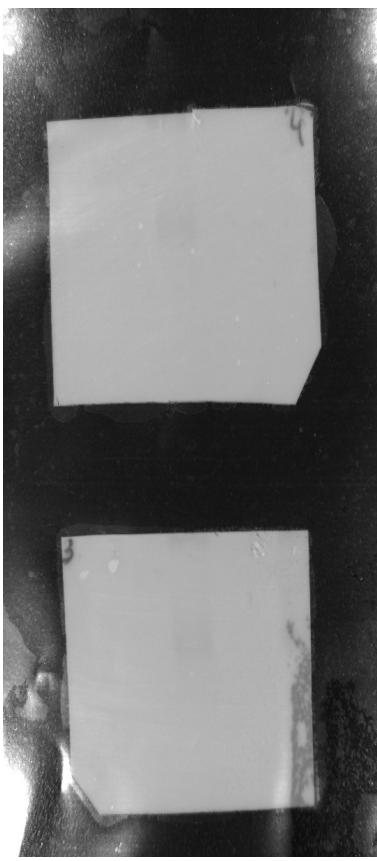


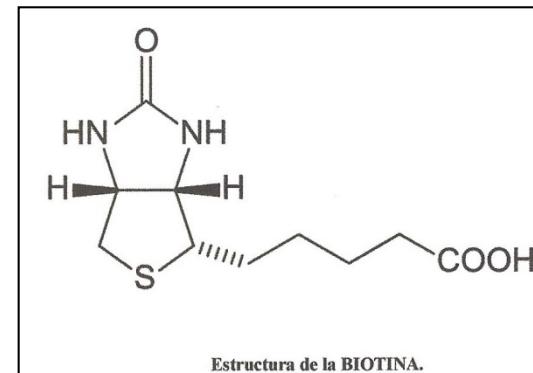
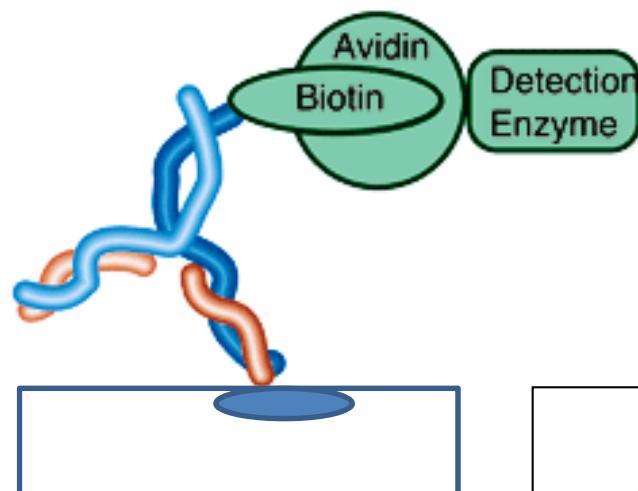
BLOT



Fluorescencia  
Quimioluminiscencia  
Actividad enzimática coloreada  
(peroxidasa, fosfatasa alcalina)



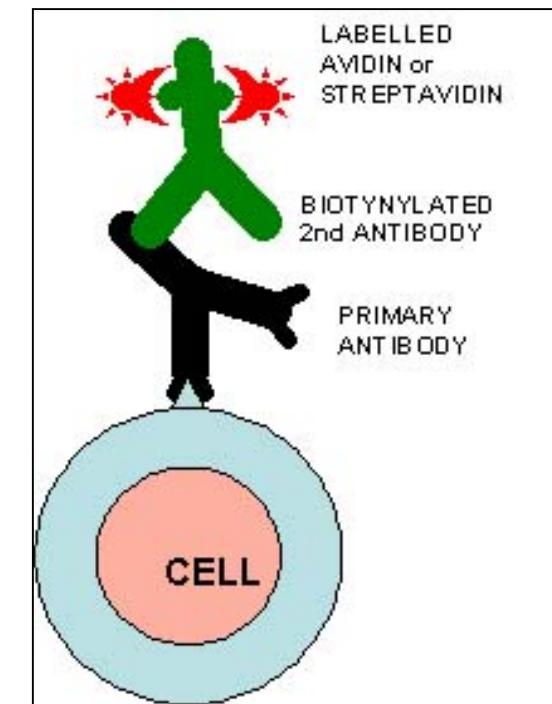


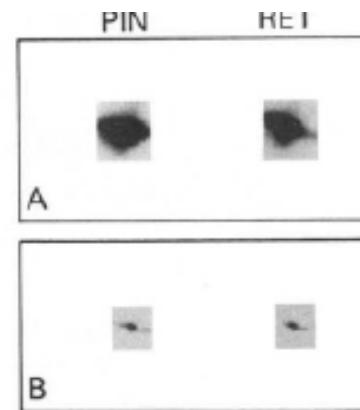
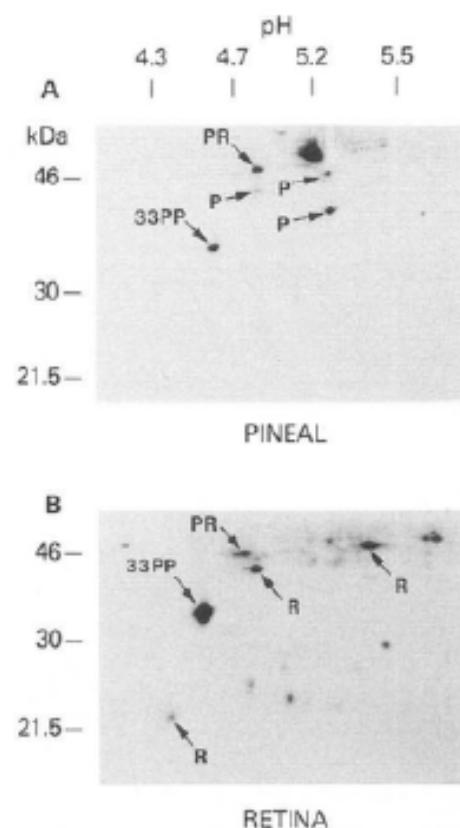


AVIDINA  
Abunda en la clara de huevo

$$K_d = 10^{-15} M$$

**AVIDINA + BIOTINA → AVIDINA-BIOTINA**





**FIG. 11. Immunological identification of pineal and retinal 33PP in two-dimensional blots as MEKA.** *A*, cytosolic fractions ( $100,000 \times g$ , 60 min) from rat pineal and retina were labeled for 10 min with [ $\gamma$ - $^{32}$ P]ATP in the presence of the catalytic subunit of the PKA and analyzed by two-dimensional-PAGE electroblotting and autoradiography. *B*, sections of the blots in *A* containing 33PP were stained immunologically using an anti-MEKA antiserum as described in the legend to Fig. 10. These results were confirmed in a second study.

**Acknowledgments**—We want to express our appreciation to the following for their assistance in obtaining antisera used in this report: K. Catt (National Institute of Child Health and Human Development), H. C. Hemmings and P. Greengard (The Rockefeller University), Allen Speigel (National Institute of Diabetes and Digestive and Kidney Diseases), Blake Kapinsky (Biogen), Toshimi Shinohara (National Eye Institute), and Tohru Abe (Akikta University School of Medicine). In addition, we would like to express our appreciation to A. Speigel and T. Shinohara for their critical and creative input during this study and to P. Roseboom (National Institute of Child Health and Human Development) for the sequence analysis.

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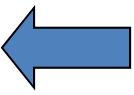
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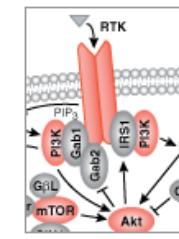
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### Vimentin (5G3F10) Mouse mAb #3390

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No.	Size	Price
3390S	100 ul ( 10 Western mini-blots )	please select country
custom	custom/drug discovery	email request

Applications	Reactivity	Sensitivity	MW (kDa)	Isotype
W IP F	H Mk	Endogenous	57	Mouse IgG1

**Applications Key:** W=Western Blotting IP=Immunoprecipitation F=Flow Cytometry  
**Reactivity Key:** H=Human Mk=Monkey  
Species cross-reactivity is determined by Western blot.

#### Specificity / Sensitivity

Vimentin (5G3F10) Mouse mAb detects endogenous levels of total vimentin protein.

#### Source / Purification

Monoclonal antibody is produced by immunizing animals with recombinant truncated human vimentin.

#### Western Blotting

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GI #: N/A NCBI Acc #: N/A Swiss Prot Acc #: N/A  
Length (aa): N/A Mol. Weight (Da): N/A Chrom Location: N/A

• Antigen: Kappa Light Chain  
• Immunogen: Mouse ? (kappa) light chain fragment  
• Family: Affinity Purified  
• Fraction: Affinity Purified IgG  
• Purity Note: This product was prepared from monospecific antiserum by immunoaffinity chromatography using antigens coupled to agarose beads followed by solid phase adsorption(s) to remove any unwanted reactivities. Assay by immunoelectrophoresis resulted in a single precipitin arc against anti-Rabbit Serum, Mouse IgG and Mouse Serum. Specificity was confirmed by ELISA at less than 1% cross reactivity against other mouse heavy or light chain isotypes.  
• Application: Suitable for immunoprecipitation, immunodiffusion, conjugation and most immunological methods requiring lot-to-lot consistency, high titer and specificity.  
• Recommended Dilution: ELISA 1:600,000  
WESTERN BLOTTING 1:5,000 - 1:50,000  
IMMUNOHISTOCHEMISTRY 1:1,000 - 1:5,000  
OTHER APPLICATIONS User Optimized  
• Physical State: Liquid (sterile filtered)

Printer Friendly Datasheet

Specific Categories

- Antibodies
- Secondary Antibodies
- Proteins
- Custom Antibody
- ELISA Kits
- Reagents
- Products by Pathway
- Seppro® Service
- Plasma Proteomics Products
- Plant Proteomics Products
- Protein Expression
- Cell Line Development

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15% Off Custom Peptide Services

Internet

Secondary Antibodies in Antibody Research - Windows Internet Explorer

http://www.sigmaaldrich.com/life-science/cell-biology/antibodies/learning-center/antibody-explorer/secondary-antibodies

Secondary Antibodies in Antibody Research

Home Products Order Center Custom Products Support All Search Advanced Sea

SIGMA Life Science

Life Science > Cell Biology > Antibodies > Learning Center > Antibody Explorer > Secondary Antibodies

## Antibodies

### Secondary Antibodies

WELCOME to the  
**ANTIBODY EXPLORER**  
Where Your Search Is Complete

Secondary Antibodies.

Antibody Explorer Home

- By Animal
- By Label
- Protein A, G and L Detection Reagents
- See also: Protein A, G and L Resins

Secondary antibodies are other host antibodies that bind to primary antibodies or antibody fragments. They are typically labeled with probe that make them useful for detection, purification or sorting applications. Secondary antibodies may be polyclonal antibodies or monoclonal antibodies. Polyclonal antibodies lack the specificity of monoclonal antibodies but frequently have higher sensitivities because polyclonal antibodies are a mixture of antibodies that may include very high affinity antibodies. Cloning may not effectively select the highest affinity antibody from a polyclonal host. Secondary antibodies are available with specificity for whole Ig molecules or antibody fragments such as the Fc or Fab regions.

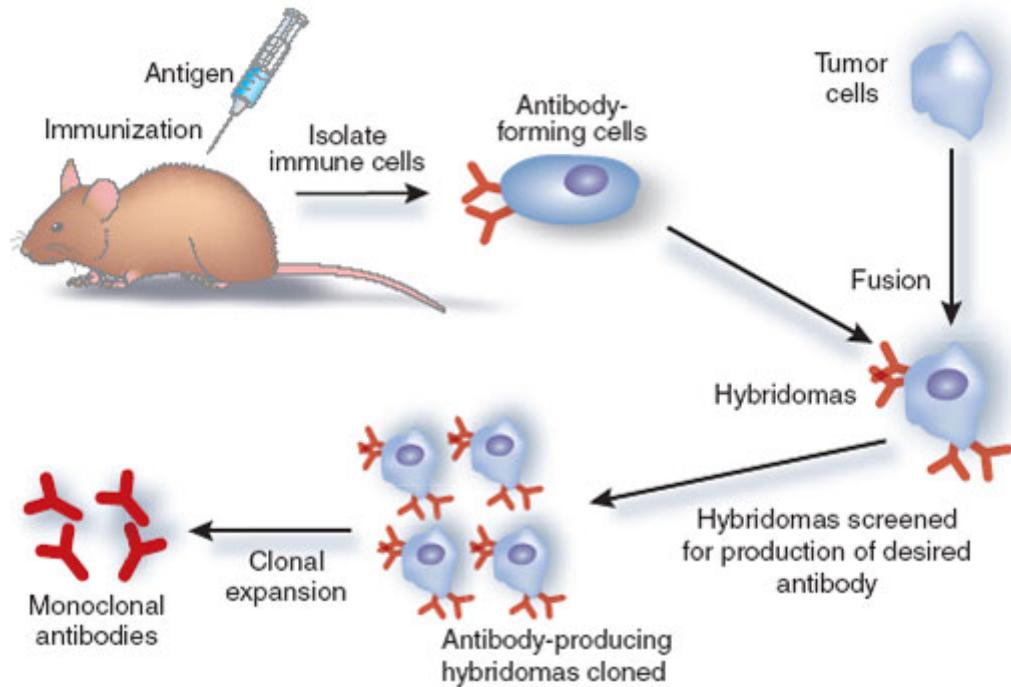
Sigma's polyclonal secondary antibodies are produced from the serum of host animals including mouse, rabbit, goat and sheep. Monoclonal secondary antibodies are produced from mouse hybridoma clones. Secondary antibodies are used in immunodetection, and immunoaffinity purification applications. Immunodetection applications include enzyme linked immunosorbent assays (ELISA); Western blotting; immunohistology, immunoblotting, immunostaining and cell based assays, such as cell-based immunochemical assays and high throughput cell-based screening assays (HTS). Secondary antibodies are useful for cell sorting, fluorescence activated cell sorting, FACS.

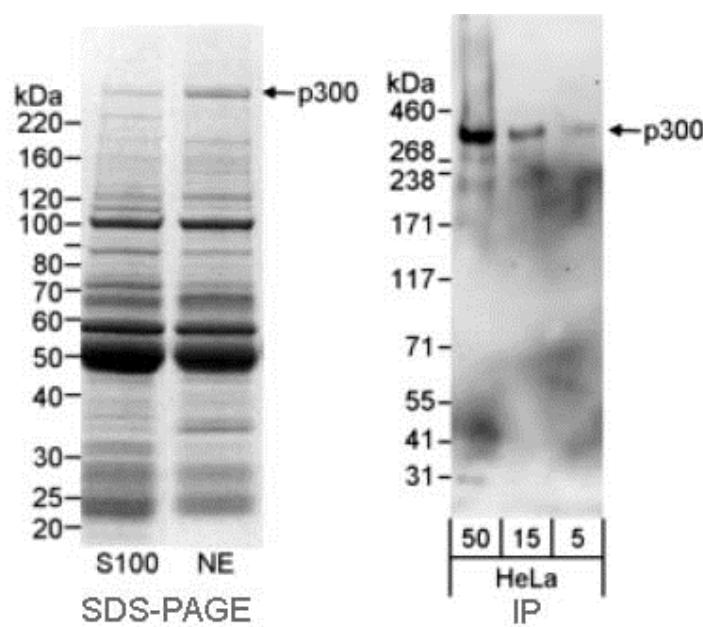
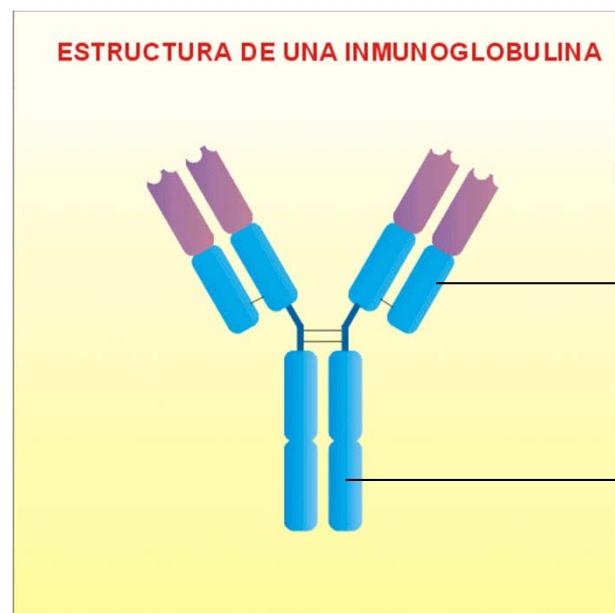
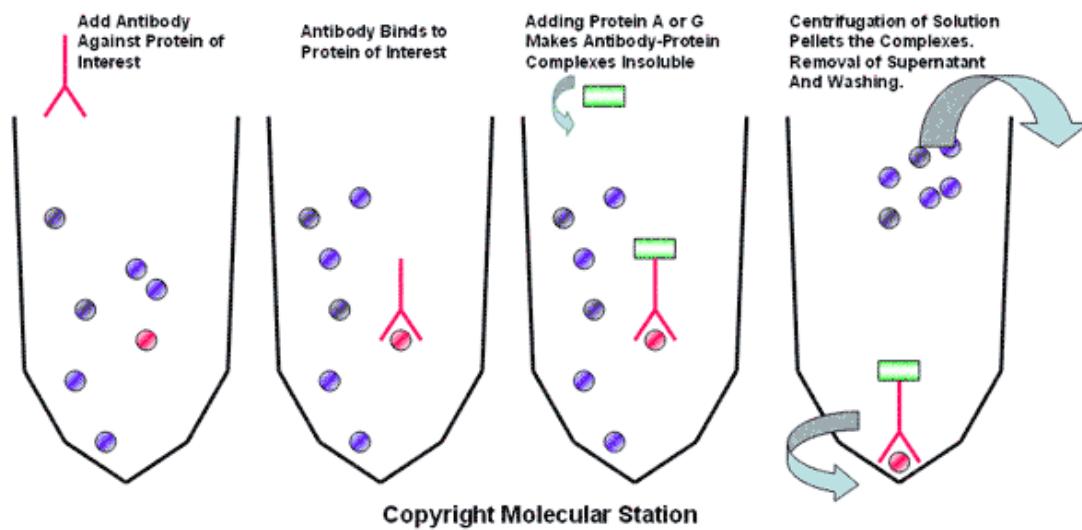
The specific utility of a secondary antibody depends upon its conjugated probe(s). Probes are molecules that support various detection technologies. The most common detection systems for conjugated secondary antibodies are colorimetric or fluorescent. Colorimetric

Internet 100% Norton™ 11:26

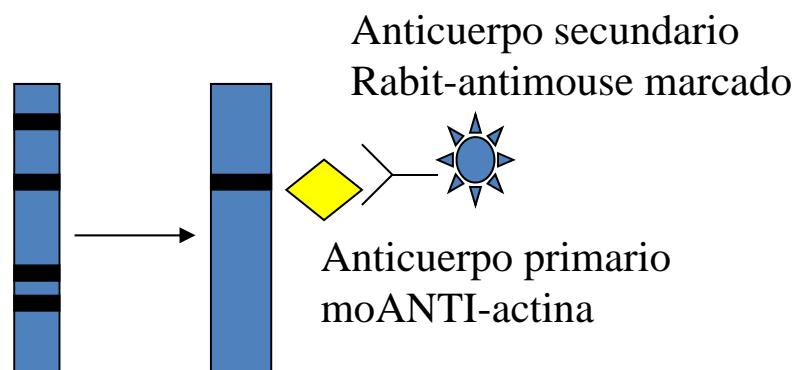
Inicio Adobe Photoshop Alb... Kingston (G:) Microsoft PowerPoint ... Secondary Antibodies... ES Norton™

## OBTENCION DE ANTICUERPOS

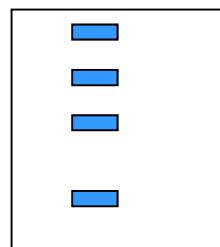




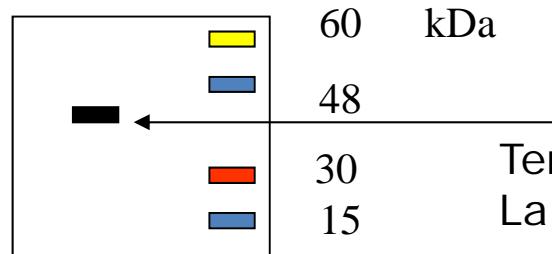
¿En que se basa el marcaje inmunológico de una proteína tras la separación electroforética, por la técnica denominada "western blot"?  
¿Que pasos y materiales se requieren para detectar ( o descartar) la presencia de una proteína concreta, por ejemplo ACTINA, en una solución tras electroforesis en SDS-poliacrilamida?



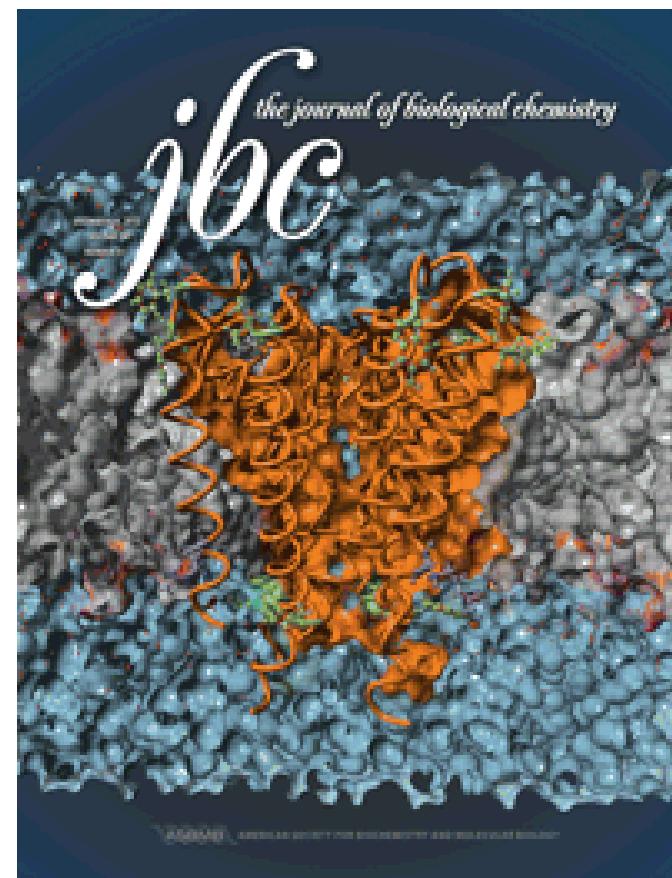
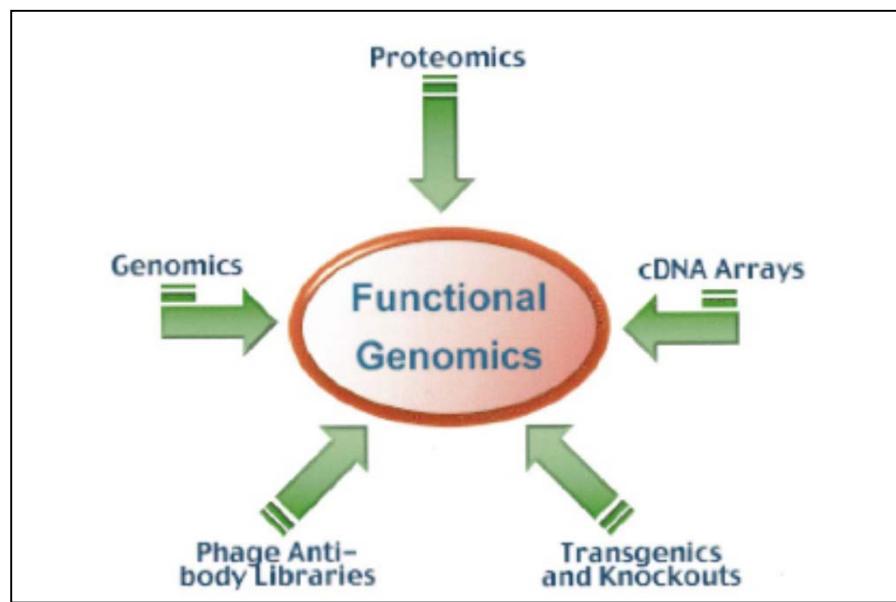
Electrotransferencia → membrana de nitrocelulosa



Rainbow markers

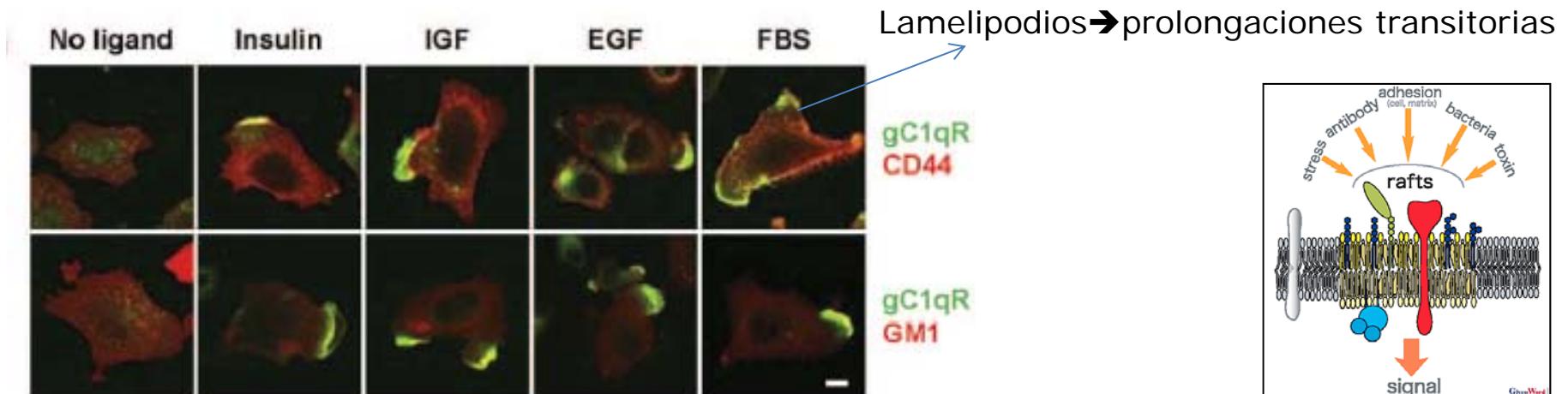


Tener una banda de un pm semejante a  
La actina no identifica a la proteína

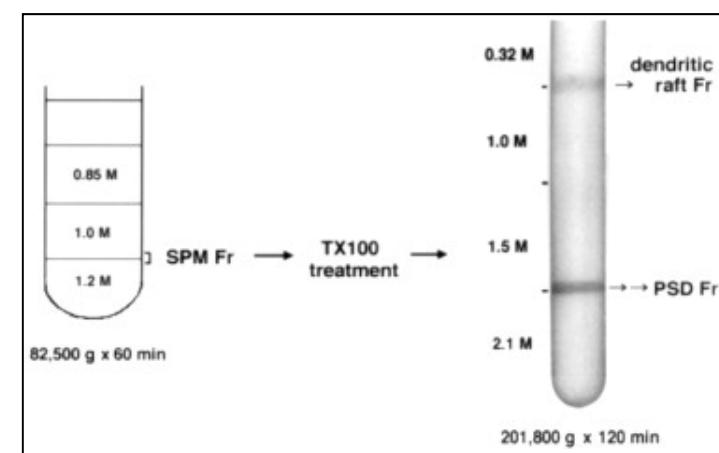
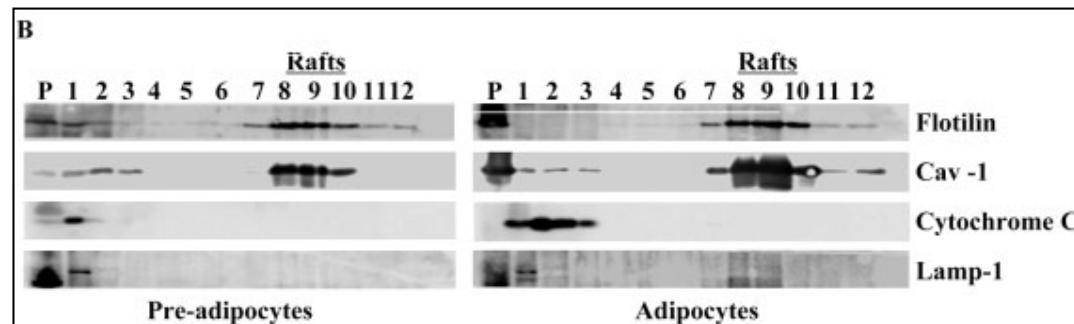


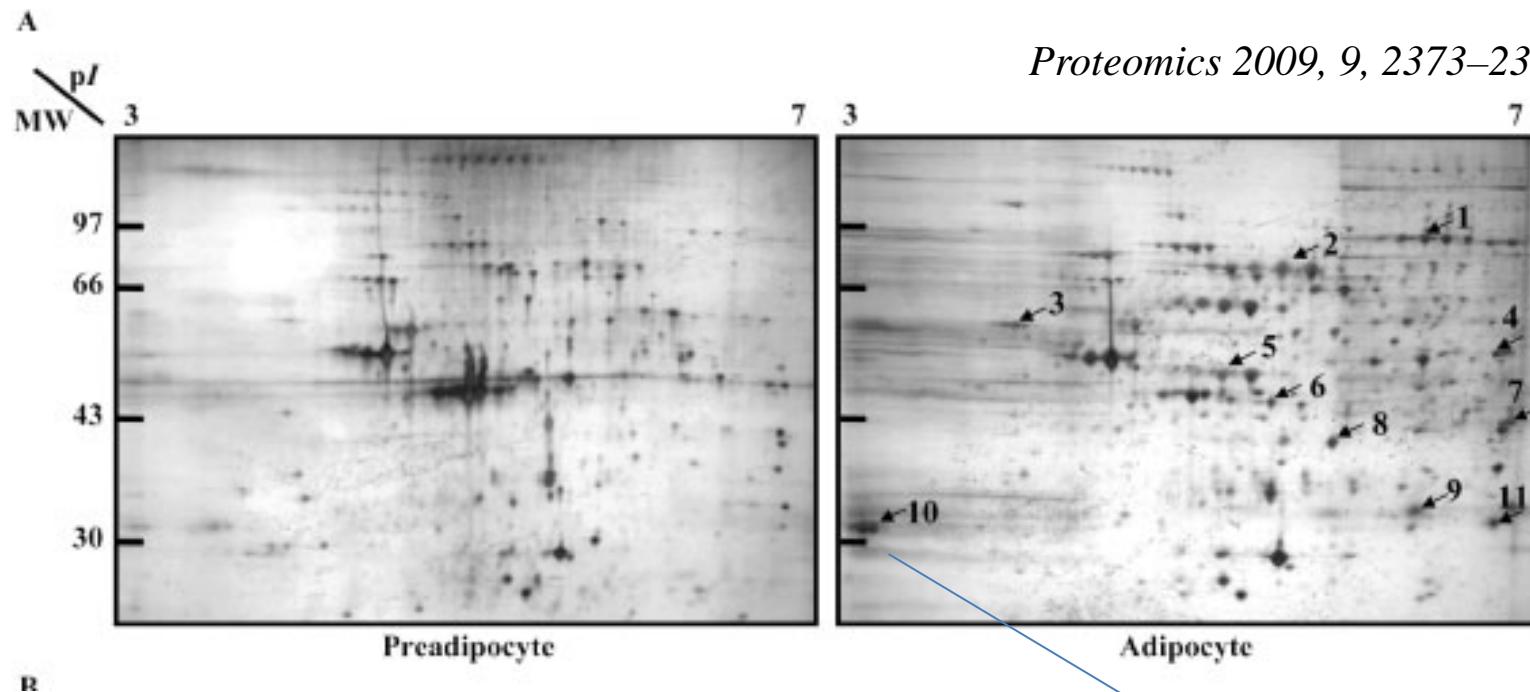
# Cell-surface Receptor for Complement Component C1q (gC1qR) Is a Key Regulator for Lamellipodia Formation and Cancer Metastasis

THE JOURNAL OF BIOLOGICAL CHEMISTRY VOL. 286, NO. 26, pp. 23093–23101, July 1, 2011



A549 cells de tumor de pulmón



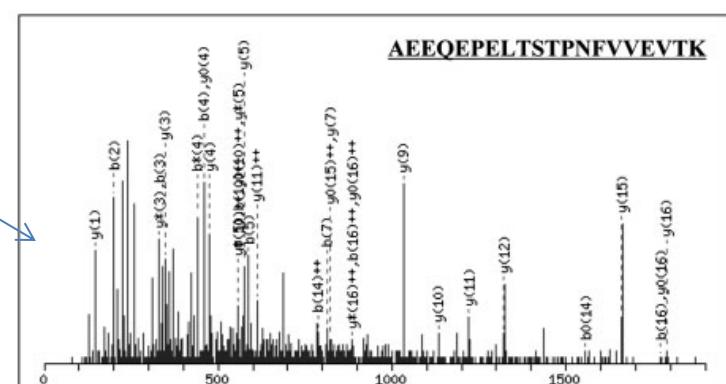


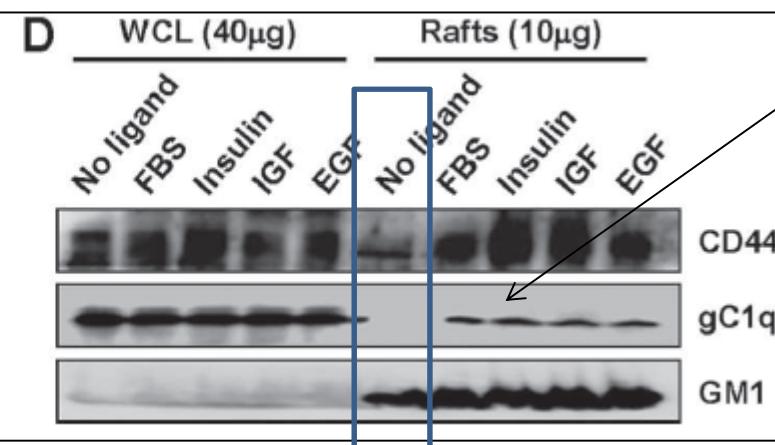
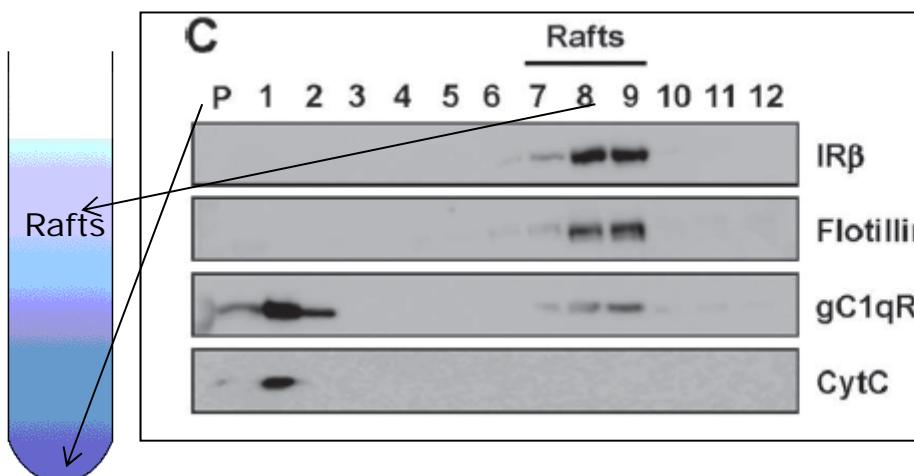
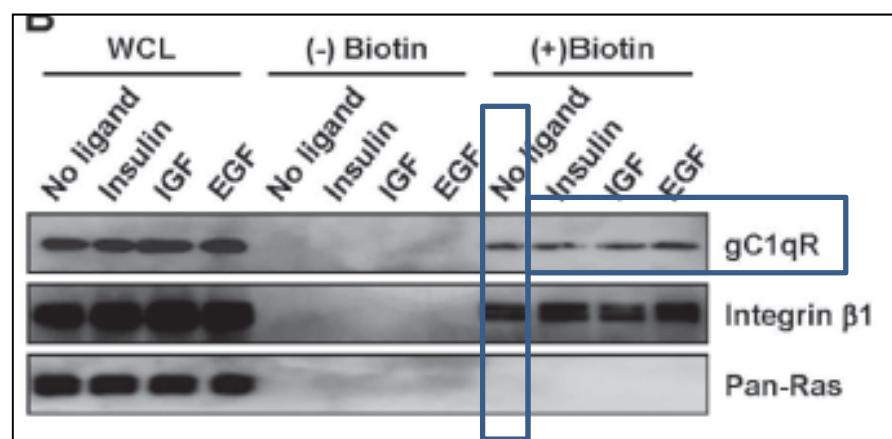
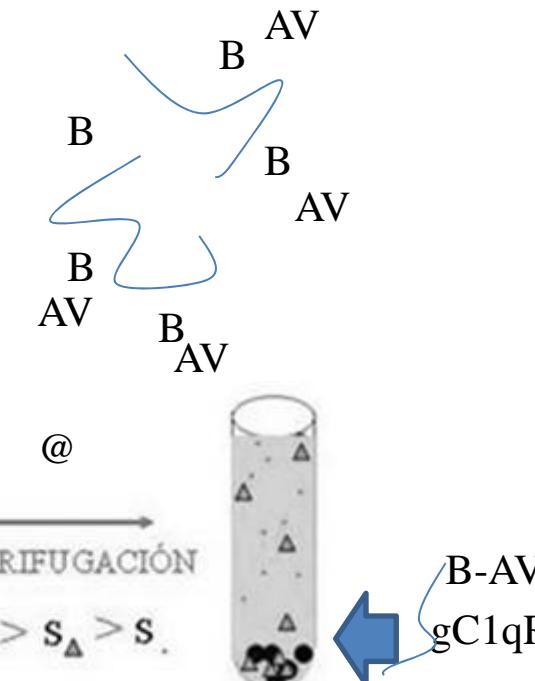
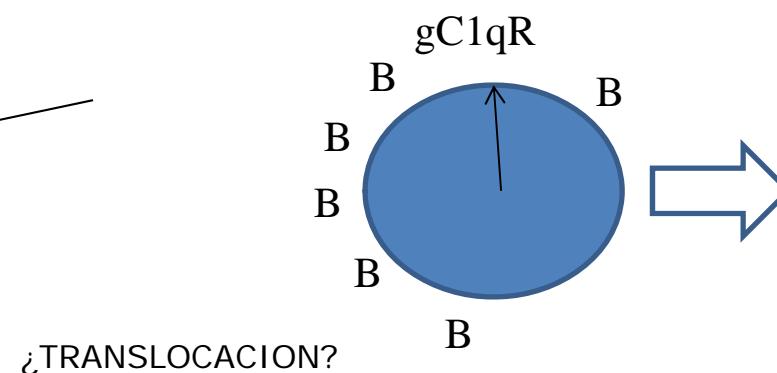
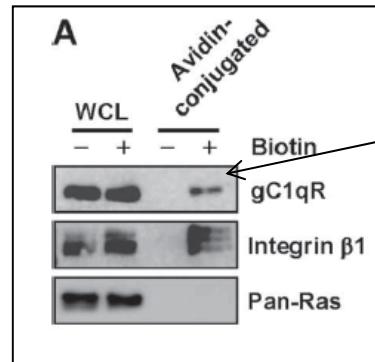
**A**

1 SARGHTVPWP GRPSLCTCPP RDAPSAALRA PRPRRRRPPA SRTAIPAQPL  
 51 RHLLQPAPRP CLRPFGLLSV RAGSARRSGL LQPPVPCACG CGALHTEGDK  
 101 AFVEFLTDEI KEEKKIQKHK SLPKMSGDWE LEVNGTEAKL LRKVAGEKIT  
 151 VTFNIINNSIP PTFDGEEEPS QGQKAEEQEP ELTSTPNFVV EVTKTDGKKT  
 201 LVLDCHYPED EIGHEDAES DIFSIKEVSF QATGDSEWRD TNYTLNTDSL  
 251 DWALYDHLMDFLADRGVDNT FADELVELST ALEHQEYITF LEDLKSFVKN  
 301 Q

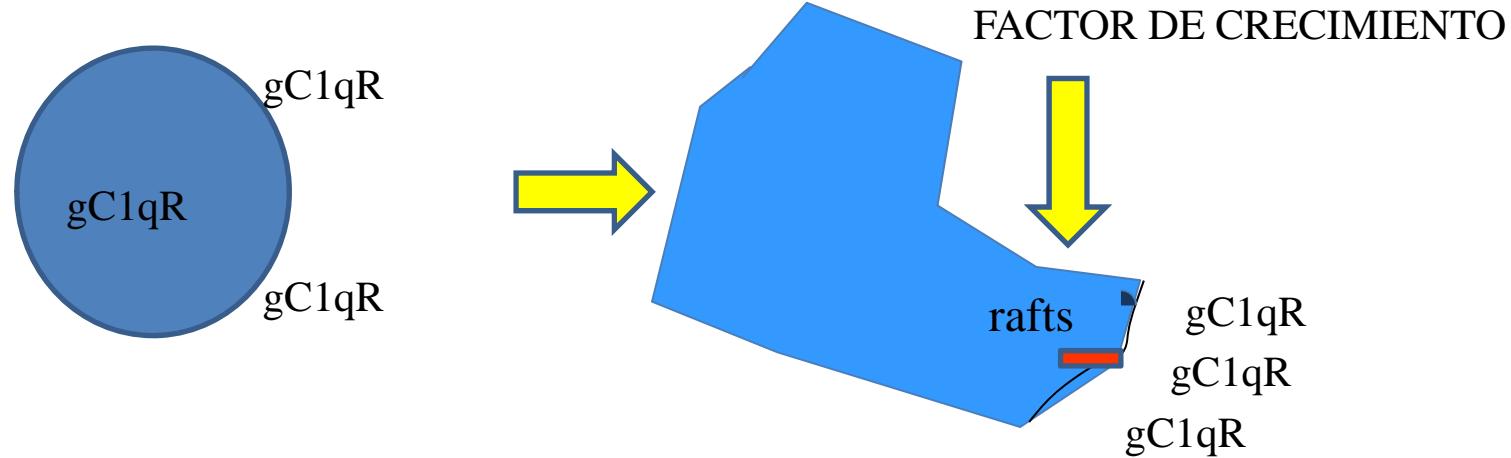
DIGESTIÓN PEPTIDICA  
 ANALISIS DE PEPTIDOS  
 POR ESPECTROMETRIA DE MASAS

**gC1qR=RECEPTOR PARA EL COMPONENTE DEL  
 COMPLEMENTO C1q**

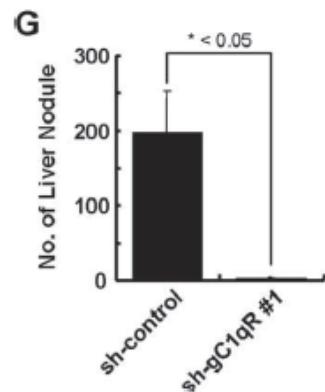
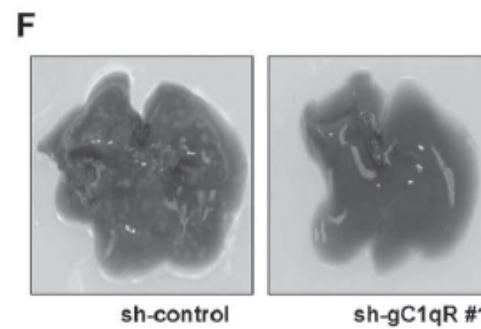
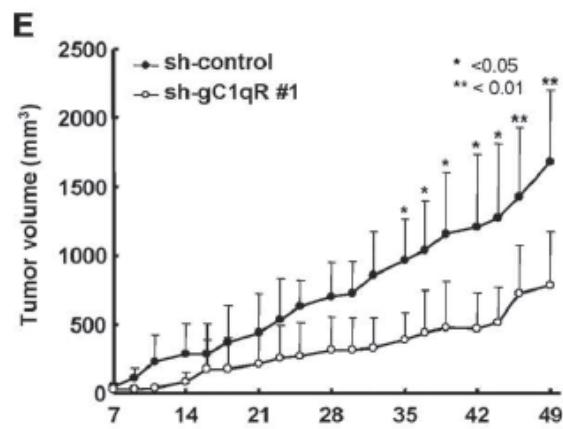
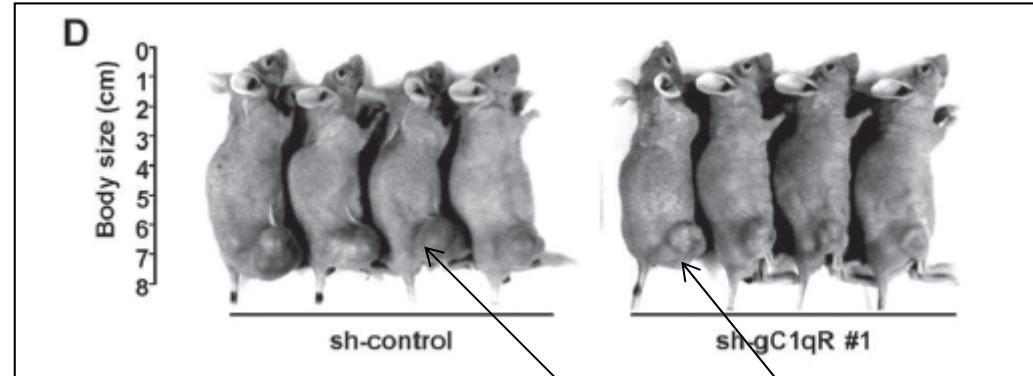




¿REORDENACION? → condensación inducida por factores



NO EXISTE TRANSLOCACION DE RECEPTORES gC1qR, SINO REORDENACIÓN O  
AGRUPACIÓN A ZONAS ESPECÍFICAS FUNCIONALES DE LA MEMBRANA (RAFTS)



El tamaño del tumor y las metastasis hepáticas Disminuyen con la inyección de las células Knockdown para gC1qR

